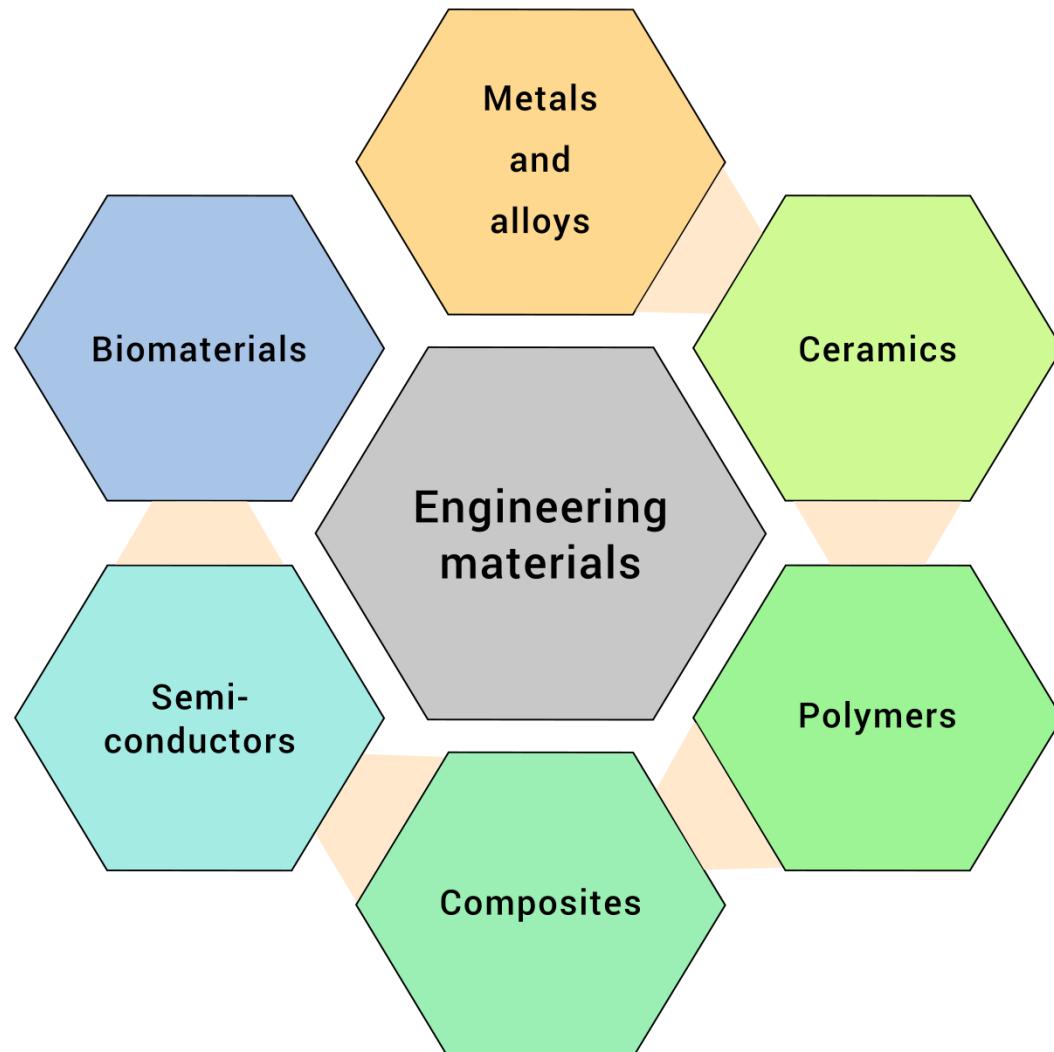




Sustainability & Materials

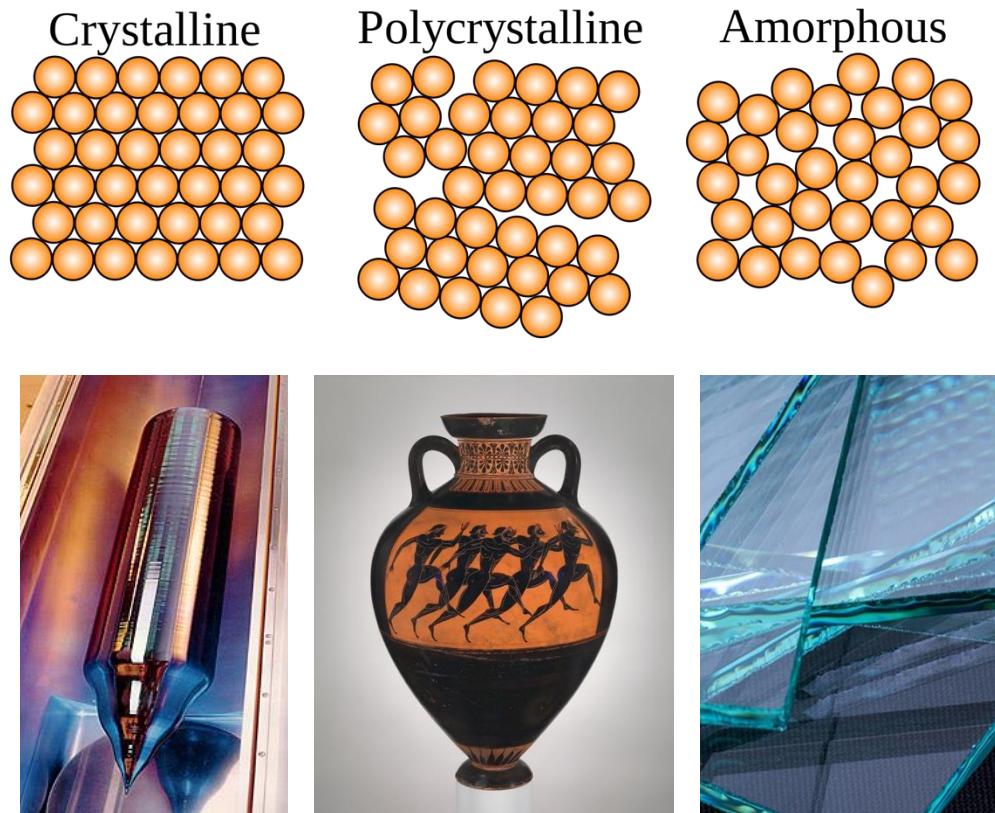
Prof. Tiffany Abitbol
2025

Engineering materials categories (variable)



Focus: Ceramics

- Non-metallic
- Inorganic
- Ionic and/or covalent bonds
- Mostly made by heat processing: heating and cooling
- Crystalline or semicrystalline except glass, which is amorphous
- Examples: alumina, silica, titania, boron nitride, etc.,



Alkalies		Alkaline Earths	
lithium 3 Li 6.941	boron 5 B 10.811	beryllium 4 B 12.011	hydrogen 1 H 1.008
sodium 11 Na 22.990	magnesium 12 Mg 24.305	aluminium 13 Al 26.982	nitrogen 7 N 14.007
potassium 19 K 39.098	calcium 20 Ca 40.078	silicon 14 Si 28.086	oxygen 8 O 15.999
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	gallium 31 Ga 69.723	fluorine 9 F 18.998
caesium 55 Cs 132.91	barium 56 Ba 137.33	germanium 32 Ge 72.64	

Partial Periodic Table

Elements that form the most common oxides
delivered to glazes by ceramic materials

Note: Classification is according to principle characteristic,
many oxides also act in other ways (e.g. colorants as fluxes, opacifiers as color modifiers).

Colorants								
titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.38
scandium 21 Sc 44.956	niobium 41 Nb 92.906	molybdenum 42 Mo 95.96	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.87	cadmium 48 Cd 112.41
zirconium 40 Zr 91.224	yttrium 39 Y 88.906	yttrium 41 Yb 187.906	niobium 41 Nb 92.906	molybdenum 42 Mo 95.96	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42
hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Ru 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.97	mercury 80 Hg 200.59

Opacifiers

Fluxes

Glass
Formers

Al₂O₃ is an
intermediate
(second most
common in glazes)

