

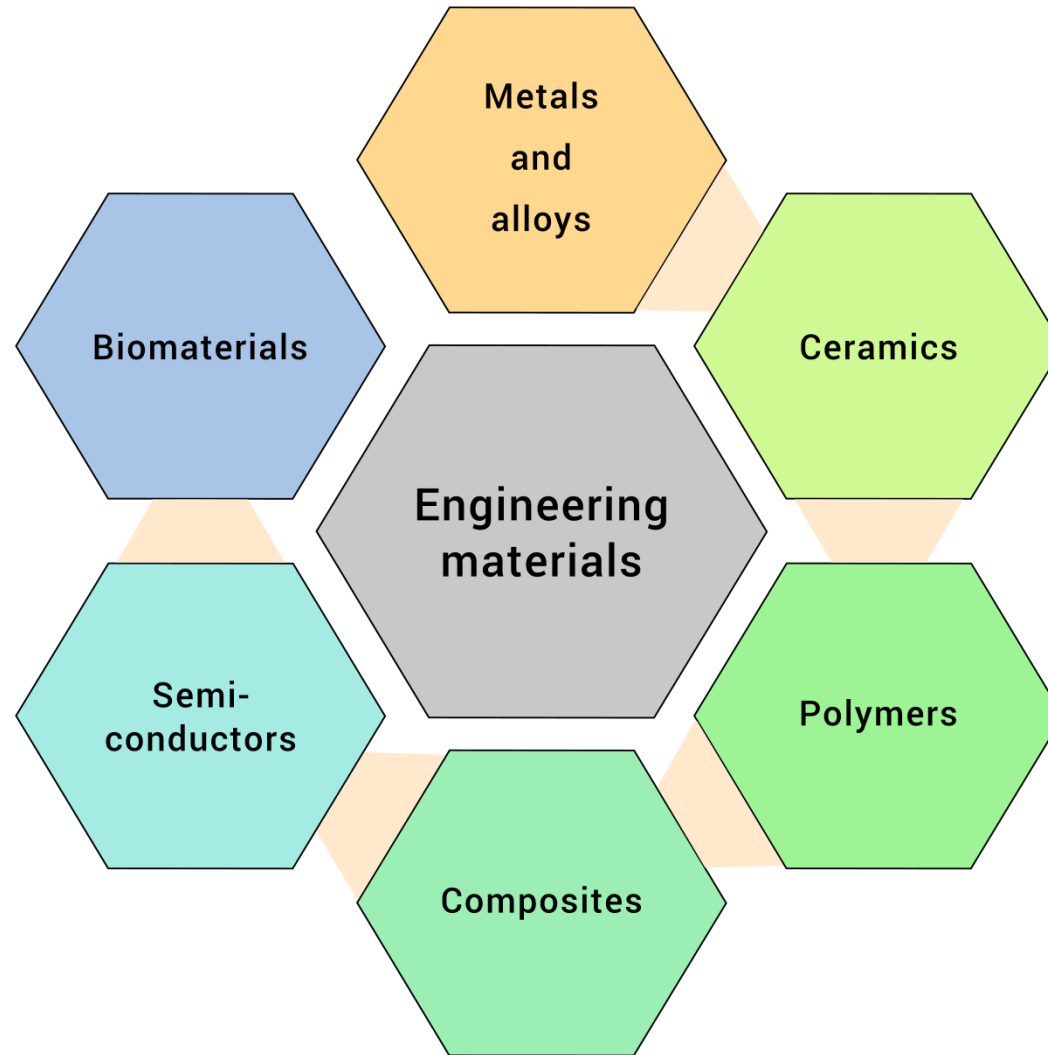


# Sustainability & Materials

Prof. Tiffany Abitbol

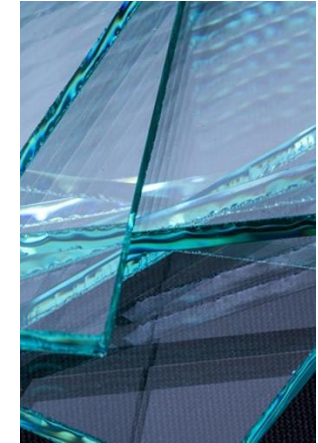
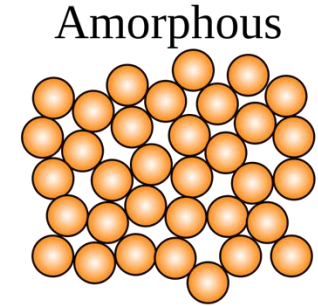
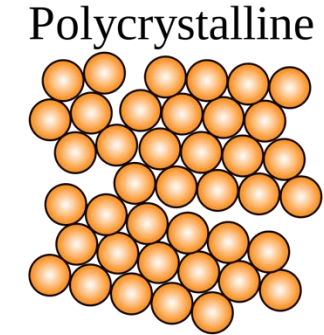
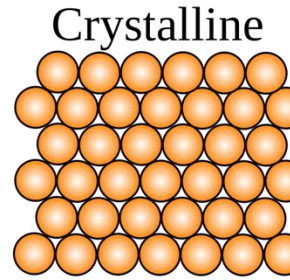
2025

# Engineering materials categories (variable)



Focus: Ceramics

- Non-metallic
- Inorganic
- Ionic and/or covalent bonds
- Mostly made by heat processing: heating and cooling
- Crystalline or semicrystalline except glass, which is amorphous
- Examples: alumina, silica, titania, boron nitride, etc.,







- Considered one of the greatest and earliest successes of humankind – control of fire
- Among the 1<sup>st</sup> objects manufactured
- From Greek “keramos” meaning burned earth
- Characteristics – long service life, low density, chemically inert, corrosion resistant, electromagnetic response, non-toxic, heat and fire resistant, sometimes electrical resistance or porosity



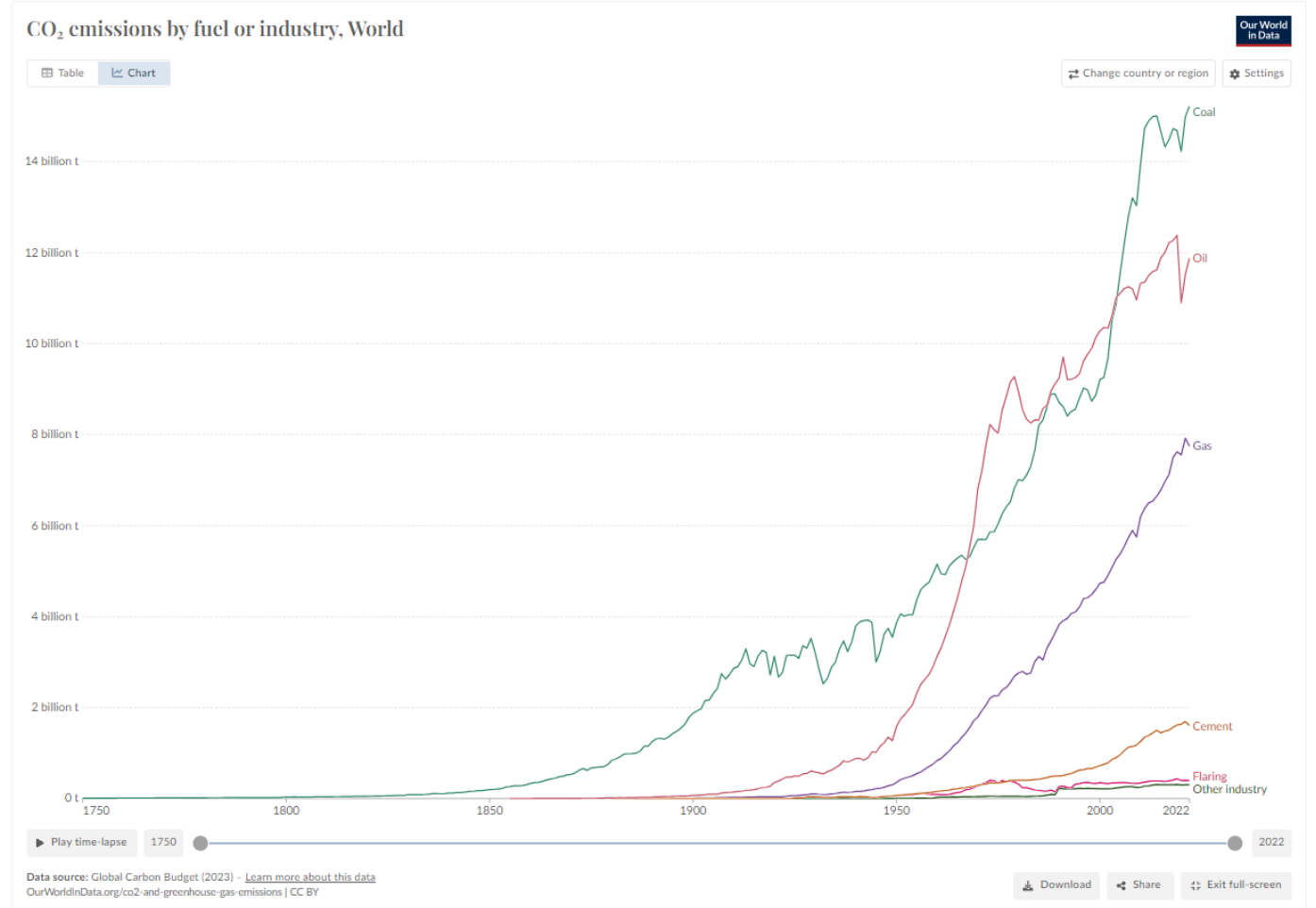
[Decarbonizing ceramics\\_2022](#)

- 1<sup>st</sup> pieces reported ca. 24,000 years ago
- Many uses today: promising for aerospace and high temperature structural applications, information storage and optical devices, oral prosthetics, water purification, bone void fillers, CO<sub>2</sub> adsorbents, etc.,
- Value of global ceramics market was approx. 230 billion in 2018, and growing due to constant growth in the construction industry, technological advancements in nanotechnology, 3D printing, and ceramics in health





- In EU: production of refractories, wall and floor tiles, and bricks and roof tiles emit around 19 Mt CO<sub>2</sub>
- Globally: brick manufacture is responsible for 2.7% of annual carbon emission; cement (considered a ceramic in this course's context) is responsible for 5-8% of annual emissions

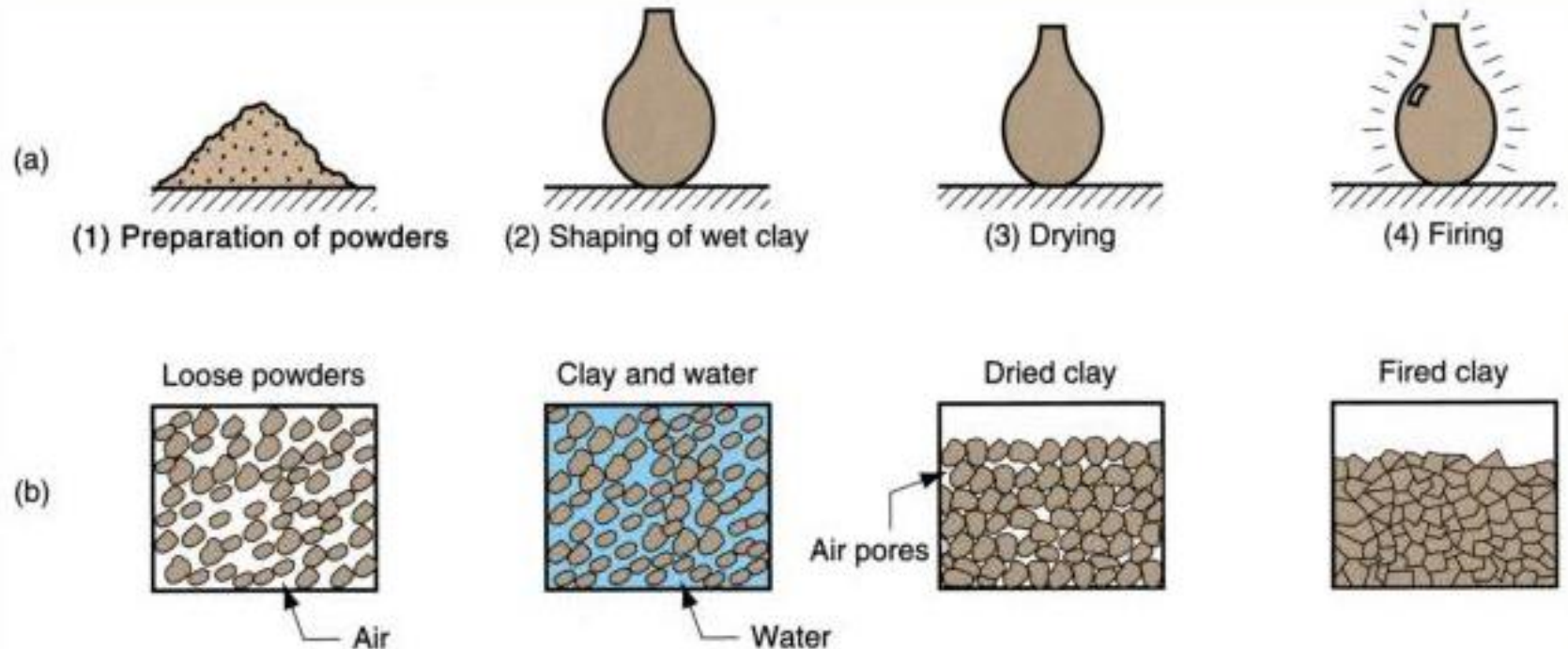




# Traditional ceramics

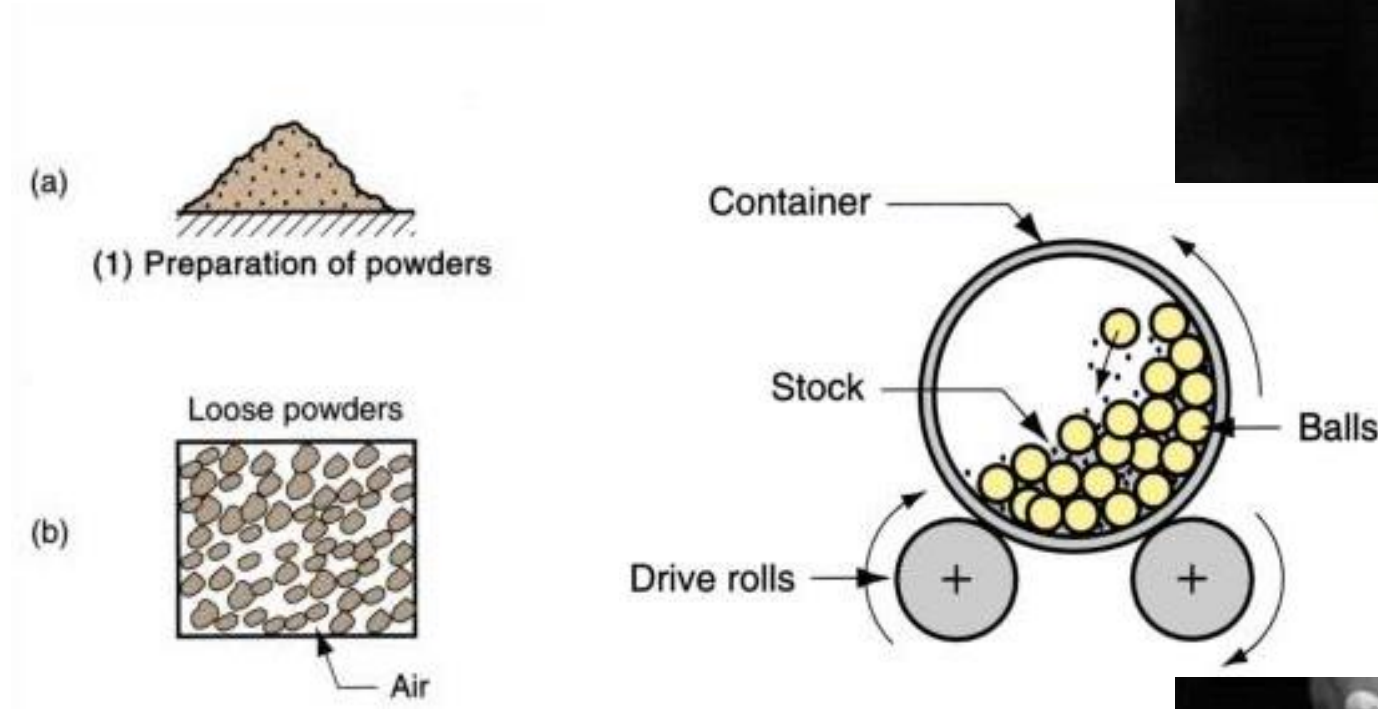
## Processing ceramics

- Traditional ceramics are made from naturally occurring minerals
- Mineral extraction has the usual consequences: habitat destruction, soil erosion, water pollution
- Open pits and quarries common