tin 50 Sn 118.71	indium 49 In 114.82
lead 82 Pb 207.2	thallium 81 Tl 204.38

Super Fluxes

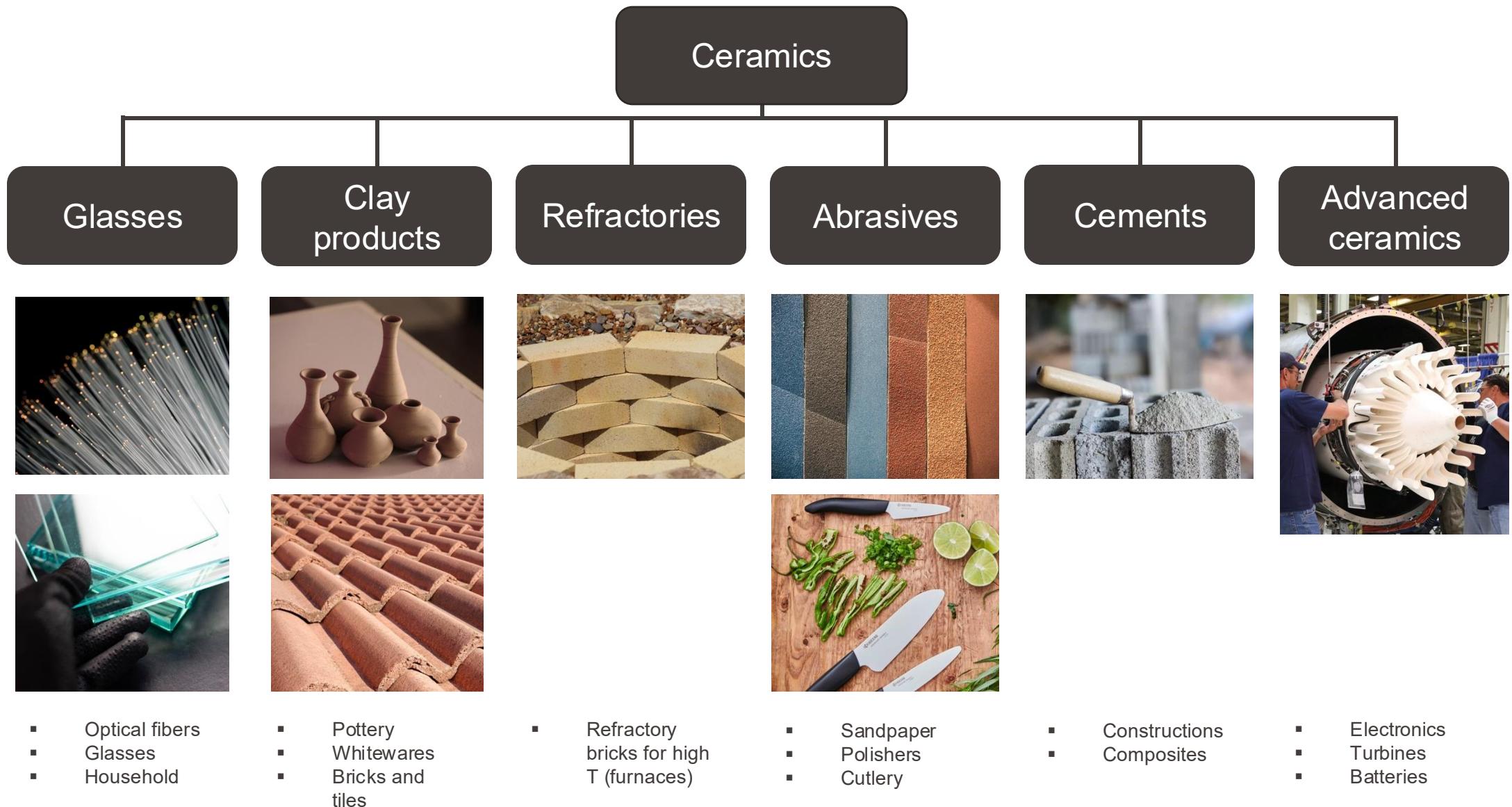
- Considered one of the greatest and earliest successes of humankind – control of fire
- Among the 1st objects manufactured
- From Greek “keramos” meaning burned earth
- Characteristics – long service life, low density, chemically inert, corrosion resistant, electromagnetic response, non-toxic, heat and fire resistant, sometimes electrical resistance or porosity



- 1st pieces reported ca. 24,000 years ago
- Many uses today: promising for aerospace and high temperature structural applications, information storage and optical devices, oral prosthetics, water purification, bone void fillers, CO₂ adsorbents, etc.,
- Value of global ceramics market was approx. 230 billion in 2018, and growing due to constant growth in the construction industry, technological advancements in nanotechnology, 3D printing, and ceramics in health

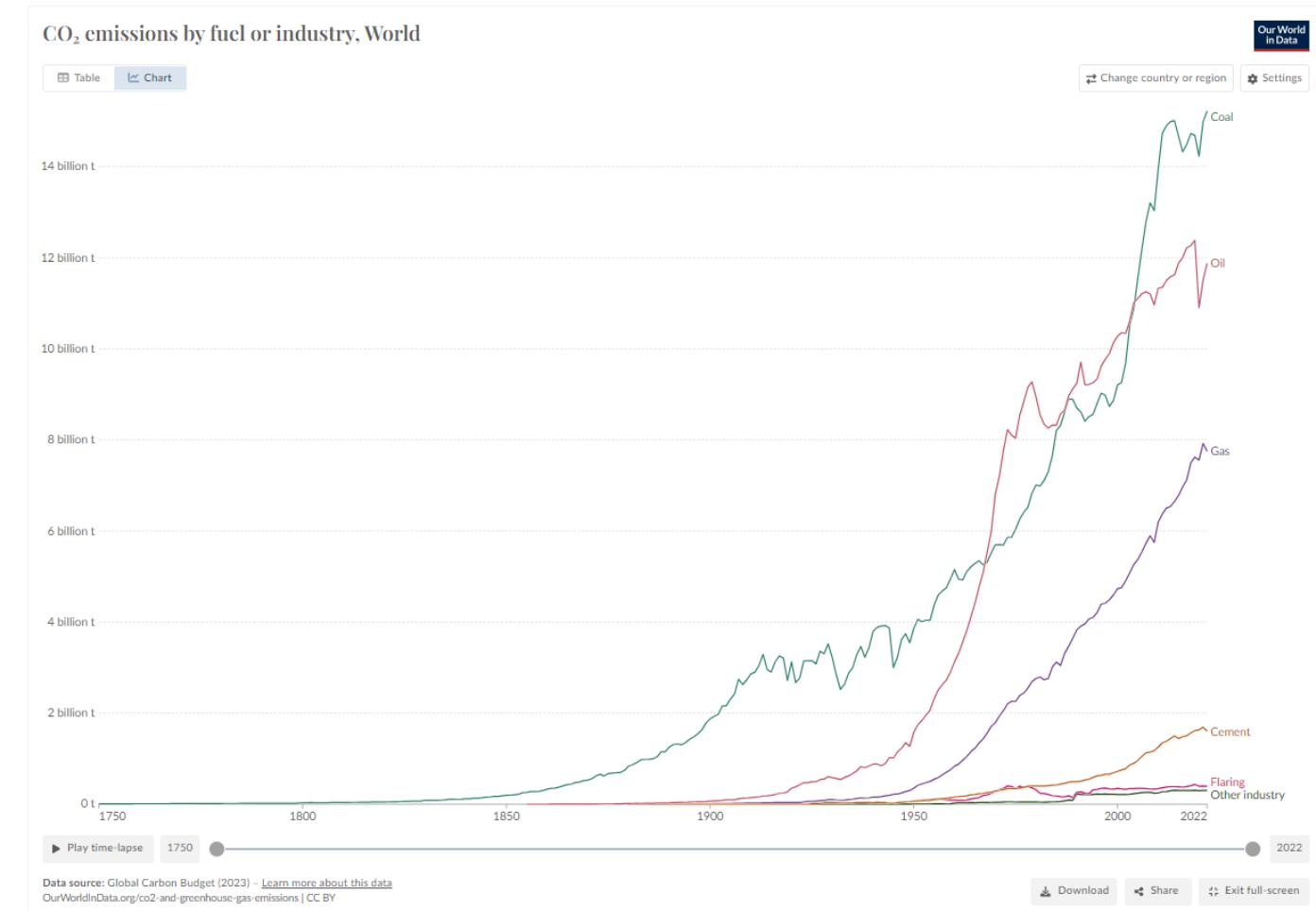


Main uses – Ceramics



Environmental cost of ceramics

- In EU: production of refractories, wall and floor tiles, and bricks and roof tiles emit around 19 Mt CO₂
- Globally: brick manufacture is responsible for 2.7% of annual carbon emission; cement (considered a ceramic in this course's context) is responsible for 5-8% of annual emissions



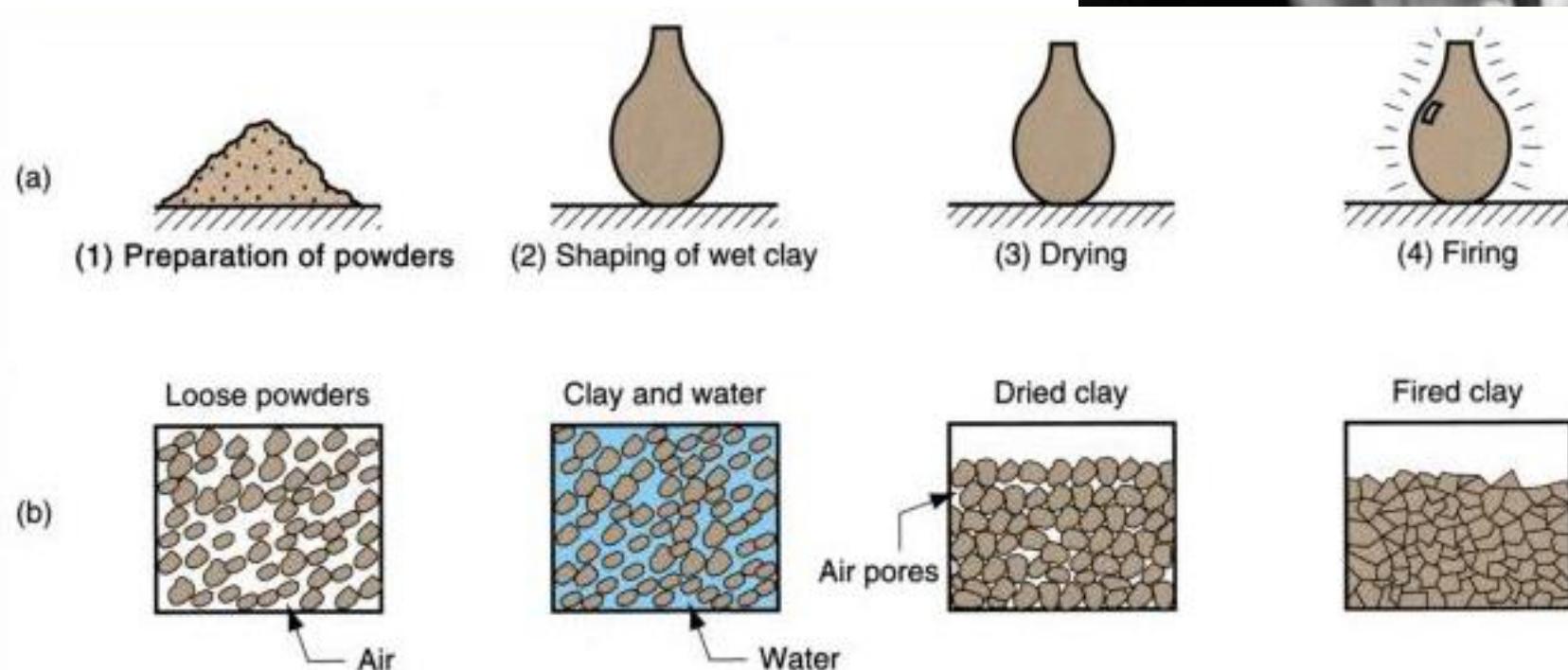
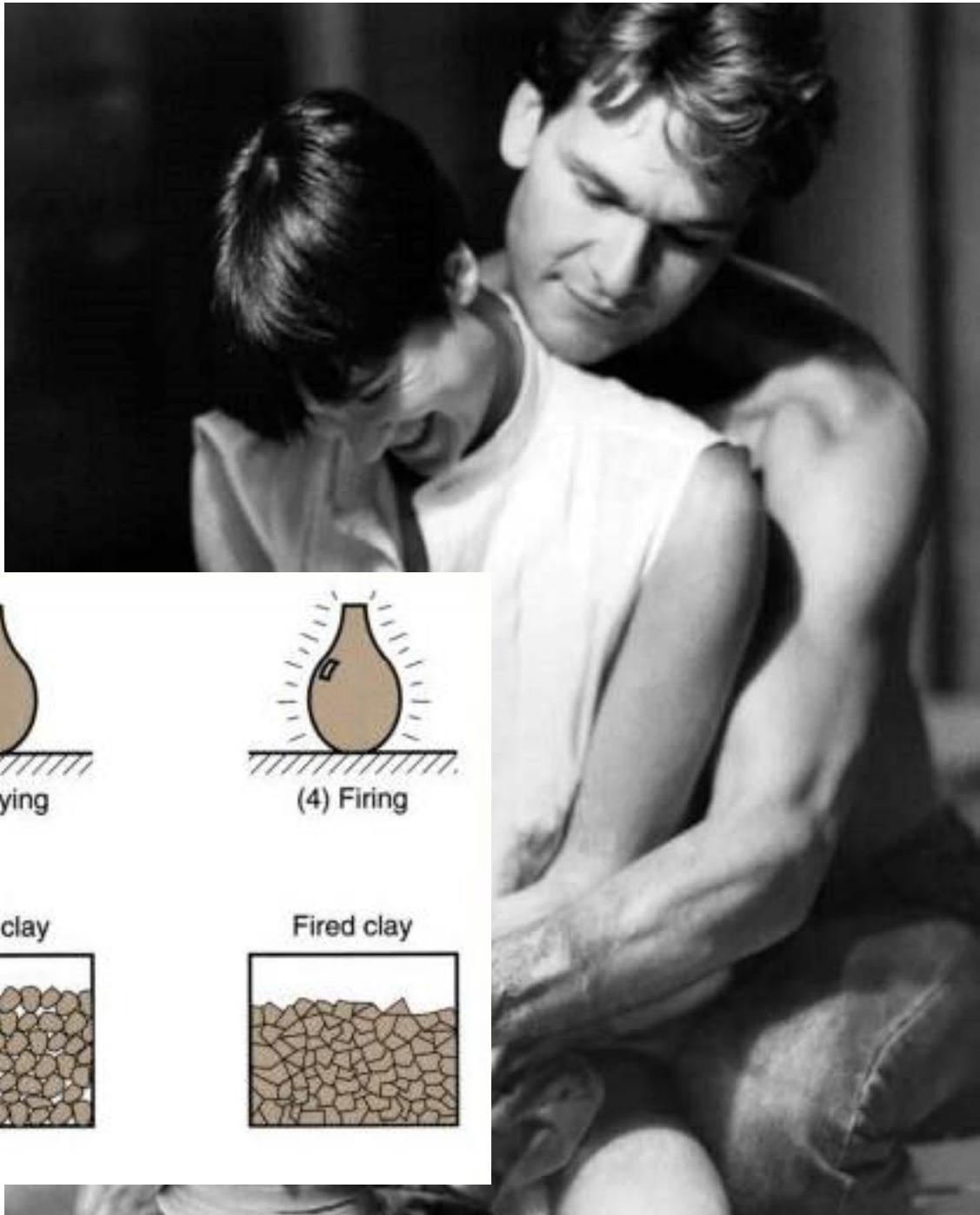
Decarbonizing ceramics_2022

Projecting future carbon emissions from cement_2023

Traditional ceramics

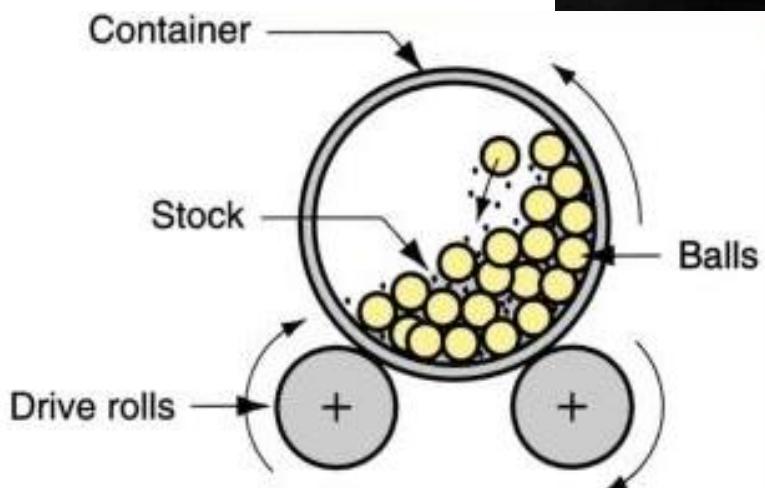
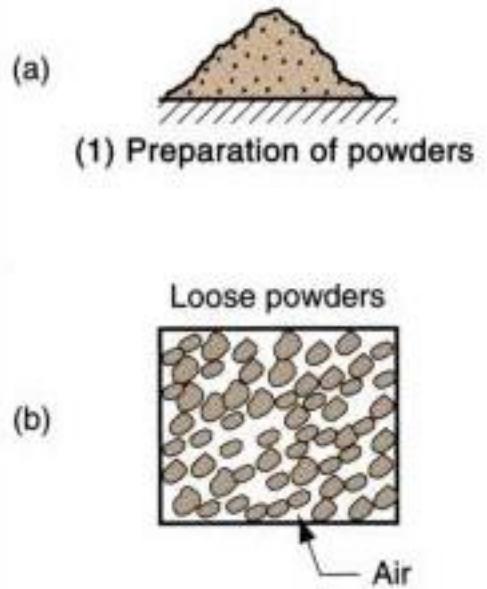
Processing ceramics

- Traditional ceramics are made from naturally occurring minerals
- Mineral extraction has the usual consequences: habitat destruction, soil erosion, water pollution
- Open pits and quarries common