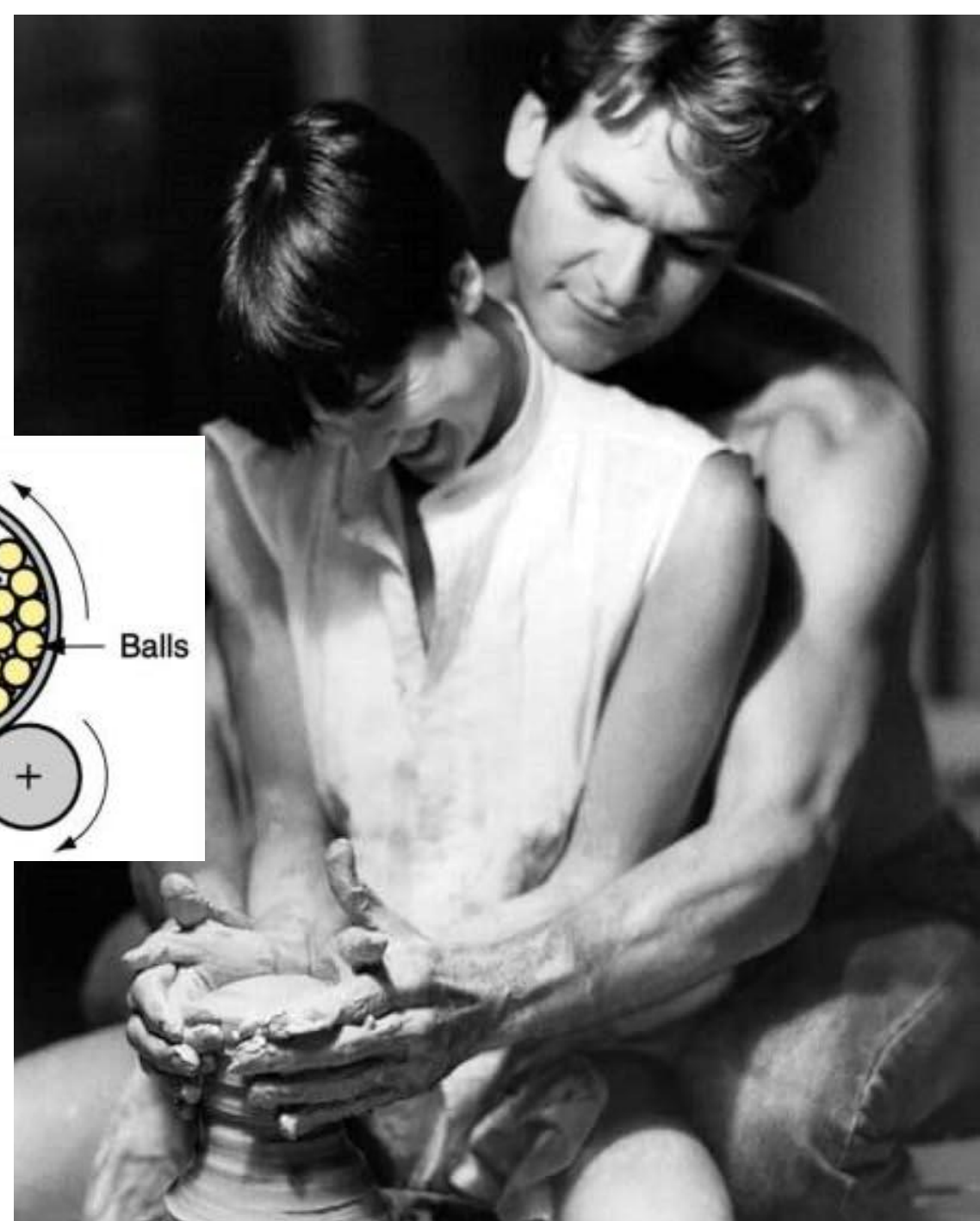


# Powders

## Processing ceramics

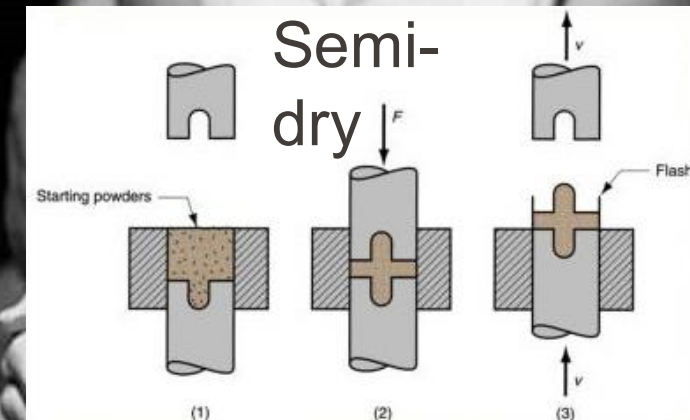
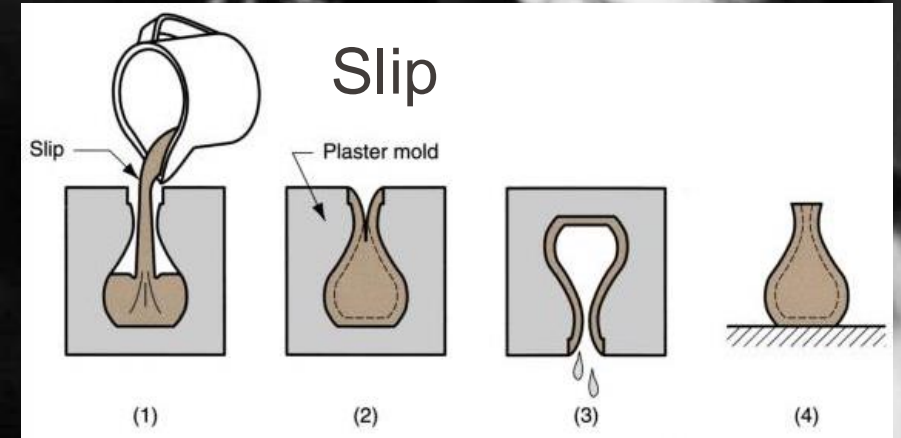
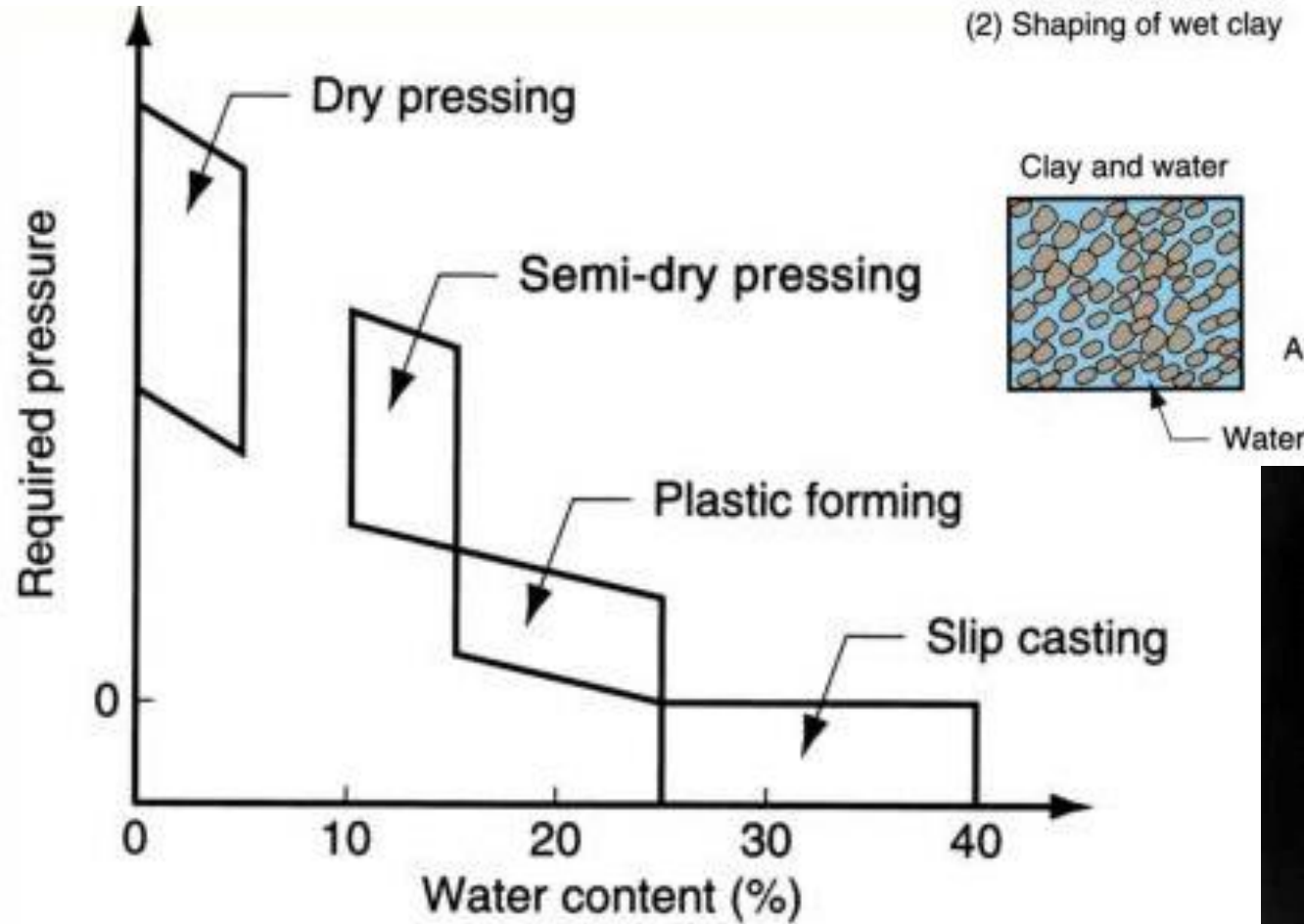


- Starts with a powder
- Powder usually made by ball crushing and then grinding, which is very energy intensive!



# Shaping

Processing ceramics

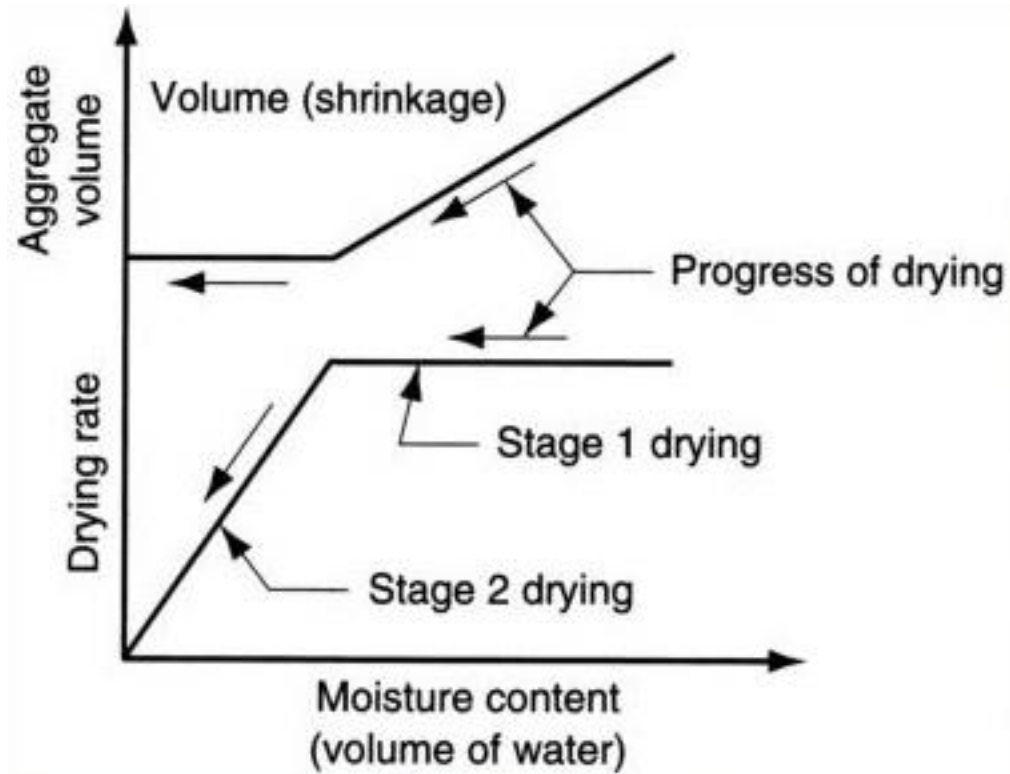


- Water is needed for shaping only/plasticity
- High pressures required for semi-dry/dry shaping



# Drying – 2 stages

Processing ceramics



- Stage 1: fast and constant rate to evaporate surface water (prone to shrinkage)
- Stage 2: enough water has been removed so particles are in contact (no shrinkage)