Powders

Processing ceramics

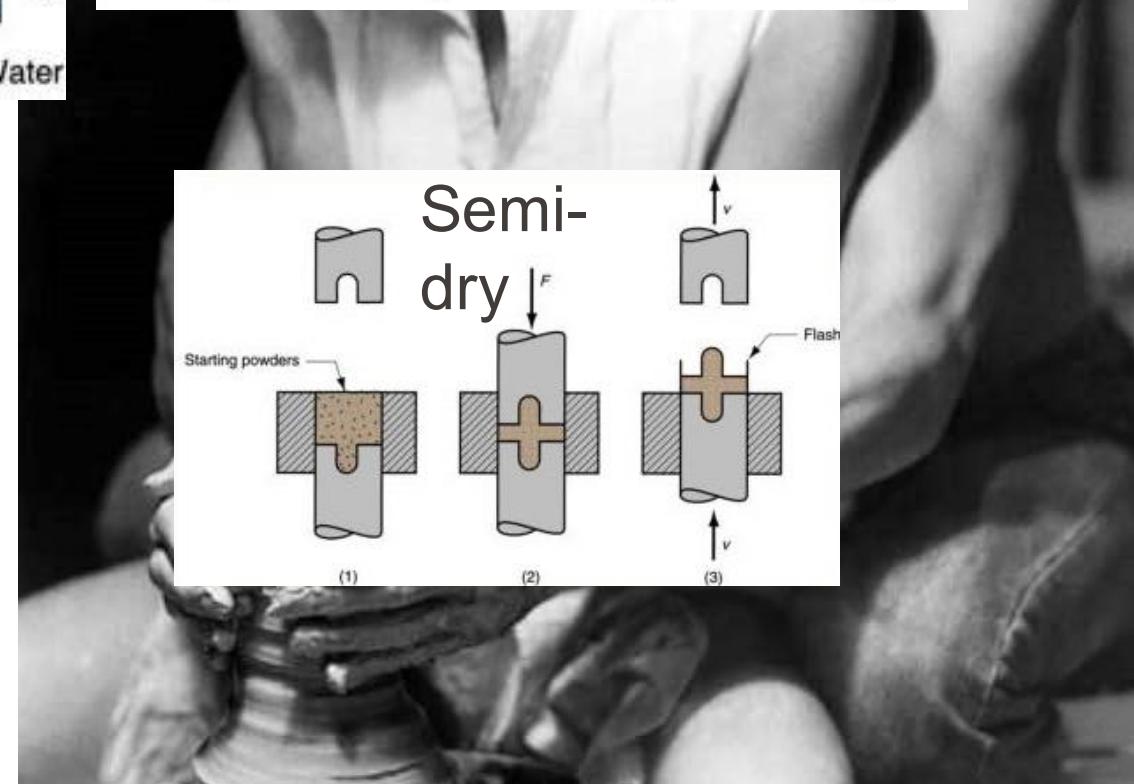
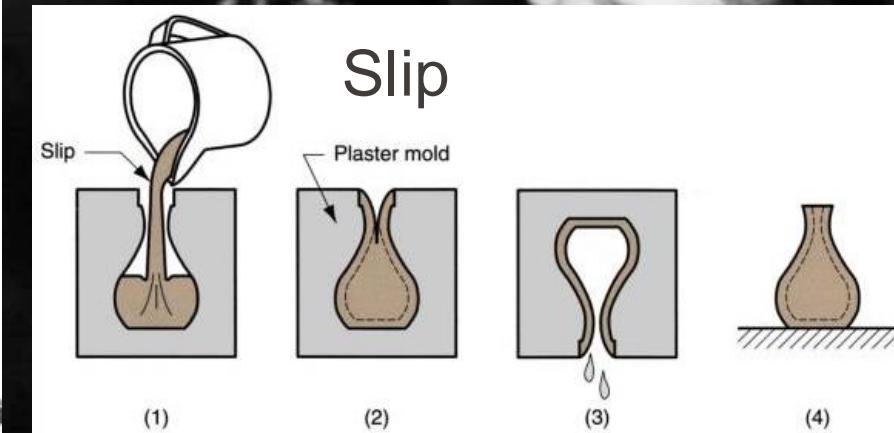
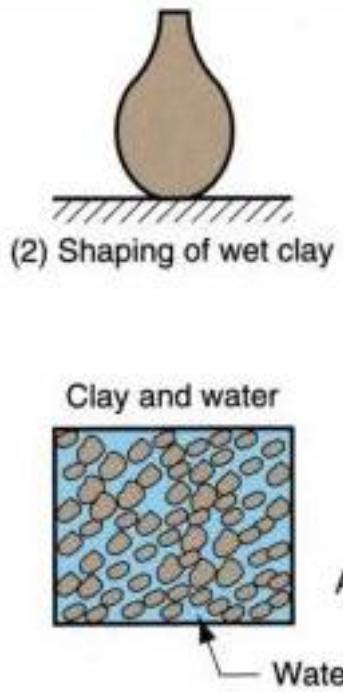
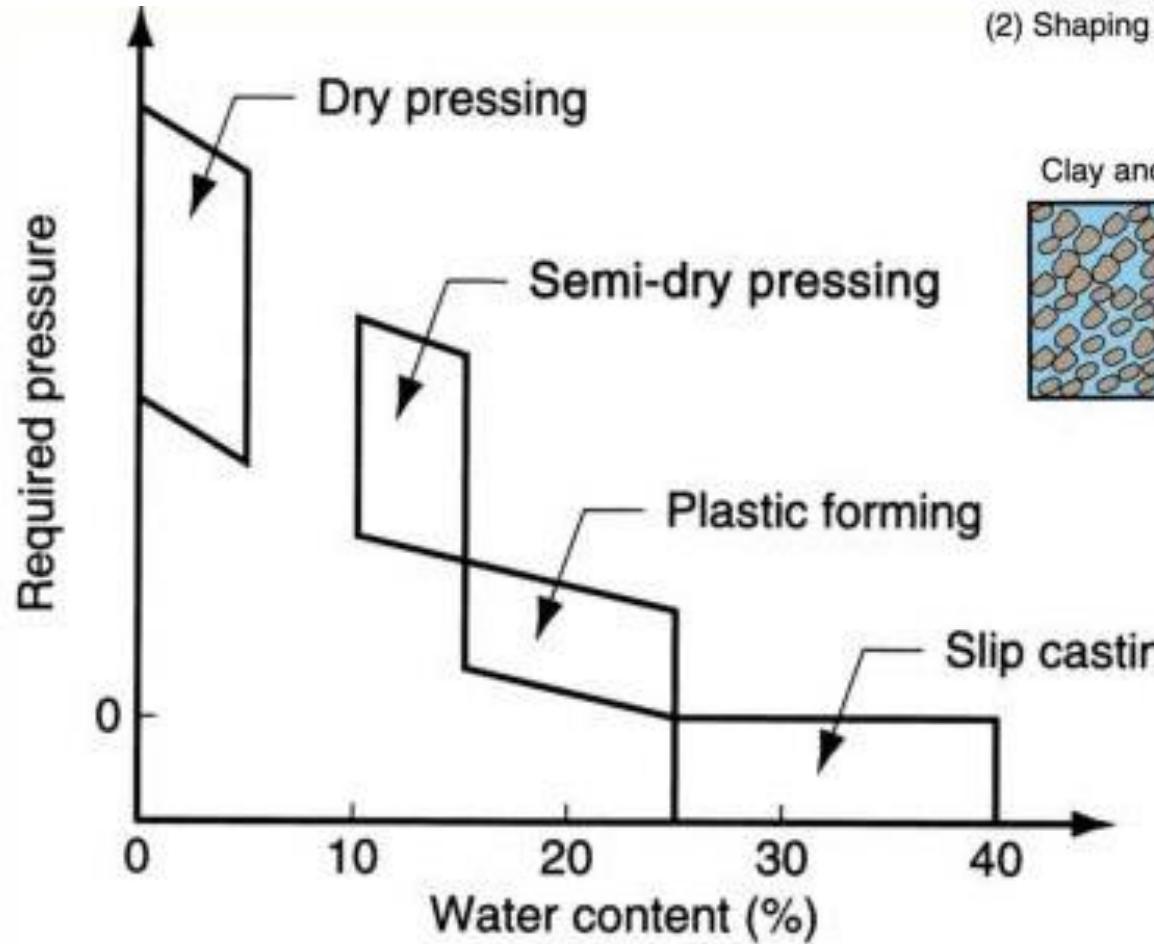


- Starts with a powder
- Powder usually made by ball crushing and then grinding, which is very energy intensive!



Shaping

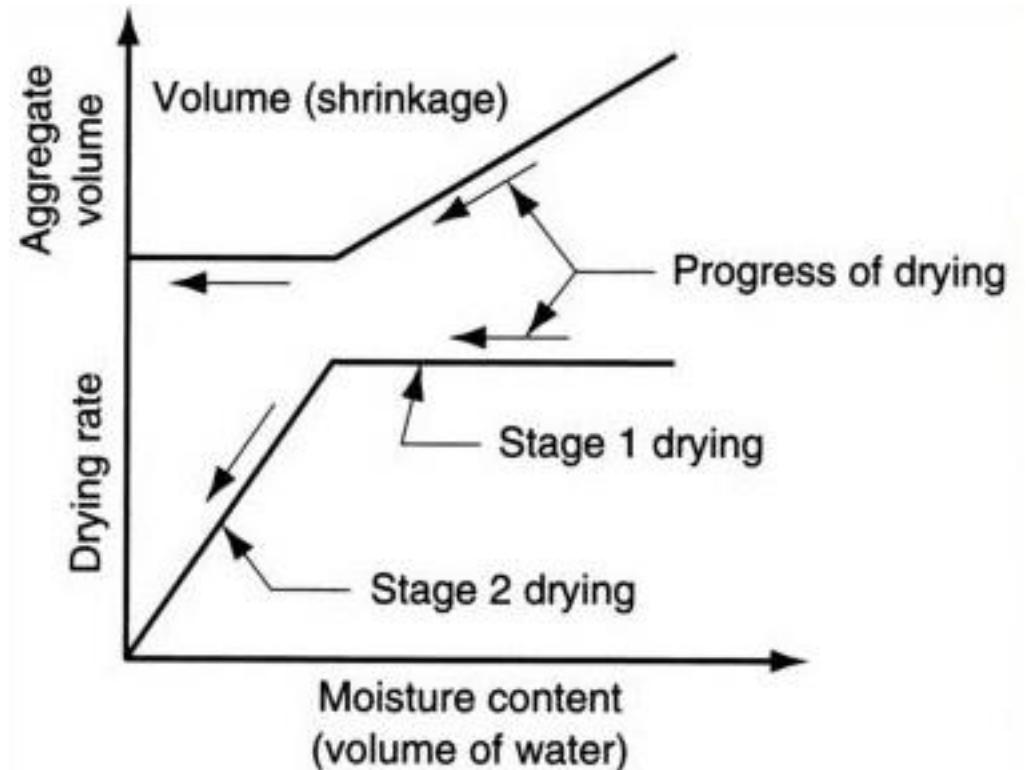
Processing ceramics



- Water is needed for shaping only/plasticity
- High pressures required for semi-dry/dry shaping

Drying – 2 stages

Processing ceramics



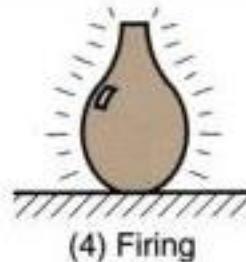
- Stage 1: fast and constant rate to evaporate surface water (prone to shrinkage)
- Stage 2: enough water has been removed so particles are in contact (no shrinkage)