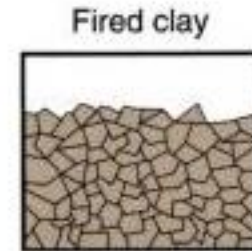




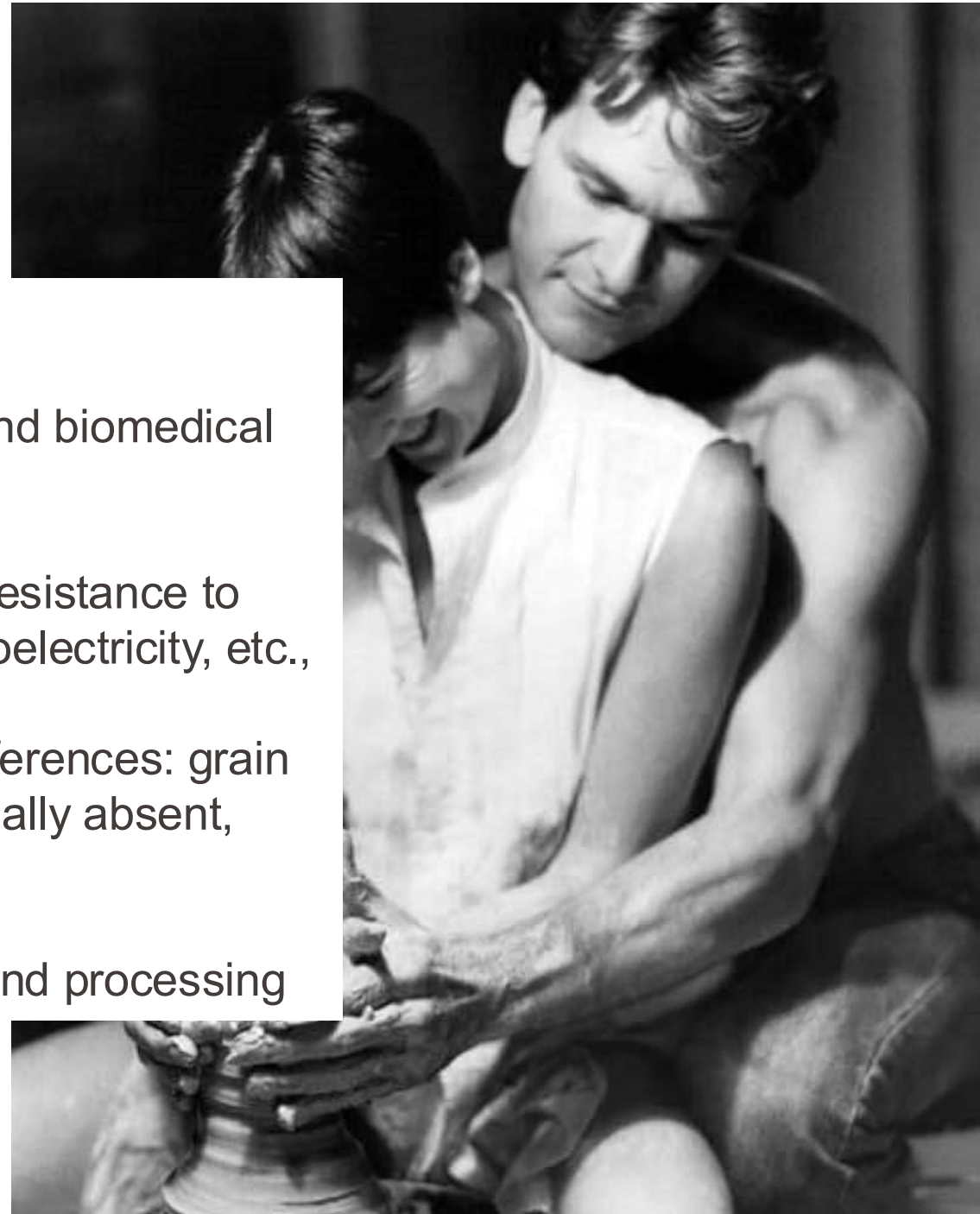
# Firing

## Processing ceramics

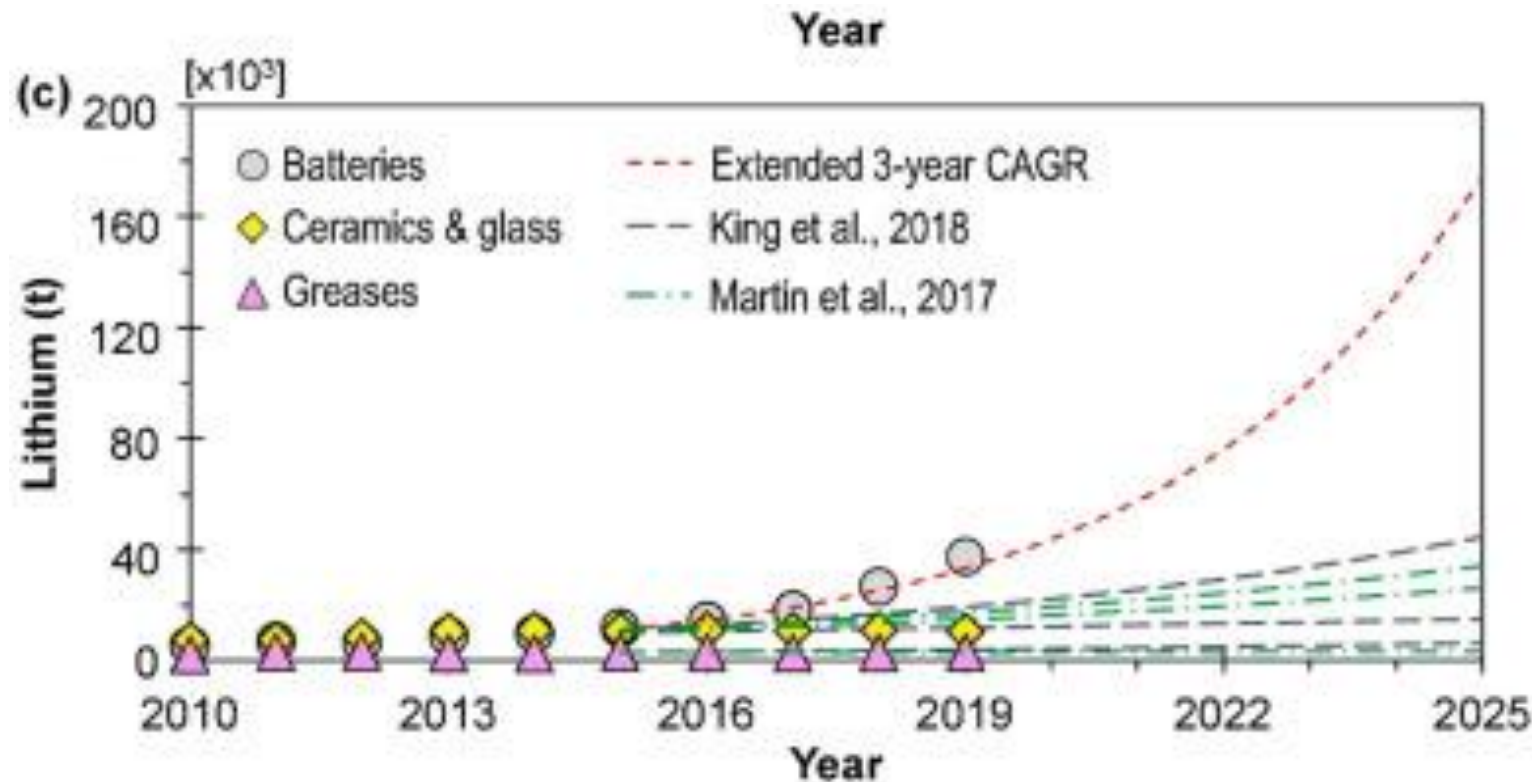
- Heat treatment to *sinter* – form bonds between particles, occurring with densification
- Performed in kiln
- Biggest impact to emissions, related to use of fossil fuels to heat the kiln and process emissions from carbonate raw materials which emit  $\text{CO}_2$  when heated
- If glaze is applied, the ceramic is fired again!



- Made from manmade raw materials
- Developed to meet industrial, technological, and biomedical needs
- Exhibit exceptional properties: high strength, resistance to wear, electrical insulation or conductivity, piezoelectricity, etc.,
- Same general processing steps with some differences: grain size can be controlled chemically, water is usually absent, etc.,
- Emissive hot spots depend on raw materials and processing



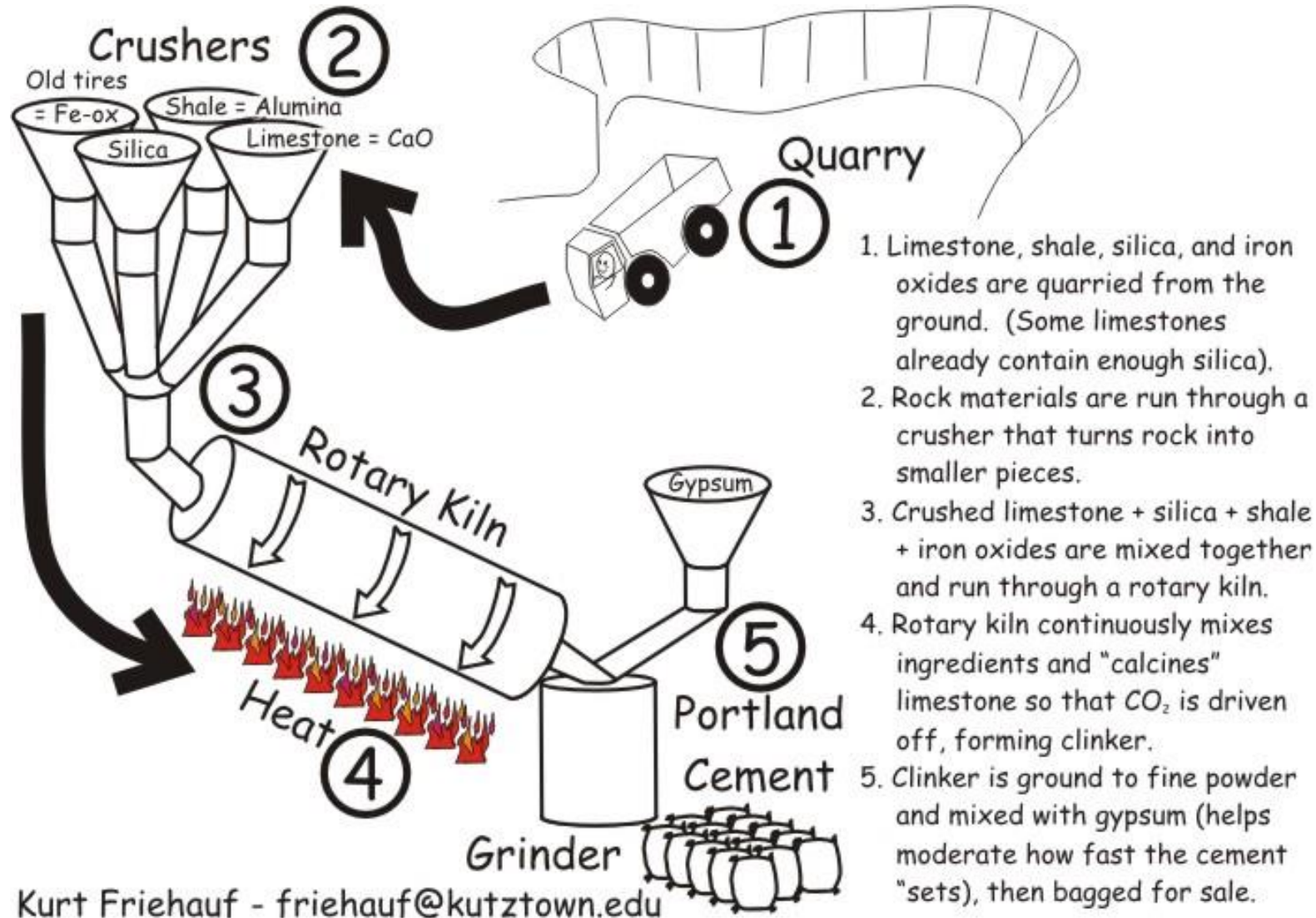
- Lithium is used in advanced ceramics due to its ability to enhance certain properties: low thermal expansion, high strength and toughness, melting point reduction, improved optical properties, enhanced electrical properties, light weight



- Cathode material in batteries
- Now batteries, but before 2015, ceramics and glass was the main industry using lithium

[Decarbonizing ceramics\\_2022](#)

# How cement is made

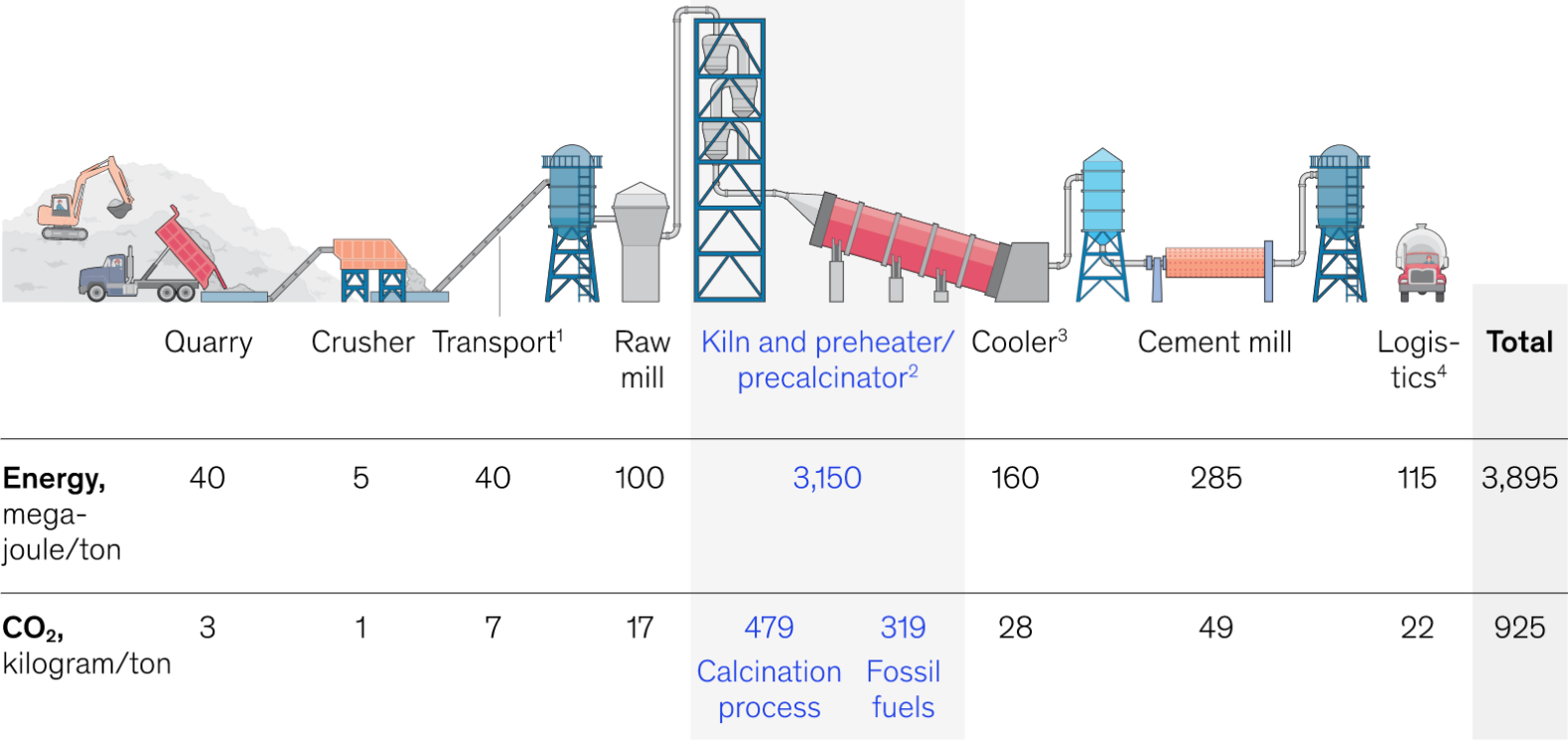


Cement manufacture



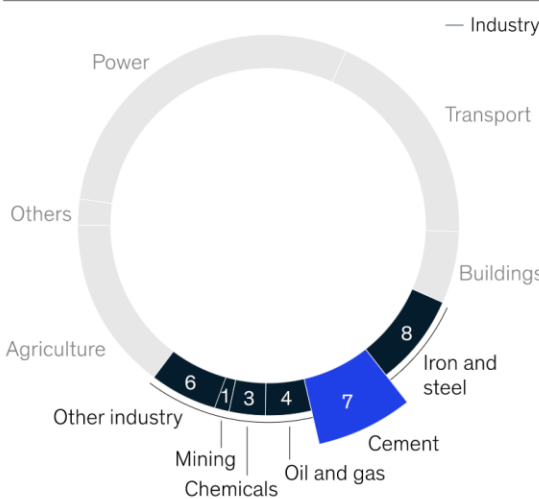
Raw materials, energy, and resources

Clinker and cement manufacturing

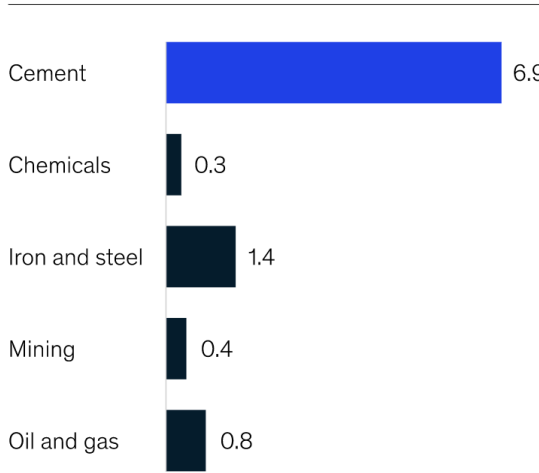


- The cement industry alone is responsible for about a quarter of all industry CO<sub>2</sub> emissions, and it also generates the most CO<sub>2</sub> emissions per dollar of revenue
- About two-thirds of those total emissions result from calcination, the chemical reaction that occurs when raw materials such as limestone are exposed to high temperatures.

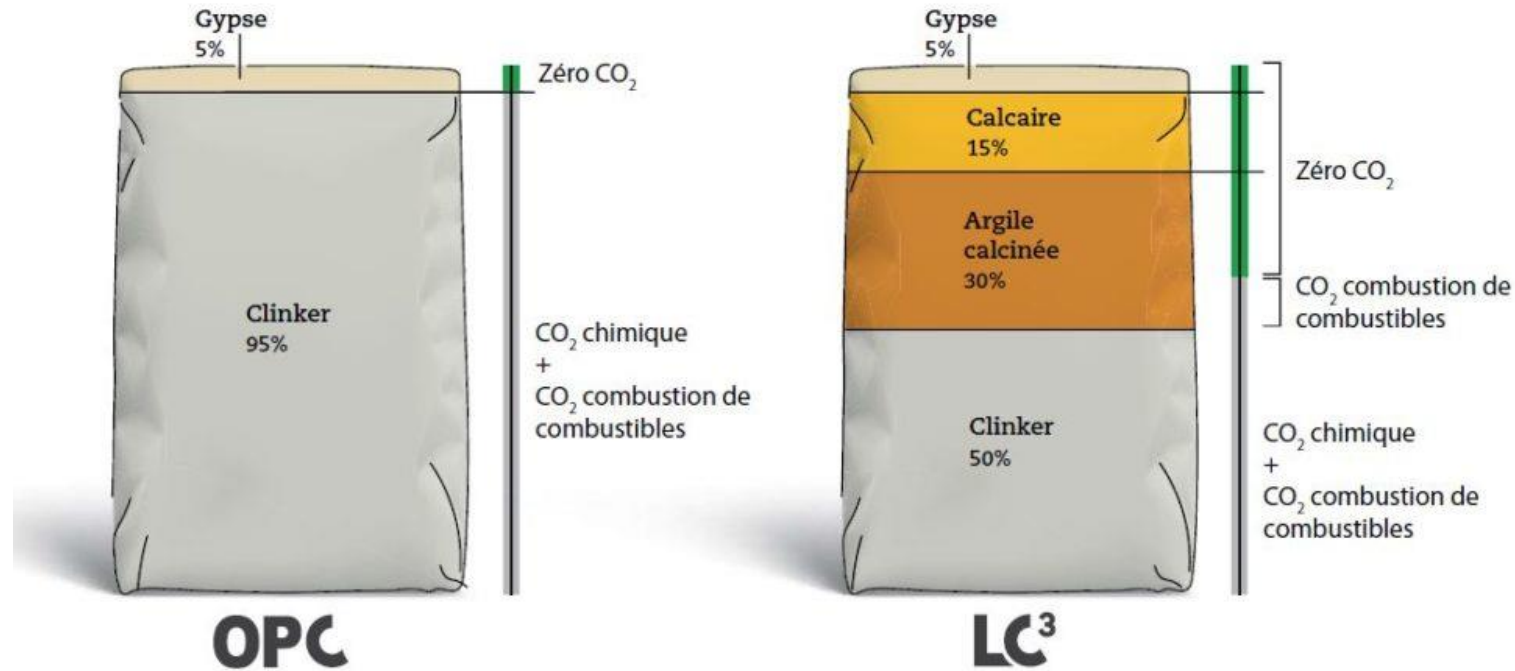
Share of global CO<sub>2</sub> emissions, % in 2017



kg of CO<sub>2</sub> per \$

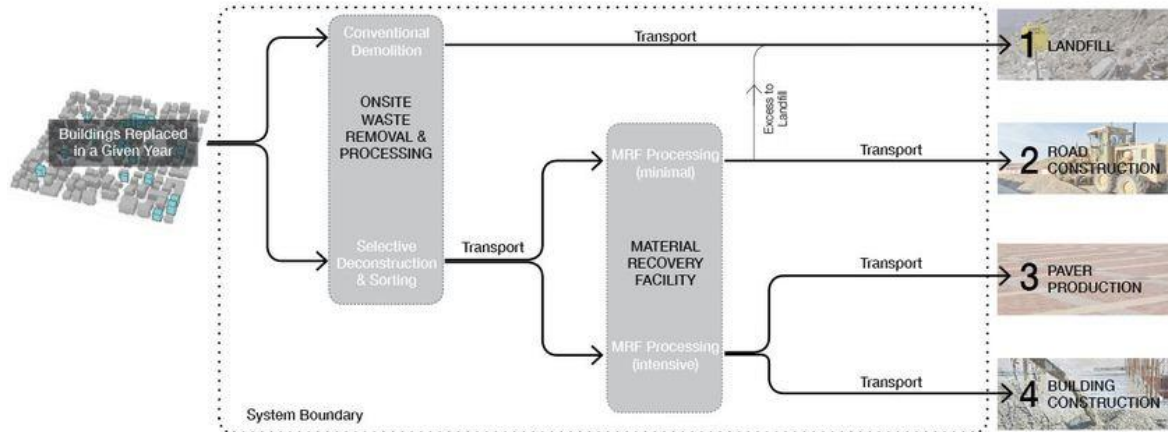


# Decarbonizing cement : EPFL connection (LMC)



- Replaces half of clinker with calcined clay and ground limestone, neither of which releases CO<sub>2</sub> when heated the way limestone does
- Clay is heated to lower temperatures, reducing fuel needed and emissions
- Lower temperatures mean that cleaner energy, like electricity can be used

# End of life – Ceramics and Glasses



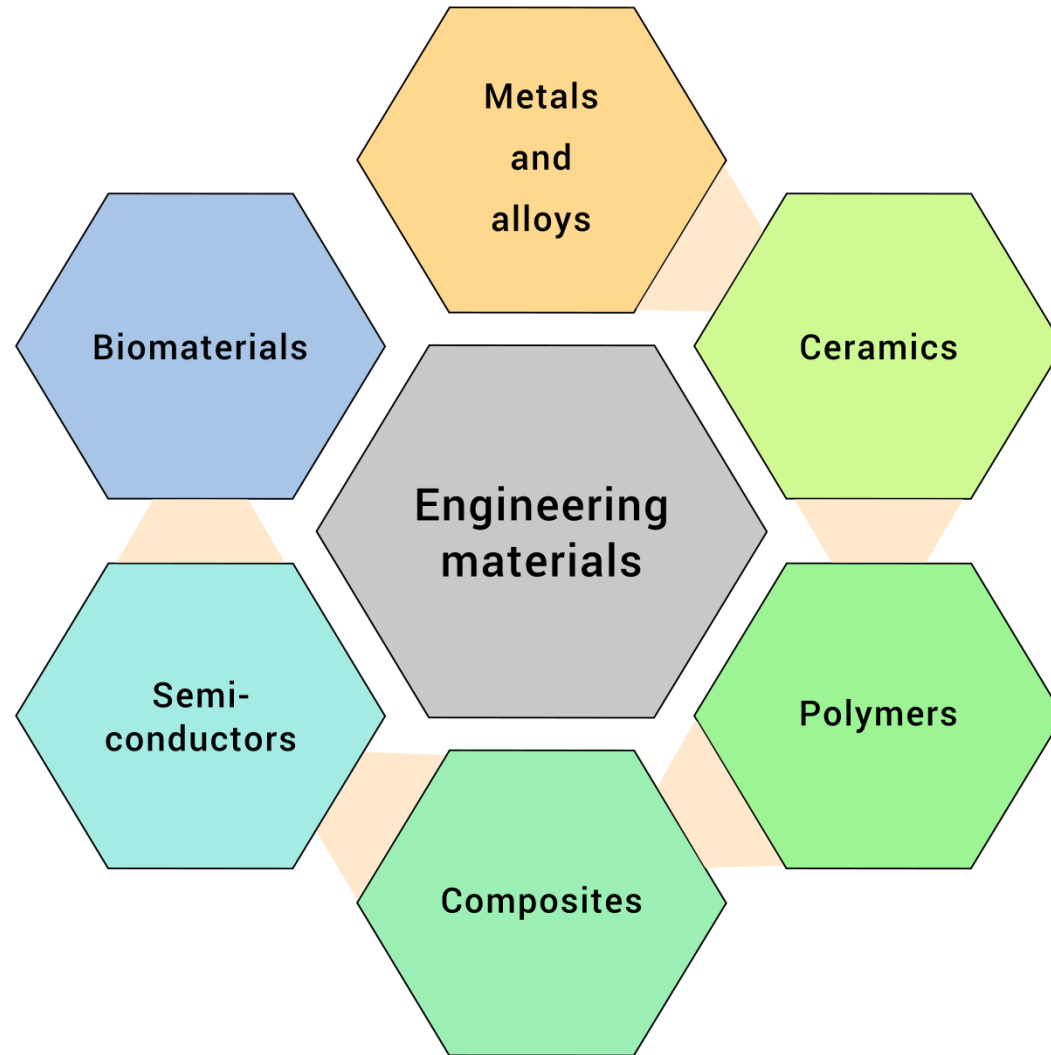
## Concrete

- About 60% of what is crushed can be used for downcycling processes (recovery of materials with a more limited range of uses than the original material).
- These fragments can be used as base materials for structures such as roads.

Recycling concrete



# Engineering materials categories (variable)

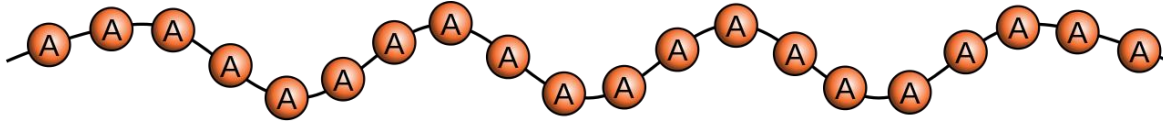


Focus: Plastics

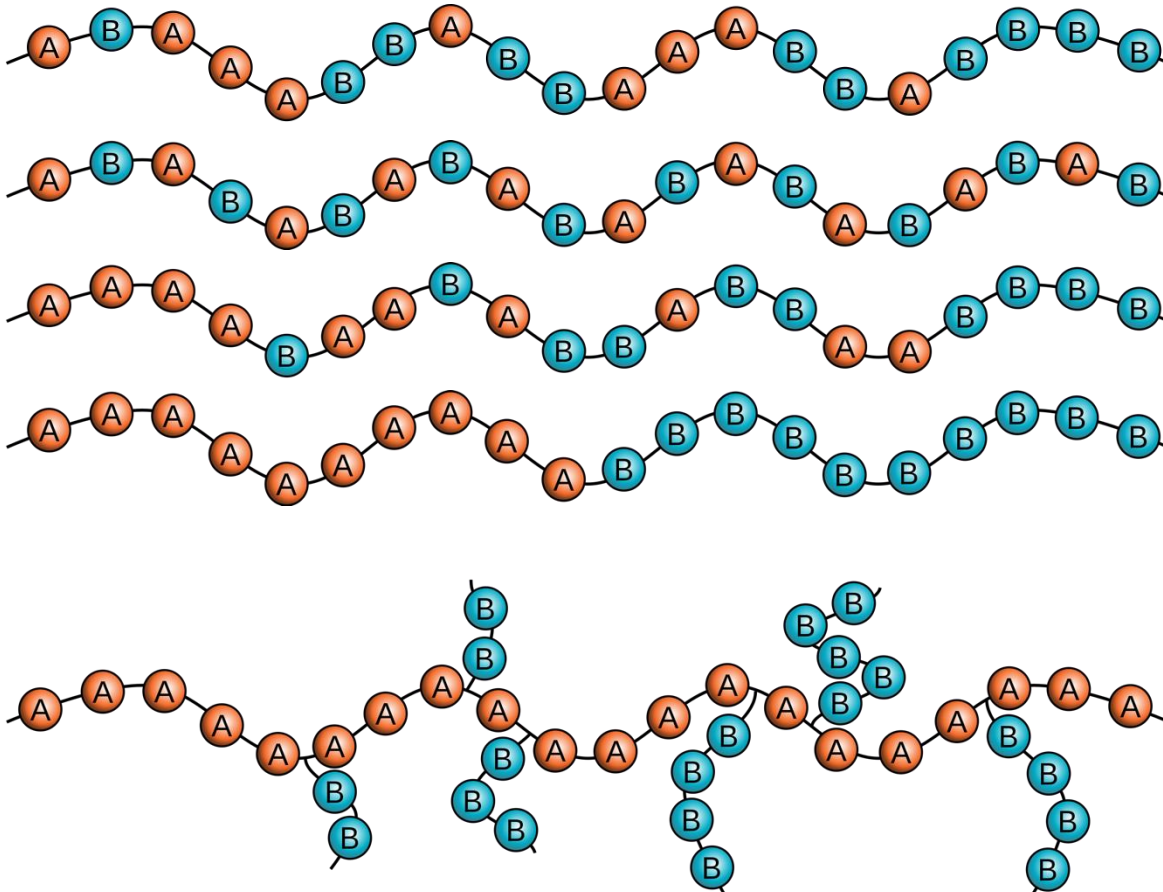


# Generic definition of a polymer

homopolymer



copolymer



- Covalently linked repeat units, building up into a long molecule, *aka* a macromolecule or a polymer
- Can be nature-derived, semi-synthetic or synthetic
- Often defined by chain length
- Versatile chemistry and chain length translates to versatile properties (crystallinity, viscosity, thermal, mechanical, etc.,)

By Minihaa - Own work, CC0,  
<https://commons.wikimedia.org/w/index.php?curid=37637259>