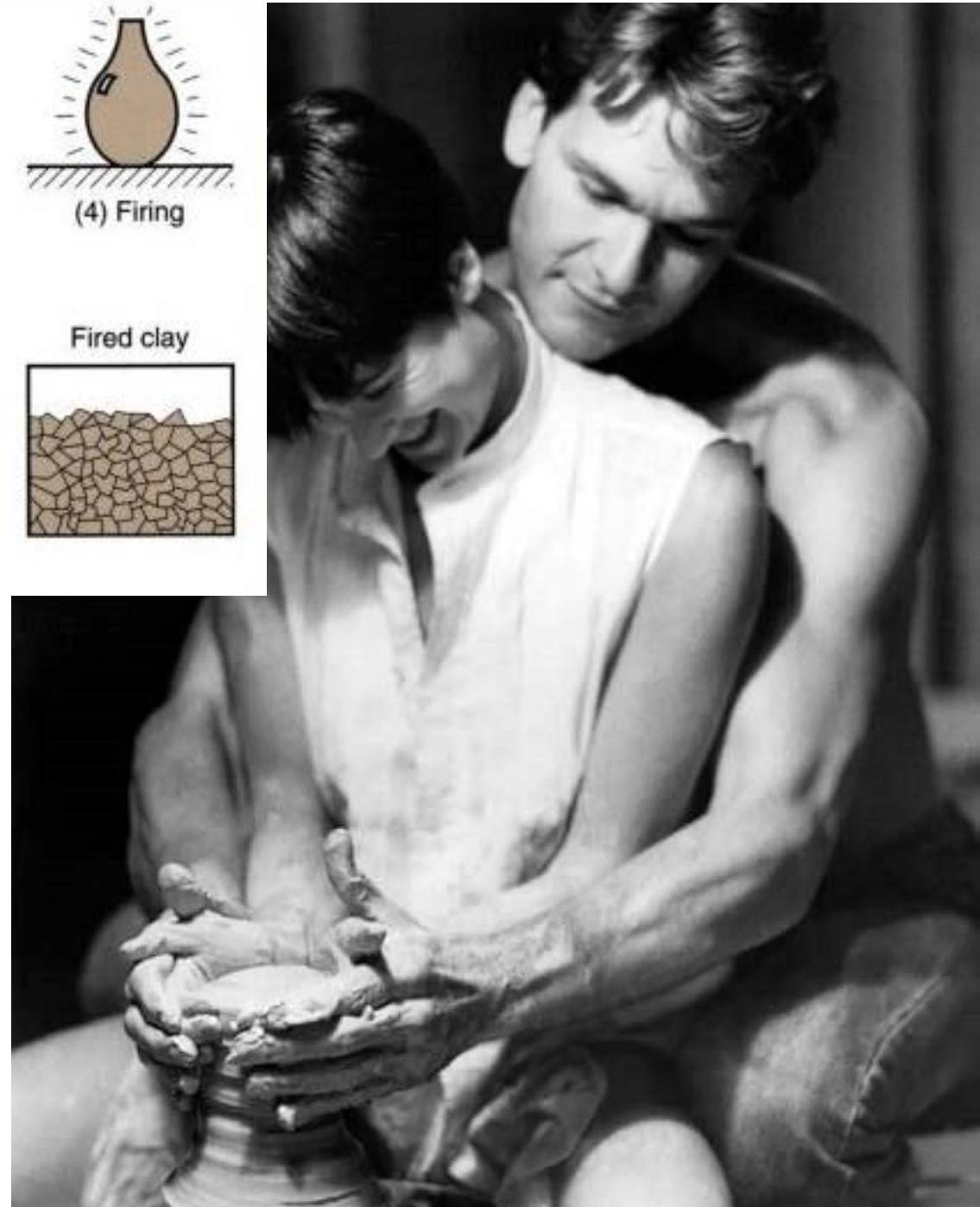
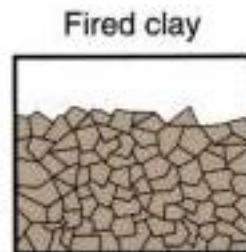
Firing

Processing ceramics

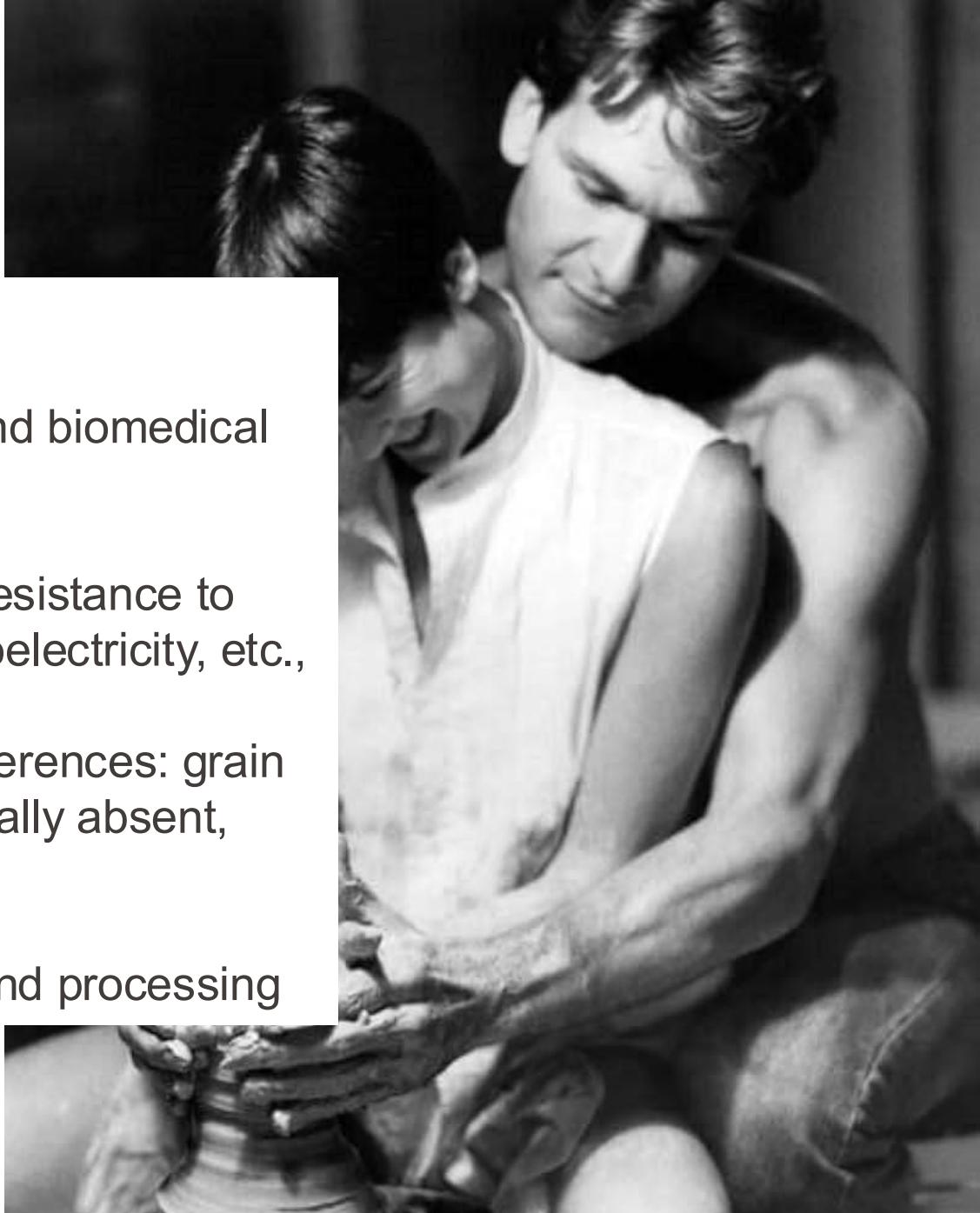
- Heat treatment to *sinter* – form bonds between particles, occurring with densification
- Performed in kiln
- Biggest impact to emissions, related to use of fossil fuels to heat the kiln and process emissions from carbonate raw materials which emit CO₂ when heated
- If glaze is applied, the ceramic is fired again!