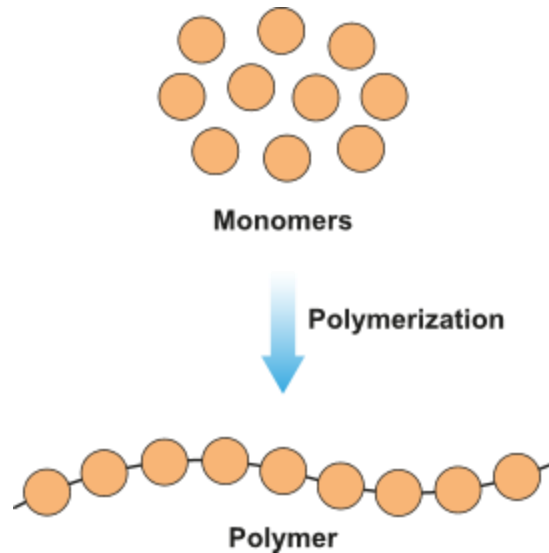
# Polymer definition

**Hermann Staudinger**  
"Father of Macromolecular Chemistry"  
1953 Chemistry Nobel Prize



**1920**

*"In a paper entitled "Über Polymerisation," Staudinger presented several reactions that form high molecular weight molecules by linking together a large number of small molecules. During this reaction, which he called "polymerization," individual repeating units are joined together by covalent bonds."*



*"...Heinrich Wieland, 1927 Nobel laureate in chemistry, wrote to Staudinger, "Dear colleague, drop the idea of large molecules; organic molecules with a **molecular weight higher than 5000 do not exist**. Purify your products, such as rubber, then they will crystallize and prove to be low molecular compounds!"*

*"My colleagues were very skeptical about this change, and those who knew my publications in the field of low molecular chemistry asked me why I was neglecting this interesting field and instead **was working on a very unpleasant field and poorly defined compounds, like rubber and synthetic polymers**. At that time the chemistry of these compounds often was designated, in view of their properties, as *Schmierenchemie* ('grease chemistry')."*

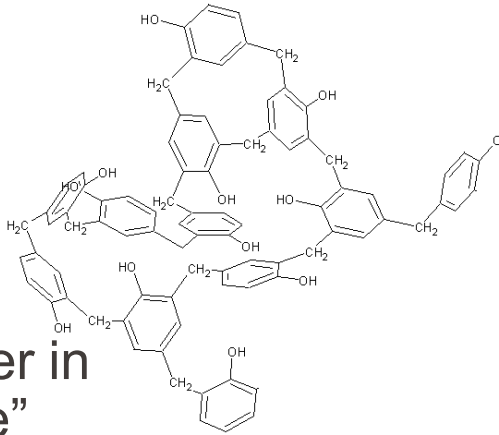
Staudinger Polymer Science

# Celluloid

- Late 19<sup>th</sup> century
- John Wesley Hyatt in response to \$10,000 reward for ivory substitute
- Celluloid = partially nitrated cellulose (“guncotton”) + camphor
- 1<sup>st</sup> semi-synthetic plastic
- Thermoplastic
- (Launching the age of man-made plastics)







- World's first fully synthetic polymer in 1907, heralding the “Polymer Age”
- “Insoluble in all solvents, does not soften. I call it Bakalite (sic)”
- Moldable
- Thermoset

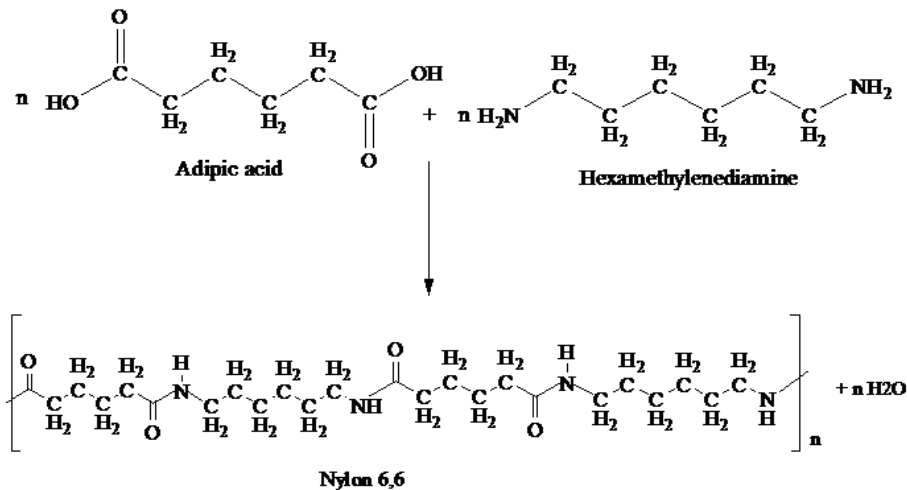


*“Von Baeyer had reported that when he mixed phenol, a common disinfectant, with formaldehyde, it formed a hard, insoluble material that ruined his laboratory equipment, because once formed, it could not be removed. Kleeburg reported a similar experience, describing the substance he produced as a hard amorphous mass, infusible and insoluble and thus of little use.”*



# Nylon

- Wallace Carothers at Dupont (US)
- Nylon developed in 1930s
- Condensation polymerization
- Thermoplastic
- High strength, rigidity, good heat and chemical resistance
- Highly used, e.g., automotive, textile fibers, etc.,



**PACK IT**—washable, quick-drying nylon man's robe! So light it takes up little room—so long-wearing you'll be packing it for many trips to come! Because nylon can be "heat-set" wrinkles and creases are no worry.

**WORK WITH IT!** In the water—out again—commercial laundry bags must be strong, light, extra long-wearing. And they are made with nylon! Tough, elastic nylon fibers have a high tensile strength when wet, are unaffected by detergents,

**WASH IT!** Hang it up! Your nylon blouse is almost ready to wear again. For nylon is that fast-drying fiber! Needs only a little wringing, can be set to hold its shape. Your blouse will keep its fresh, crisp and feminine look—wear and WEAR!

## news about NYLON

*it all started with a stocking*  
... a sheerest, liveliest, long-wearing stocking that women had ever dreamed of!

Today, you're choosing nylon hose as fashion-right, costume-bright, color-shed, resistant, out-possibly strong for their weight.

You're looking nylon over on sweaters, panties, slips, lace, gloves—on to leotards, from to hat's tough new and elastics in countless industrial uses.

Du Pont makes the nylon fibers used in the products shown. The manufacturers of these products use nylon because nylon products can have these outstanding properties:

- ✓ STRENGTH
- ✓ LIGHTNESS
- ✓ ELASTICITY
- ✓ TOUGHNESS
- ✓ LONG WEAR
- ✓ EASY WASHING
- ✓ FAST DRYING
- ✓ FLAME RESISTANCE
- ✓ RESISTANCE TO MOTHS AND PERSPIRATION
- ✓ CAN BE "SET" TO HOLD SHAPE

**DU PONT**  
REG. U.S. PAT. OFF.

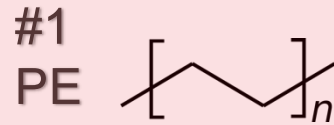
BETTER THINGS FOR BETTER LIVING  
... THROUGH CHEMISTRY

TO MANUFACTURERS: Are there textile fibers in your product? Then you'll want to read "Nylon Textile Fibers in Industry." Write for this important book now!

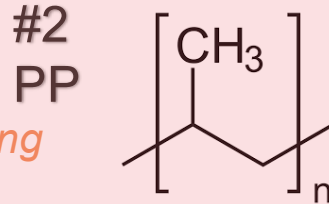
NEW! FREE book for women, teachers, students—  
"About Du Pont Nylon." Write to Nylon Division, E. I. du Pont de Nemours & Co. (Inc.), Wilmington 98, Del.

FOR NYLON FOR RAYON... FOR FIBERS TO COME... LOOK TO DU PONT

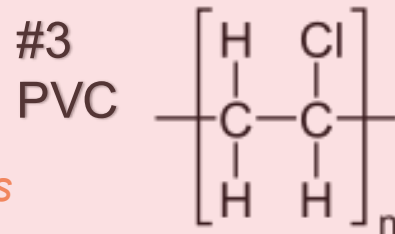
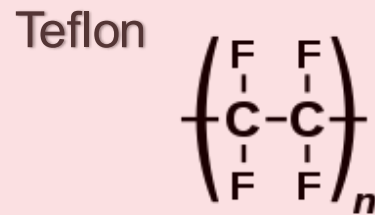
## ADDITION POLYMERS



packaging



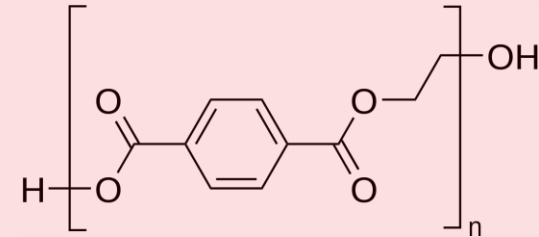
packaging



pipes

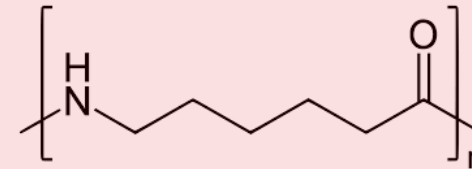
## CONDENSATION POLYMERS

PET



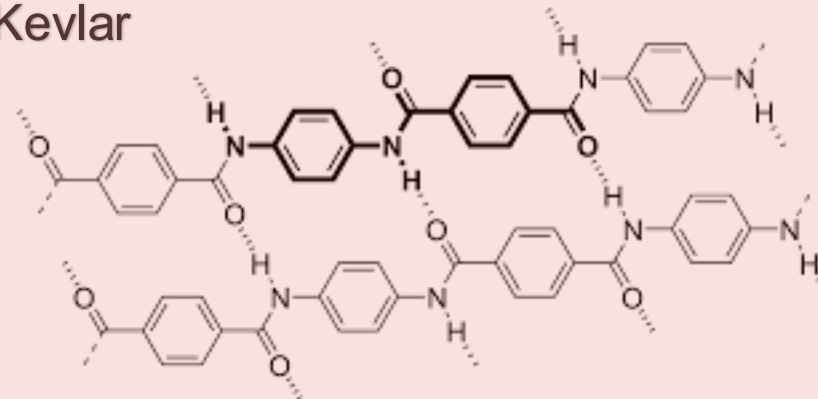
Polyester  
Carbonated drinks

Nylon 6,6



Polyamide  
Textiles

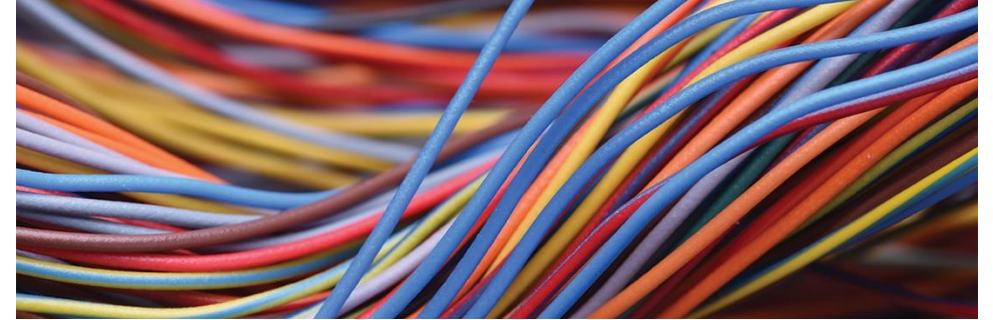
Kevlar



aromatic polyamide  
Engineering materials

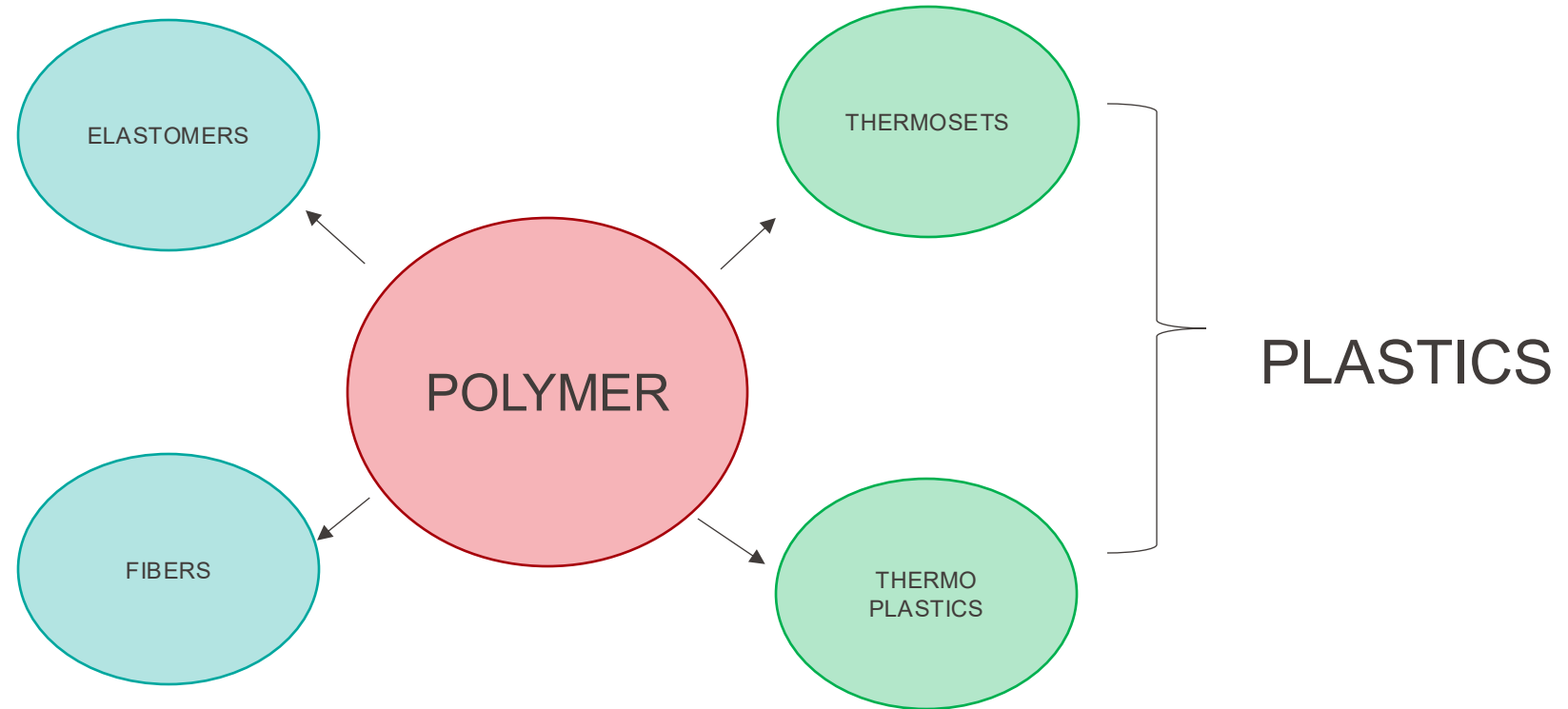
# General characteristics of plastics

- Cheap & versatile
- Excellent strength to weight ratio
- Highly processible
- Non-reactive – inert, not easily corroded, BUT can be degraded and broken down into microplastics by UV-light, abrasion
- Good insulators
- Tailorable appearance and properties
- Life cycle & end of life complex for synthetic polymers



# Polymer vs. plastic

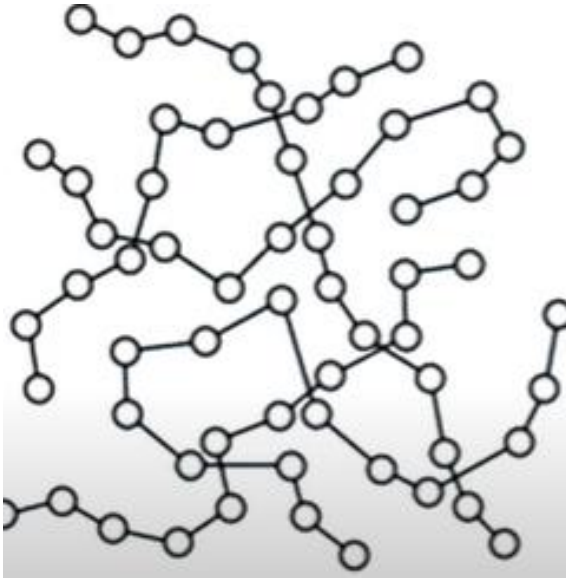
- *Polymer is from Greek for “of many parts”*
- *Plastic is from “Plastikos”, Greek for pliable*





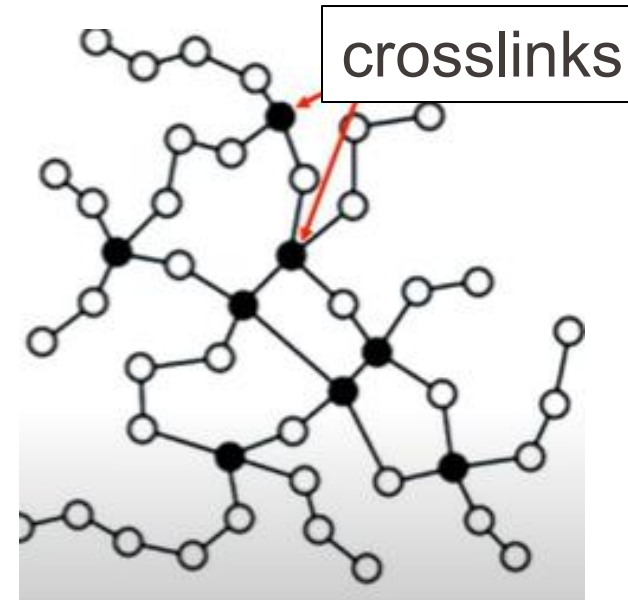
# Thermoplastics vs. Thermosets

## THERMOPLASTICS



- Weak intermolecular bonds
- Softens when heated
- Hardens when cooled
- Reversible

## THERMOSETS



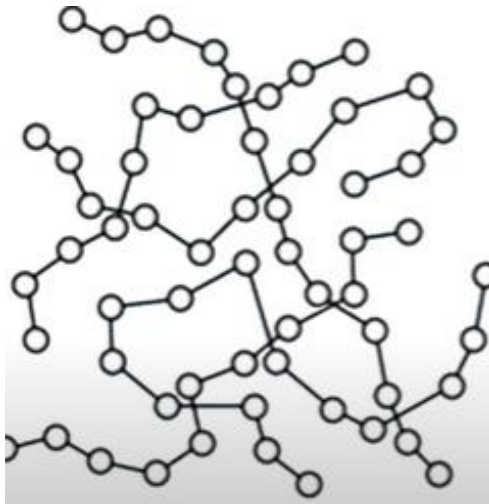
- Strong covalent bonds (“crosslinks”)
- Rigid, prevents melting when heated
- Cannot be reshaped or remolded

[thermoplastics vs thermosets](https://www.youtube.com/watch?v=4fTtrKPySm0)

<https://www.youtube.com/watch?v=4fTtrKPySm0>

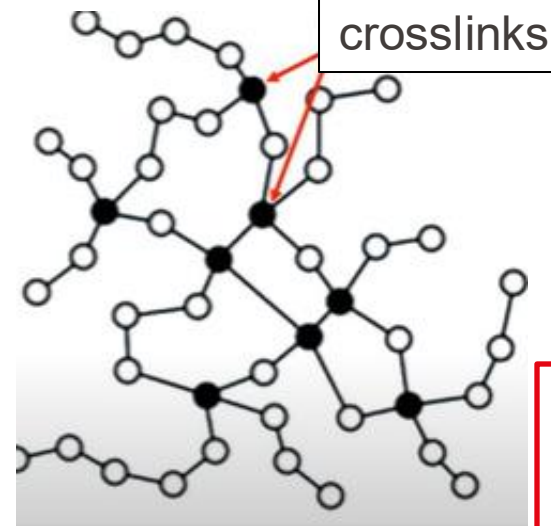
# Thermoplastics vs. thermosets

## THERMOPLASTICS



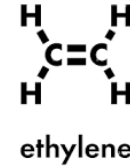
- Remoldable
- Recyclable
- High strength
- Shrink resistance
- Flexibility

## THERMOSETS

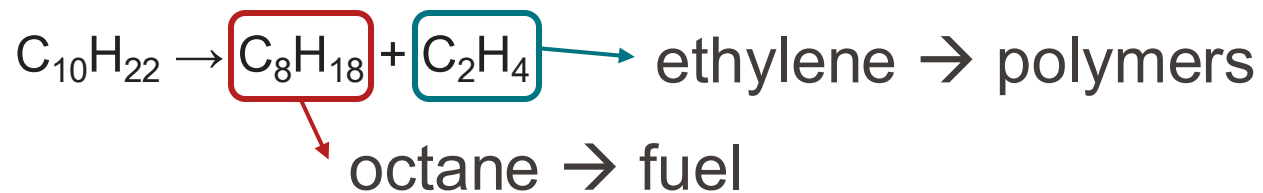


- Heat resistant
- Corrosion resistant
- High strength
- Mechanical creep resistance
- High specific strength (strength to weight ratio)

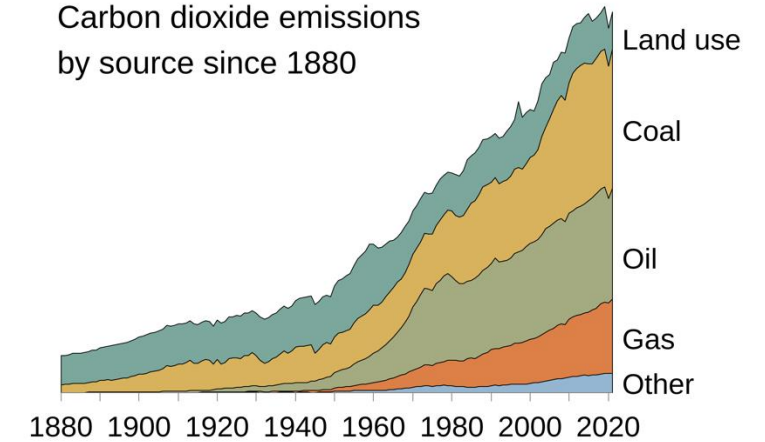
# Plastic resources



- Most plastic in use today comes from hydrocarbons derived from crude oil, natural gas, and coal – fossil fuels
- Crude oil is a combination of hundreds of different hydrocarbons
- Fractional distillation is used to separate them into their various components
- Higher molecular weight hydrocarbons are then selected and further processed (cracking) to yield lower molecular weight alkenes and alkanes (e.g., decane cracking)

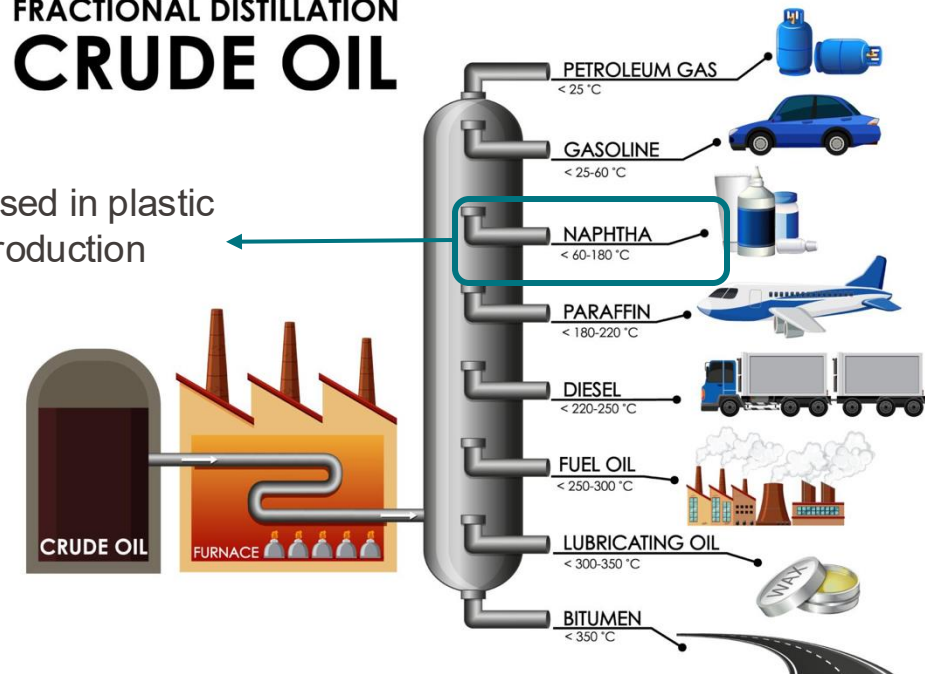


Carbon dioxide emissions  
by source since 1880



## FRACTIONAL DISTILLATION CRUDE OIL

Used in plastic  
production

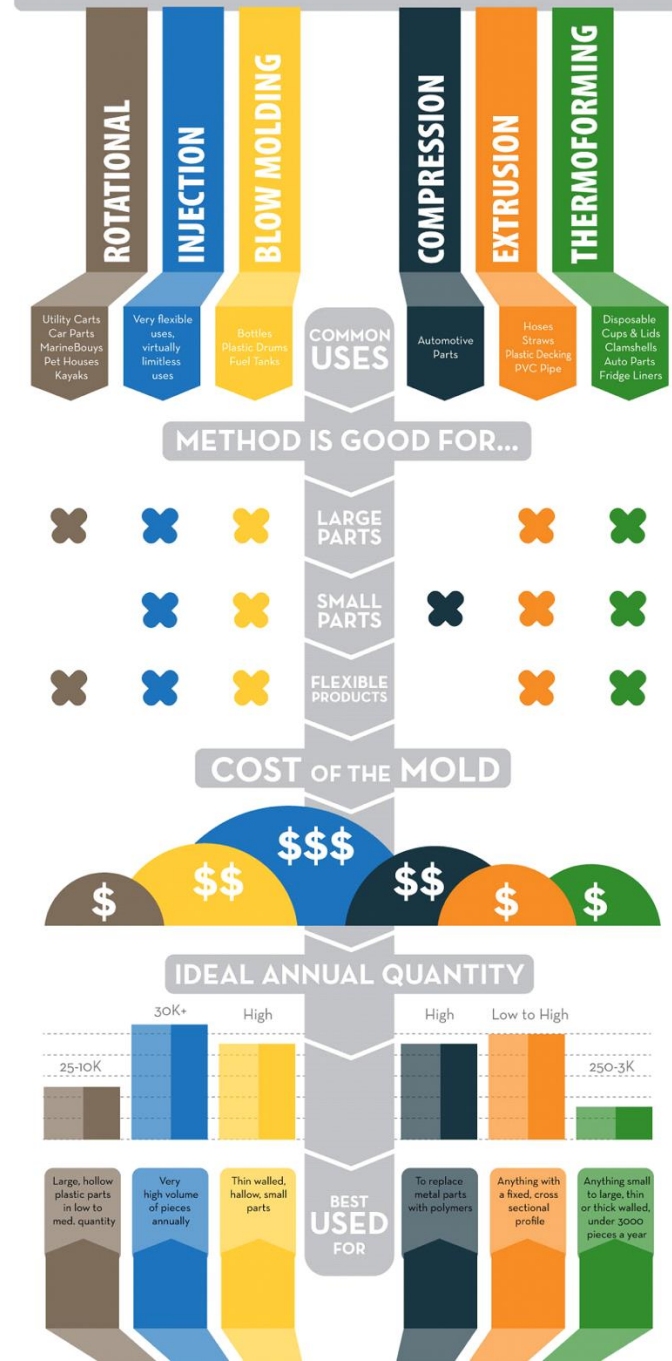


# Plastic processing

- Rotational molding
- Injection molding
- Blow molding
- Compression molding
- Extrusion molding
- Thermoforming



## PLASTIC MOLDING COMPARISON







**DU PONT**

REG. U.S. PAT. OFF.

BETTER THINGS FOR BETTER LIVING  
... THROUGH CHEMISTRY

- Neoprene
- Nylon
- Teflon
- Lycra
- Kevlar

Lucia





# Plasticene

*A NEW GEOLOGICAL AGE...(last 60-70 years!)*

n. & adj. (2011) an era in Earth's history, within the Anthropocene, commencing in the 1950s, marked stratigraphically in the depositional record by a new and increasing layer of plastic ([Stager, 2011](#), attributed to Matt Dowling). The history and etymology of plasticene in this sense is not related to the word *plasticene*, a common early spelling (for example, [Cooper, 1901](#)) of the popular molding clay, plasticine.



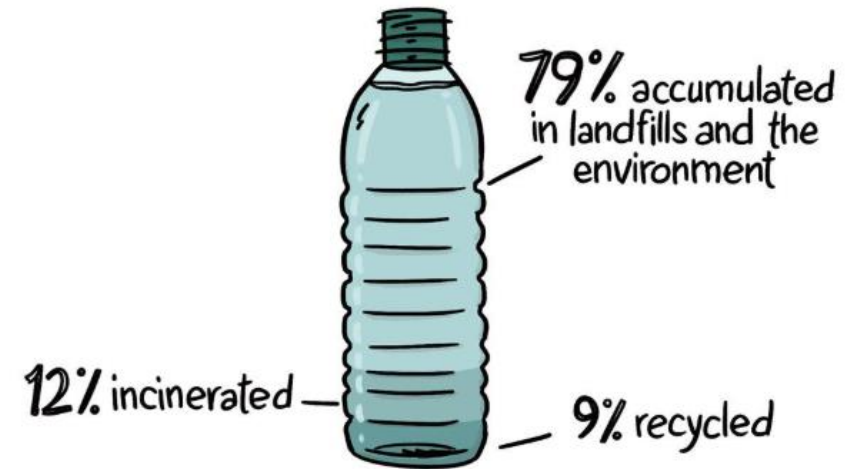
Plastics everywhere

# Plastic promises: How'd we do?

Data from 1950-2015:

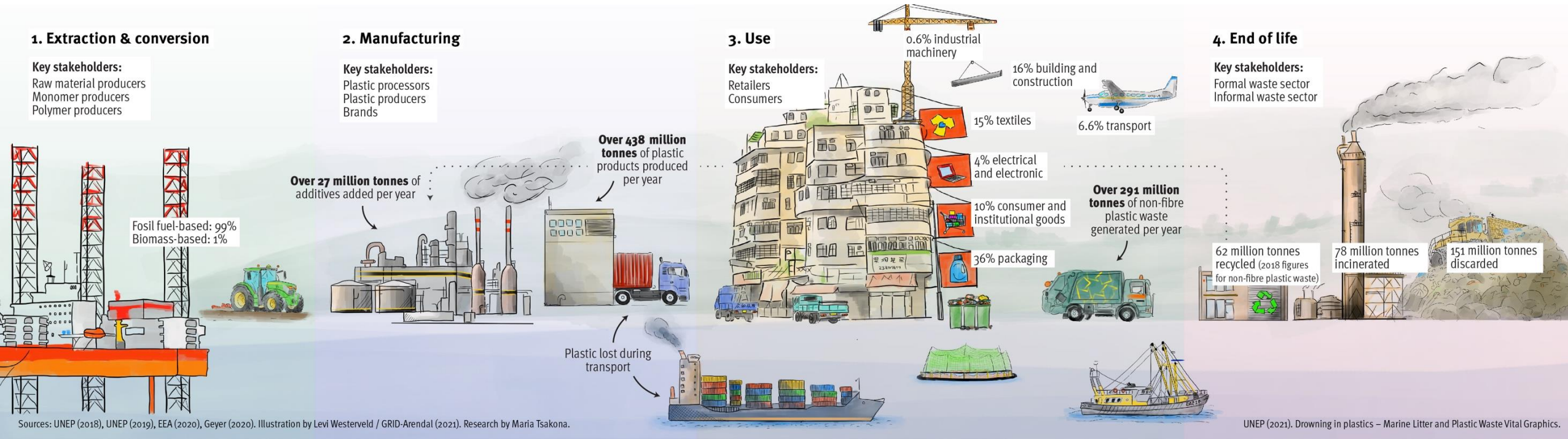
- 8300 Mt of plastics
- 6300 Mt of plastic waste
- 12000 Mt in landfill/nature projected by 2050
- Conventional plastics do not biodegrade
- Increased demand for plastics
- Permanent elimination only by incineration

Only about 30% of plastic ever produced is still in use — the rest has been disposed of in one of three ways:



Geyer R. et al., Science Advances 2017, 3(7), 1-5.