(4) Firing

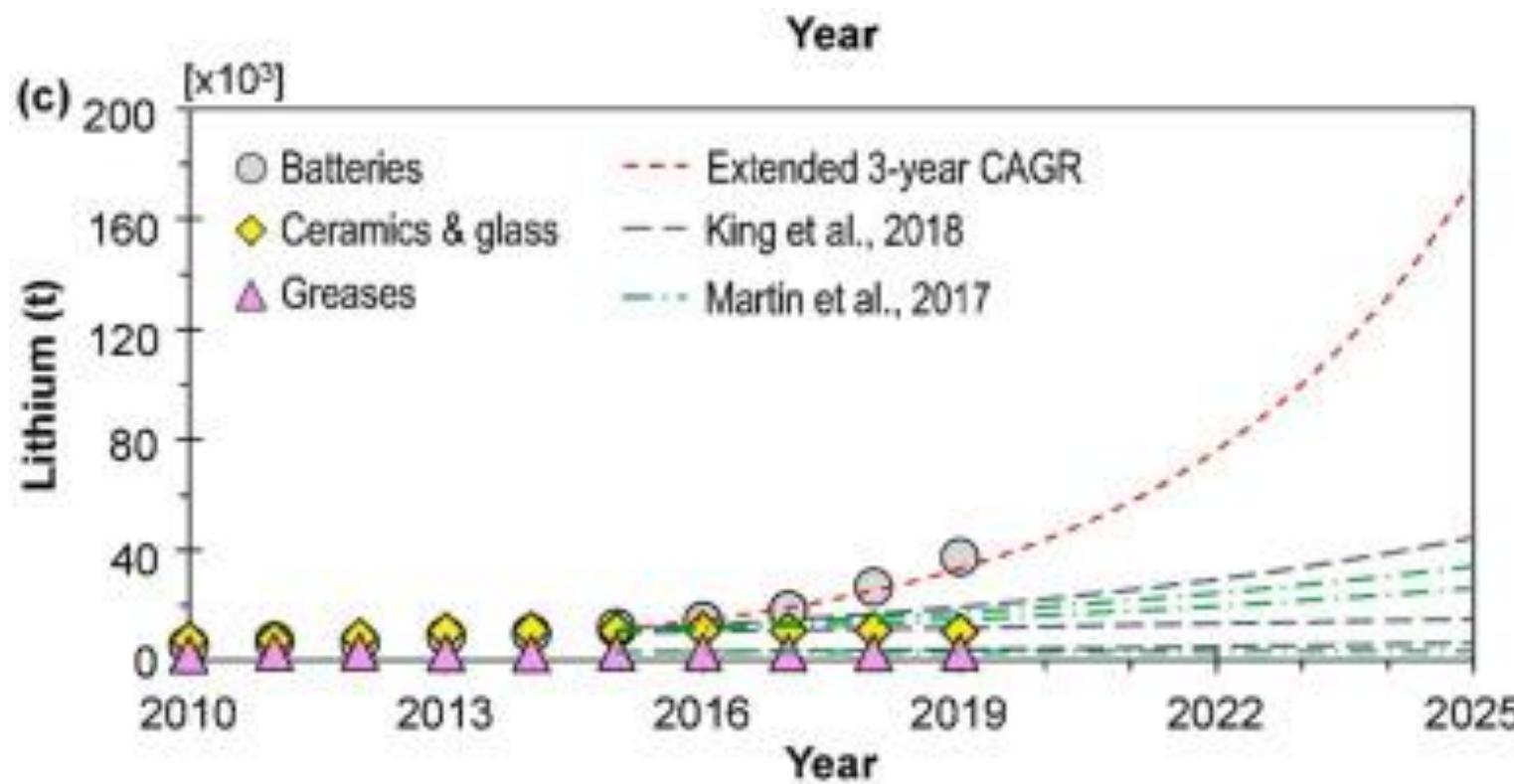


- Made from manmade raw materials
- Developed to meet industrial, technological, and biomedical needs
- Exhibit exceptional properties: high strength, resistance to wear, electrical insulation or conductivity, piezoelectricity, etc.,
- Same general processing steps with some differences: grain size can be controlled chemically, water is usually absent, etc.,
- Emissive hot spots depend on raw materials and processing



Lithium connection?

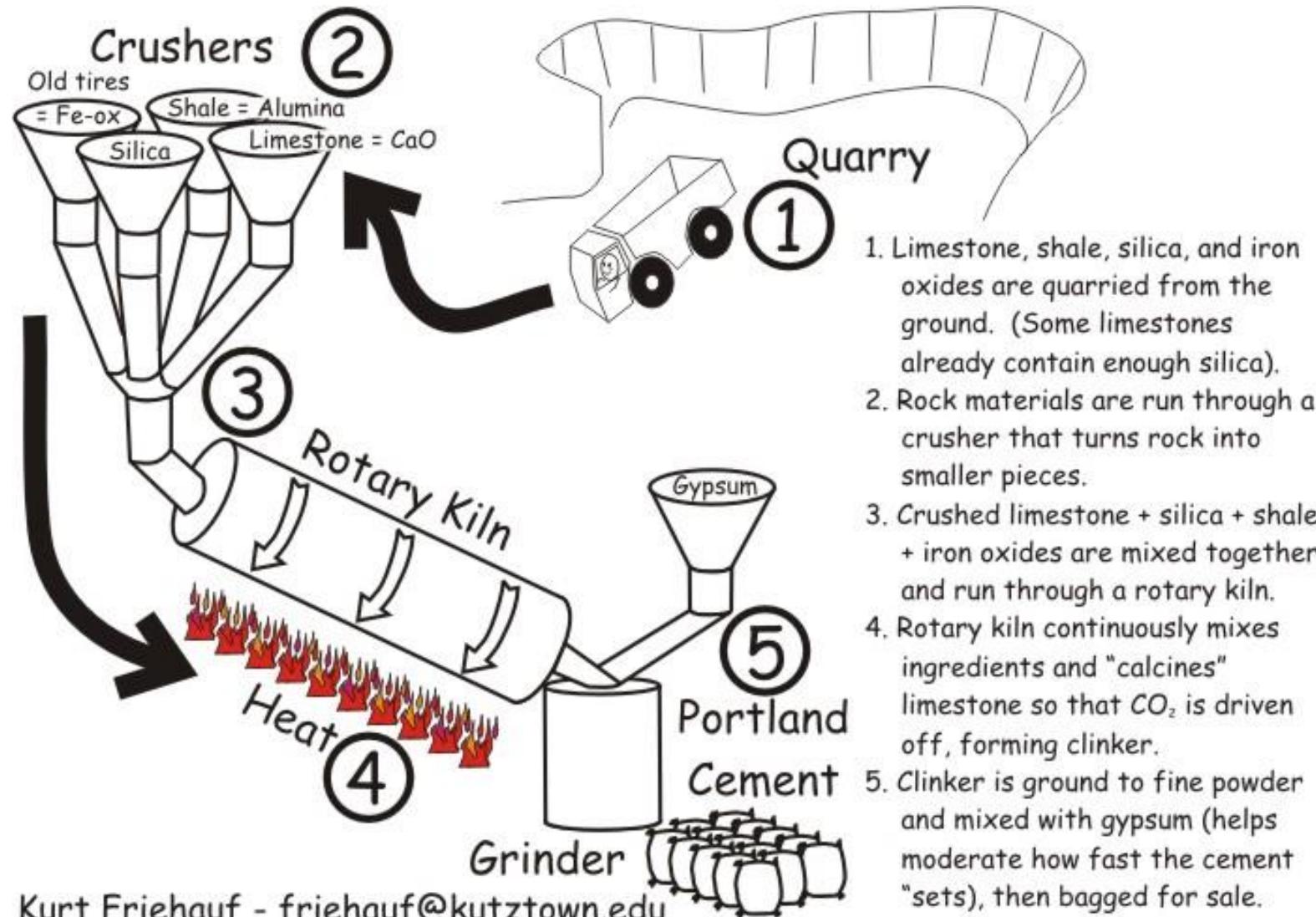
- Lithium is used in advanced ceramics due to its ability to enhance certain properties: low thermal expansion, high strength and toughness, melting point reduction, improved optical properties, enhanced electrical properties, light weight



- Cathode material in batteries
- Now batteries, but before 2015, ceramics and glass was the main industry using lithium

[Decarbonizing ceramics_2022](#)

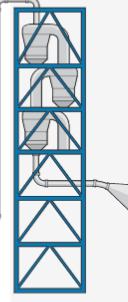
How cement is made



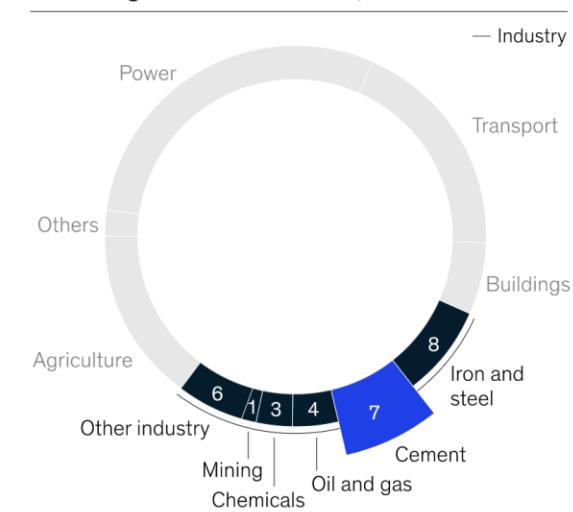
[Cement manufacture](#)

<https://www.mckinsey.com/industries/chemicals/our-insights/laying-the-foundation-for-zero-carbon-cement>

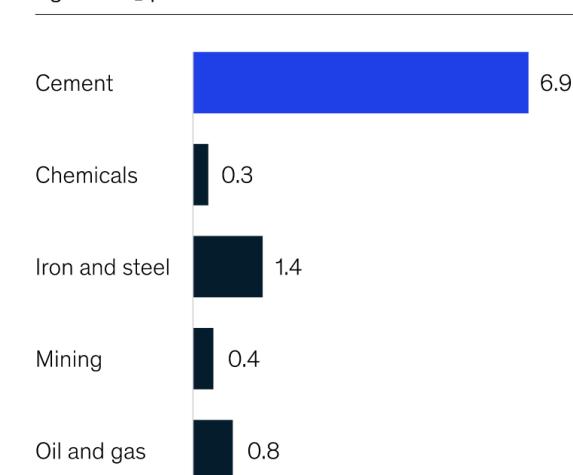
Raw materials, energy, and resources

	Clinker and cement manufacturing									
	Quarry		Crusher	Transport ¹		Raw mill		Kiln and preheater/ precalibrator ²		
								Cooler ³		
								Cement mill		
								Logistics ⁴		
								Total		
Energy, mega-joule/ton	40	5	40	100	3,150	160	285	115	3,895	
CO₂, kilogram/ton	3	1	7	17	479 Calcination process	319 Fossil fuels	28	49	22	925