- Of the 438 million tons of plastic produced every year, only about 14% is recycled
- A total of 99% of feedstock for plastic production is fossil fuel-based, accounting for around 8-9% of global oil and gas consumption



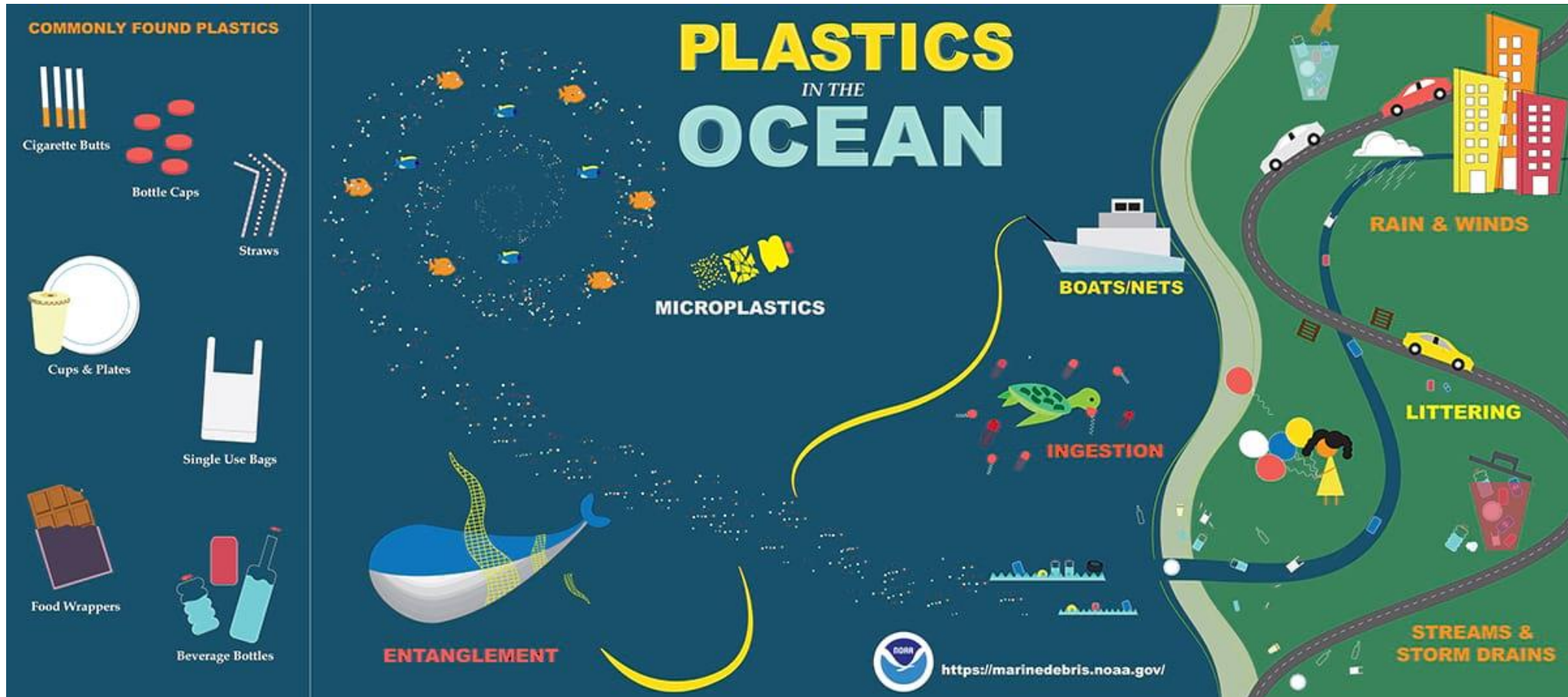
*The New York Times*

STEM WRITING CONTEST WINNER

## *An Unexpected Dinner Guest: Marine Plastic Pollution Hides a Neurological Toxin in Our Food*

We are honoring each of the top eight winners of our Student STEM Writing Contest by publishing their essays. This one is by Vivian Li.





- 80% of marine pollution is plastic, 8-10 Mt of plastic/year
- Essentially all manmade plastics are still in existence (except what has been incinerated, approx. 9%)
- Plastics either break down into microplastics or form garbage patches





# The Lifecycle of Plastics

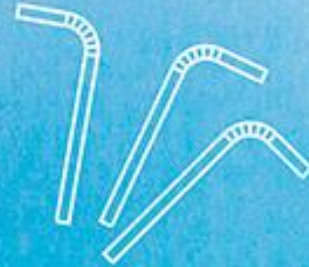
- 500-1000 years degradation time
- Even then, it becomes microplastics.
- **50-75 trillion pieces of plastic** and microplastics in the ocean.