Share of global CO₂ emissions, % in 2017

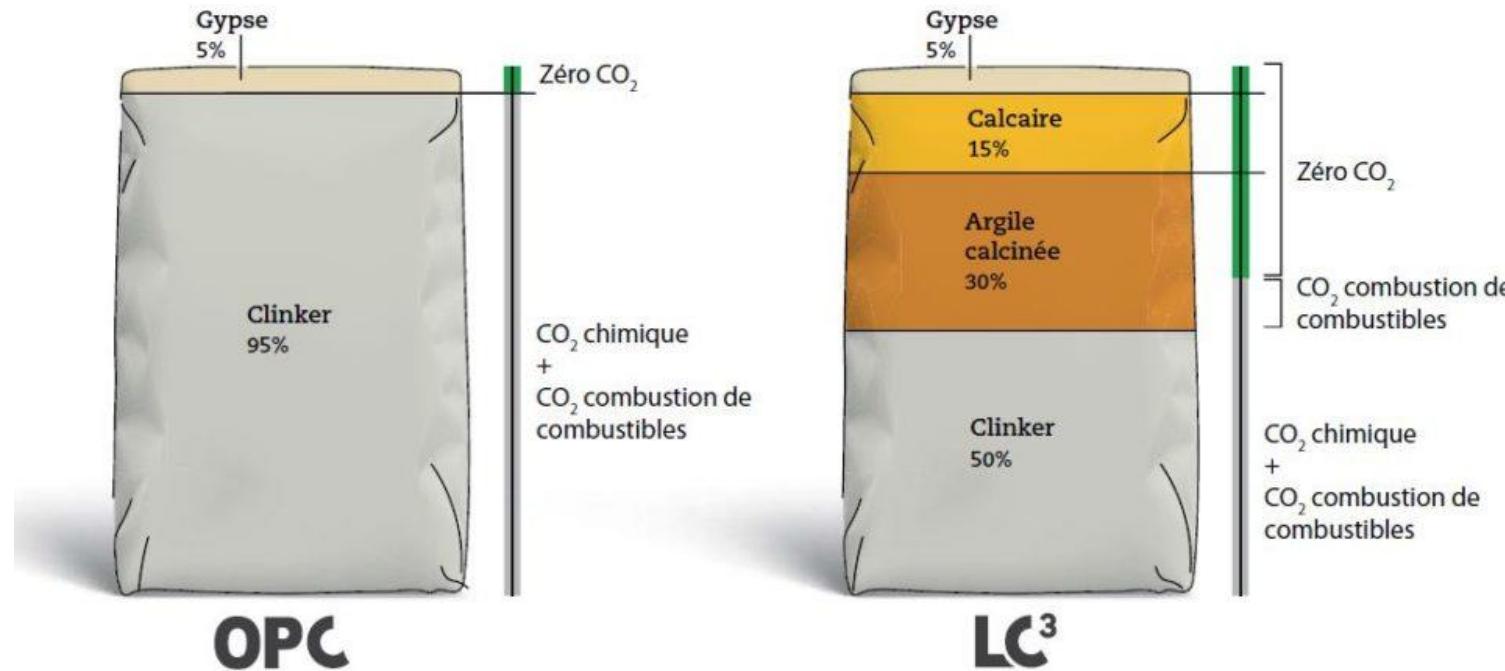


kg of CO₂ per \$



- The cement industry alone is responsible for about a quarter of all industry CO₂ emissions, and it also generates the most CO₂ emissions per dollar of revenue
- About two-thirds of those total emissions result from calcination, the chemical reaction that occurs when raw materials such as limestone are exposed to high temperatures.

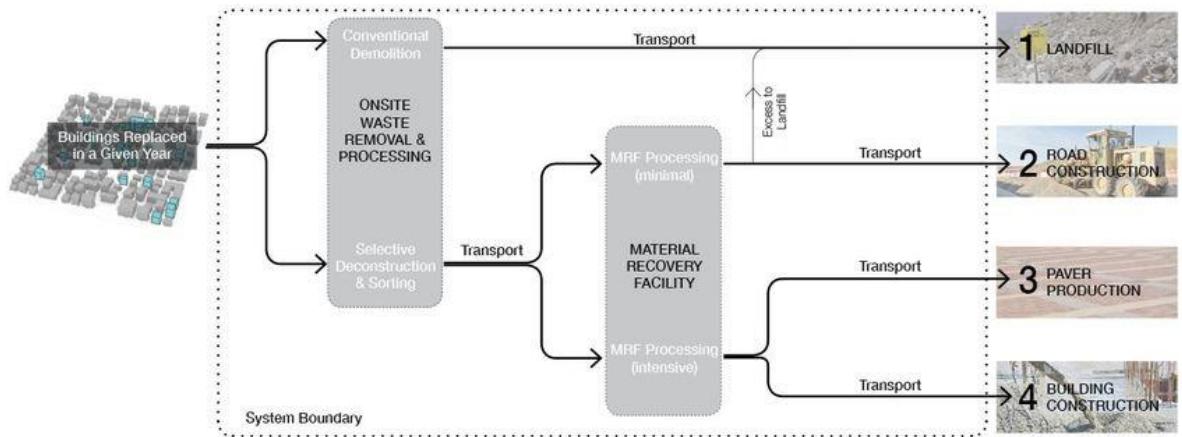
Decarbonizing cement: EPFL connection (LMC)



- Replaces half of clinker with calcined clay and ground limestone, neither of which releases CO₂ when heated the way limestone does
- Clay is heated to lower temperatures, reducing fuel needed and emissions
- Lower temperatures mean that cleaner energy, like electricity can be used

[LC3 - more sustainable cement](#)

End of life – Ceramics and Glasses



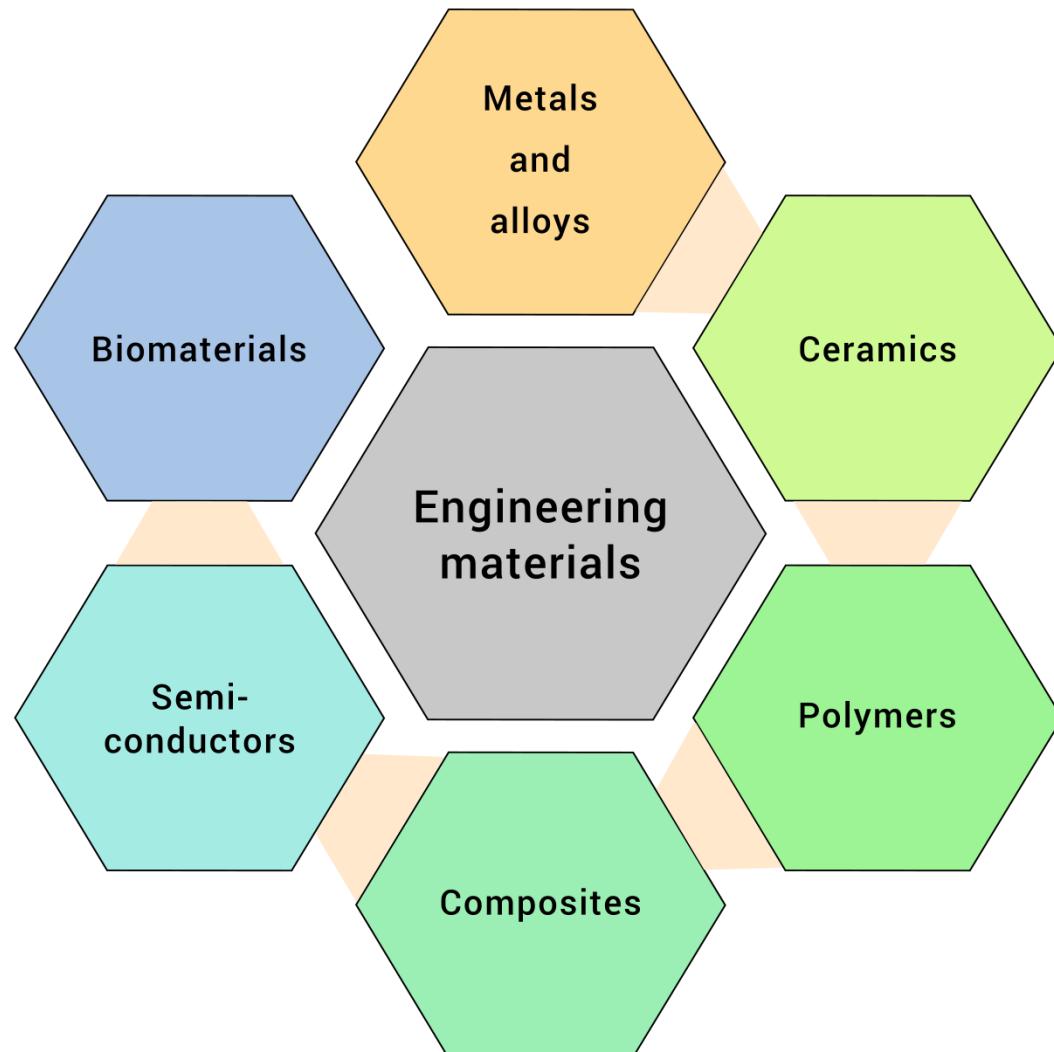
Concrete

- About 60% of what is crushed can be used for downcycling processes (recovery of materials with a more limited range of uses than the original material).
- These fragments can be used as base materials for structures such as roads.