Plastic bag  
**20 years**



Coffee cup  
**30 years**



Plastic straw  
**200 years**



6-pack plastic rings  
**400 years**



Plastic water bottle  
**450 years**



Coffee pod  
**500 years**



Plastic cup  
**450 years**



Disposable diaper  
**500 years**



Plastic toothbrush  
**500 years**

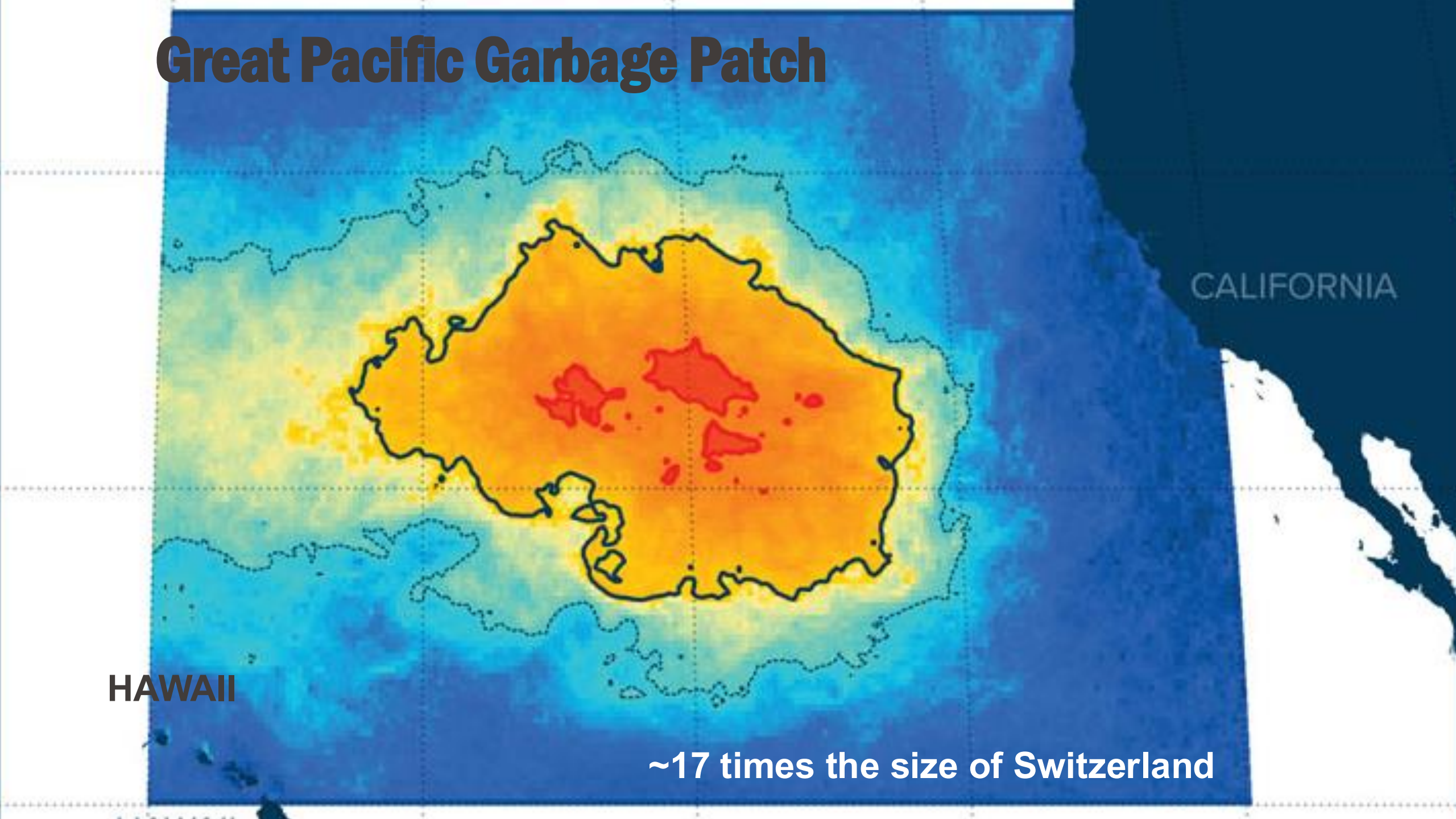


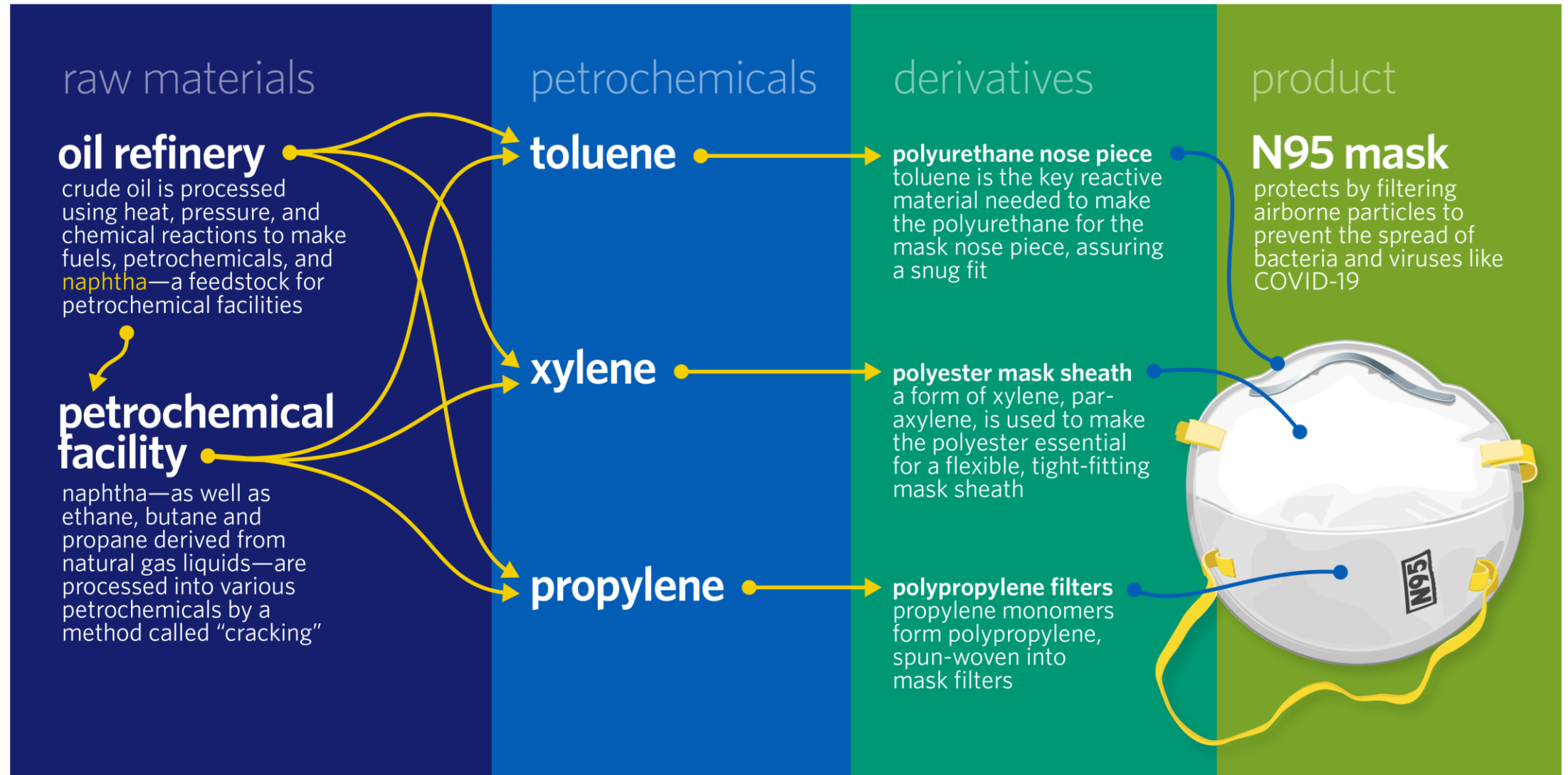
# Great Pacific Garbage Patch

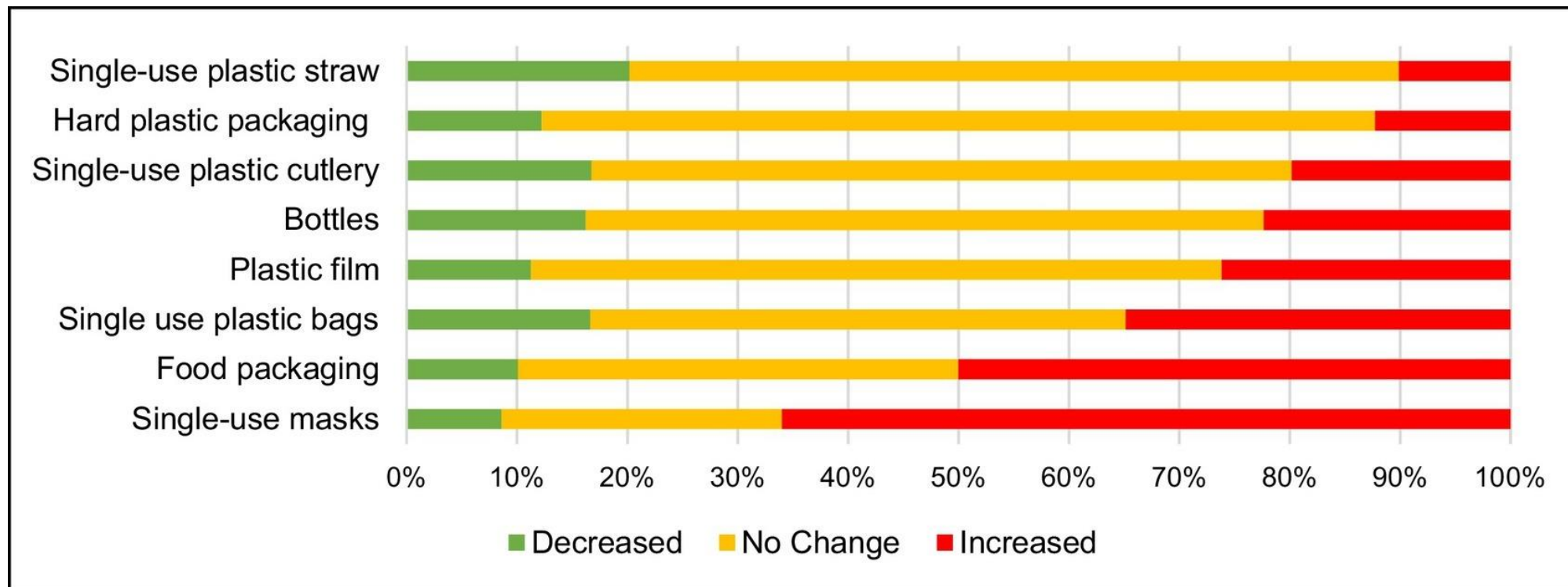
HAWAII

CALIFORNIA

~17 times the size of Switzerland





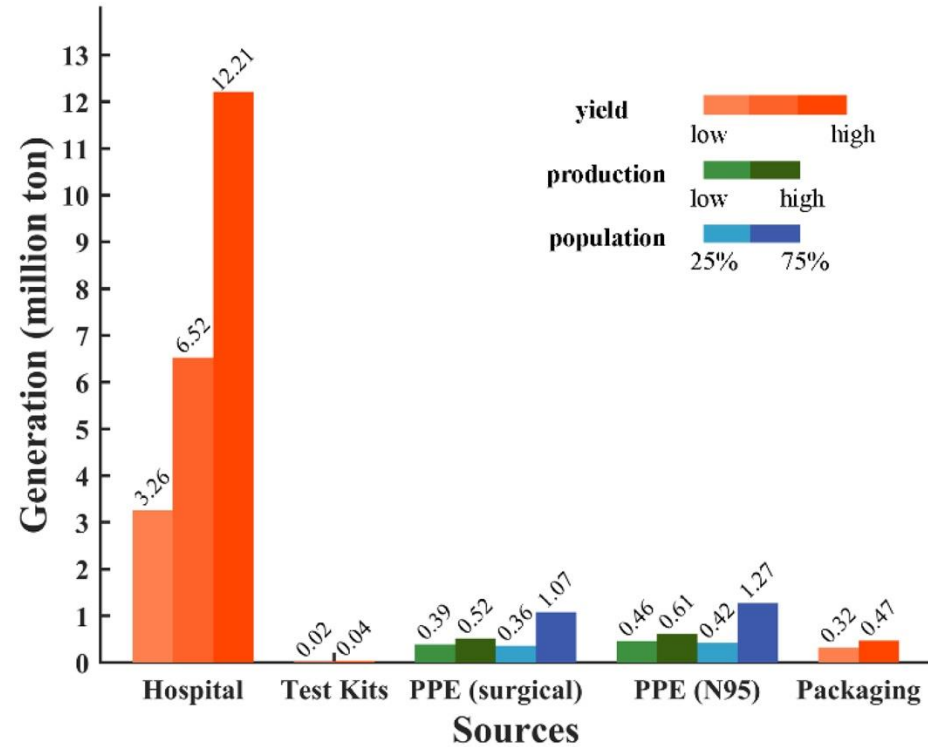


[SUPs in the Pandemic](#)

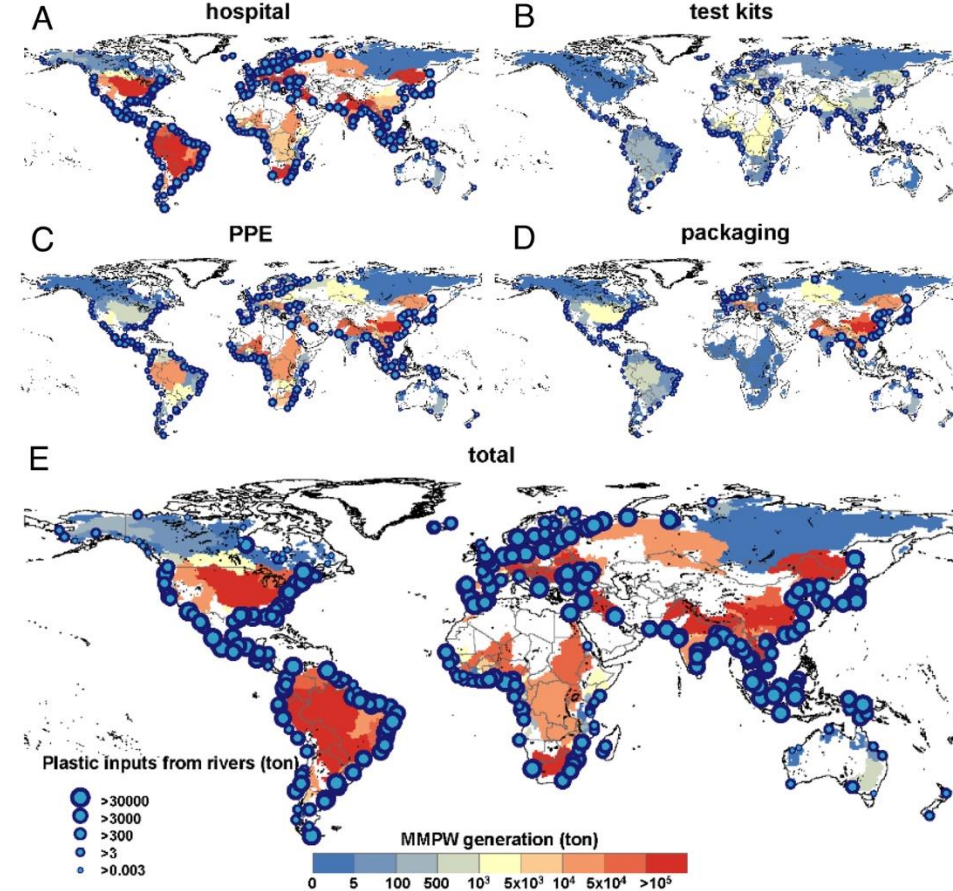
\*Based on 202 respondents from 41 countries



# Pandemic plastics in marine environment



- Excess mismanaged plastic waste (MMPW) 4.4-15.1 million tons (8.4 million tons best estimate)
- 87% from hospitals, 8% PPE by individuals, packaging 4.7%, test kits 0.3%



- Top river discharges from – Shatt al Arab, Indus, Yangtze River, Ganges, Danube, Amur
- 73% of discharge is from Asia, 11% from Europe



**Persistent and ubiquitous**

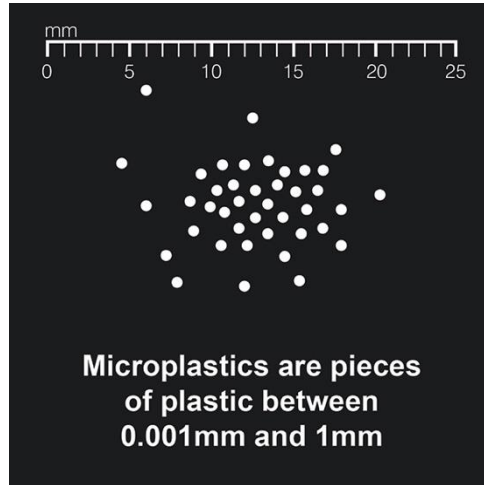




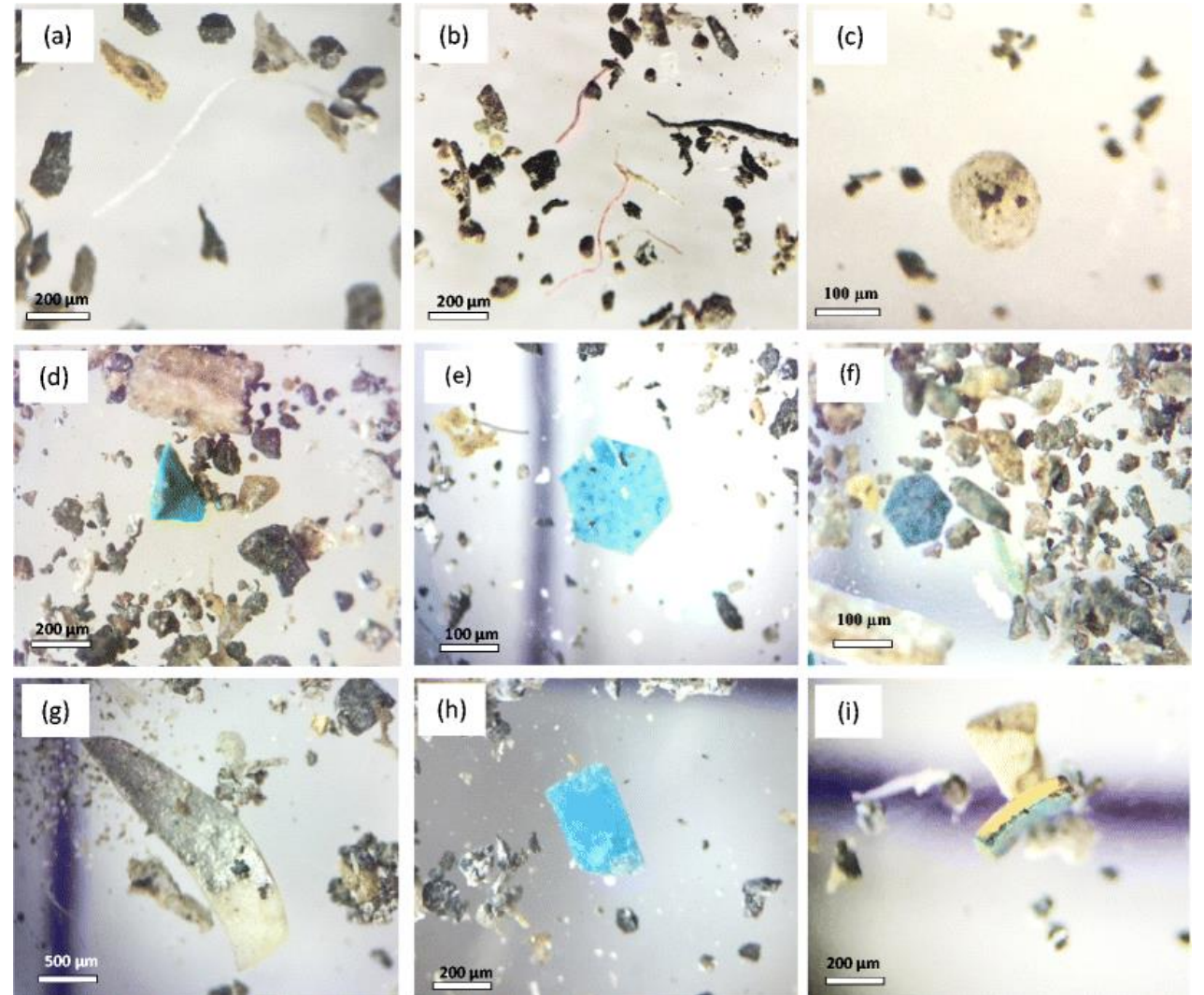


*This collage of Arctic Ocean plastic, assembled by Andres Cozar, the lead author of a 2017 study "The Arctic Ocean as a dead end for floating plastics in the North Atlantic branch of the Thermohaline Circulation," and his colleagues, shows just some of the estimated billions of pieces of plastic in ice-free Arctic waters. (Nunatsiaq News file photo)*

# Microplastics



1. Plastic litter
2. From washing synthetic clothes
3. Cosmetics
4. Toothpaste
5. Abandoned and Derelict vessels
6. Car tires and rubbers
7. Scrubbing agents in toiletries



Microplastics in urban dust, Tehran

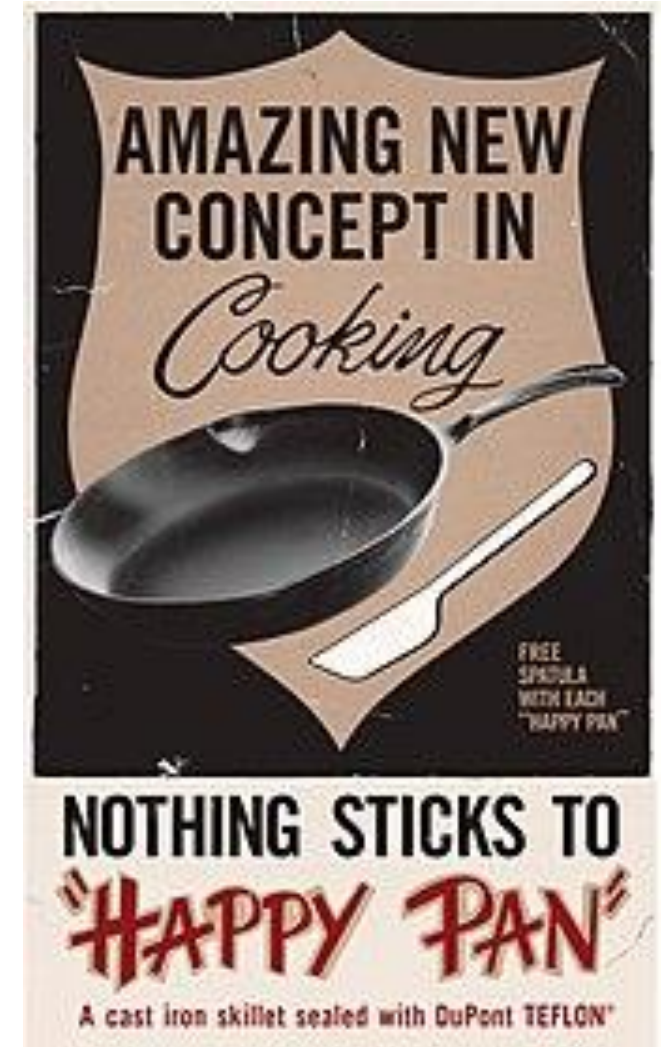


# What went wrong?

- Perception of unlimited resources
- Plastics are largely stable and inert, e.g., PFAS “forever chemicals”
- Complicated EoL due to versatility in chemistry and function
- Mismanaged waste
- Limited recycling infrastructure
- Limited technologies for sorting and separation
- Global movement of microplastics
- Shipping trash may not be the answer (see China’s ban on plastic import)
- Hard to compete! Plastic is cheap, highly functional, growing in demand!



Easy, drop-in solution probably not possible





# Packaging...

# Next week!

Problematic because  
of plastics &  
multimaterials, high  
volume, short life  
span, and recycling  
challenges



# Key take aways

- Ceramics: main impacts related to mining, processing, and difficulties in recycling
- Plastics: it's complicated...
- Next week: Packaging and guest lecture on Sustainability at EPFL (what's the carbon footprint of a lab?)
- May 14: Glencore activity (please prioritize attendance – will not be recorded)
- May 21 and 30: Final project presentations! Summer in sight!