Recycling concrete



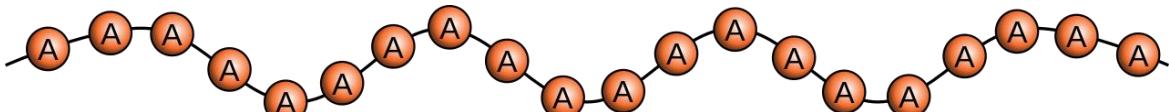
Engineering materials categories (variable)



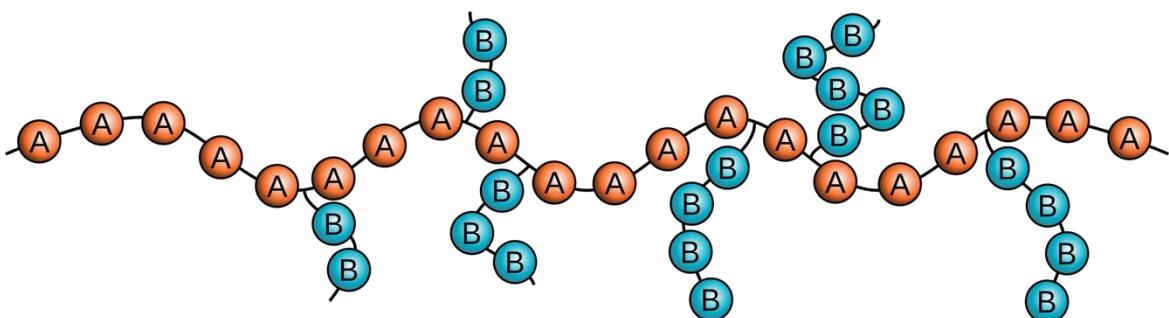
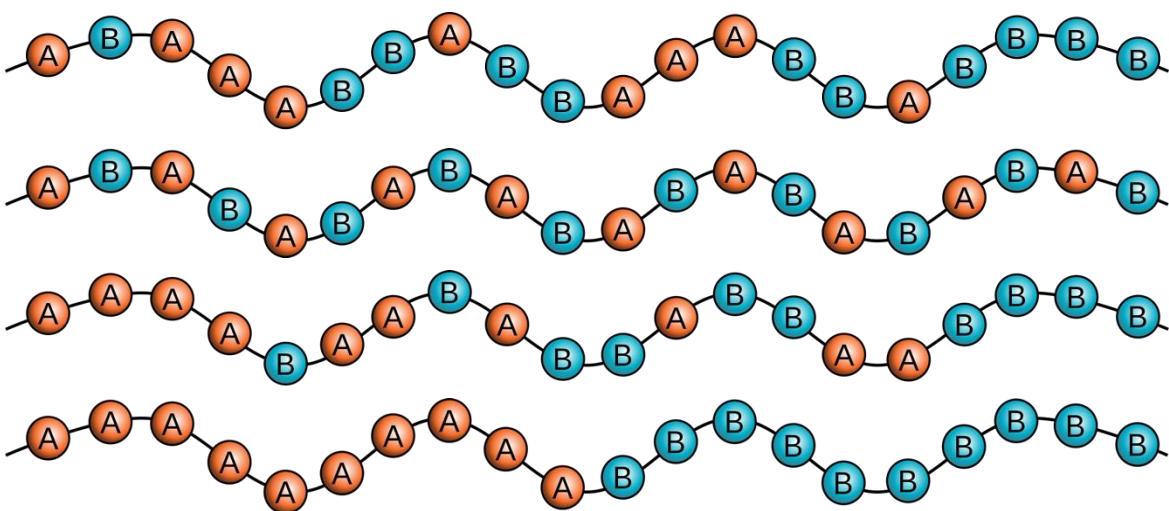
Focus: Plastics

Generic definition of a polymer

homopolymer

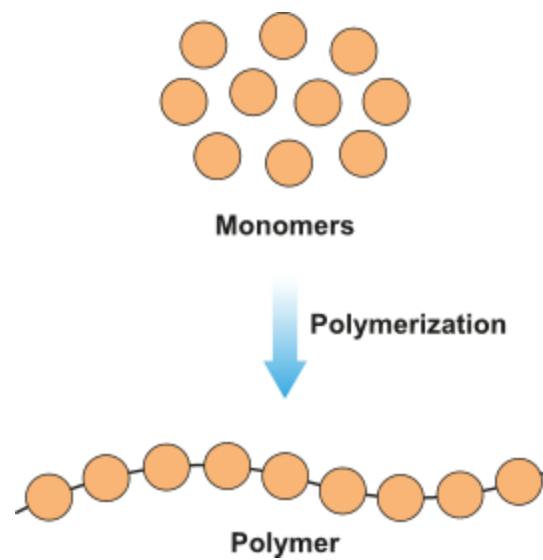


copolymer



- Covalently linked repeat units, building up into a long molecule, aka a macromolecule or a polymer
- Can be nature-derived, semi-synthetic or synthetic
- Often defined by chain length
- Versatile chemistry and chain length translates to versatile properties (crystallinity, viscosity, thermal, mechanical, etc.,)

Polymer definition



Hermann Staudinger
"Father of Macromolecular Chemistry"
1953 Chemistry Nobel Prize



1920
"In a paper entitled "Über Polymerisation," Staudinger presented several reactions that form high molecular weight molecules by linking together a large number of small molecules. During this reaction, which he called "polymerization," individual repeating units are joined together by covalent bonds."

"...Heinrich Wieland, 1927 Nobel laureate in chemistry, wrote to Staudinger, "Dear colleague, drop the idea of large molecules; organic molecules with a **molecular weight higher than 5000 do not exist**. Purify your products, such as rubber, then they will crystallize and prove to be low molecular compounds!"

"My colleagues were very skeptical about this change, and those who knew my publications in the field of low molecular chemistry asked me why I was neglecting this interesting field and instead was working on a very unpleasant field and poorly defined compounds, like rubber and synthetic polymers. At that time the chemistry of these compounds often was designated, in view of their properties, as Schmierenchemie ('grease chemistry')."

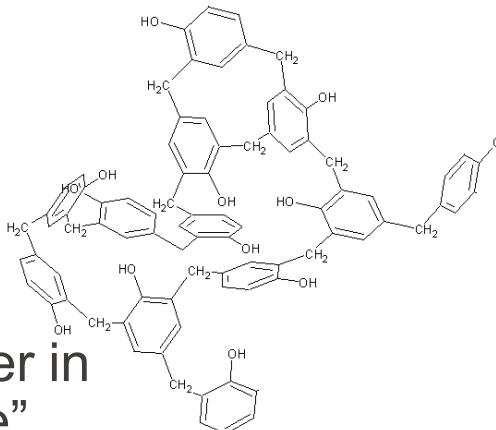
Celluloid

- Late 19th century
- John Wesley Hyatt in response to \$10,000 reward for ivory substitute
- Celluloid = partially nitrated cellulose (“guncotton”) + camphor
- 1st semi-synthetic plastic
- Thermoplastic
- (Launching the age of man-made plastics)



Bakelite

- World's first fully synthetic polymer in 1907, heralding the "Polymer Age"



- "Insoluble in all solvents, does not soften. I call it Bakelite (sic)"

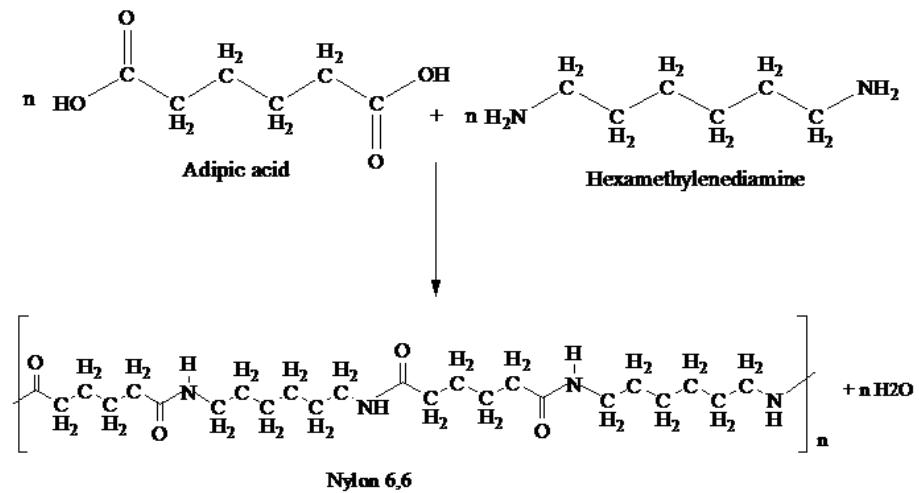
- Moldable
- Thermoset



"Von Baeyer had reported that when he mixed phenol, a common disinfectant, with formaldehyde, it formed a hard, insoluble material that ruined his laboratory equipment, because once formed, it could not be removed. Kleeburg reported a similar experience, describing the substance he produced as a hard amorphous mass, infusible and insoluble and thus of little use."

Nylon

- Wallace Carothers at Dupont (US)
- Nylon developed in 1930s
- Condensation polymerization
- Thermoplastic
- High strength, rigidity, good heat and chemical resistance
- Highly used, e.g., automotive, textile fibers, etc.,



Examples of common synthetic polymers

ADDITION POLYMERS

#1

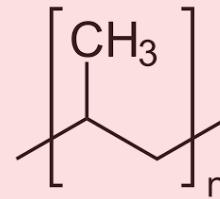
PE



packaging

#2

PP



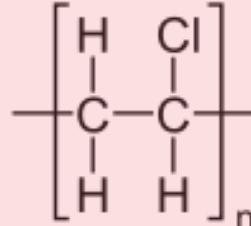
packaging

Teflon



#3

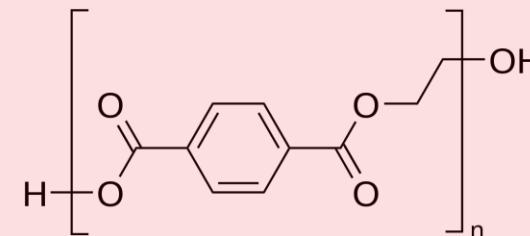
PVC



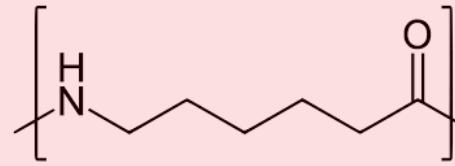
pipes

CONDENSATION POLYMERS

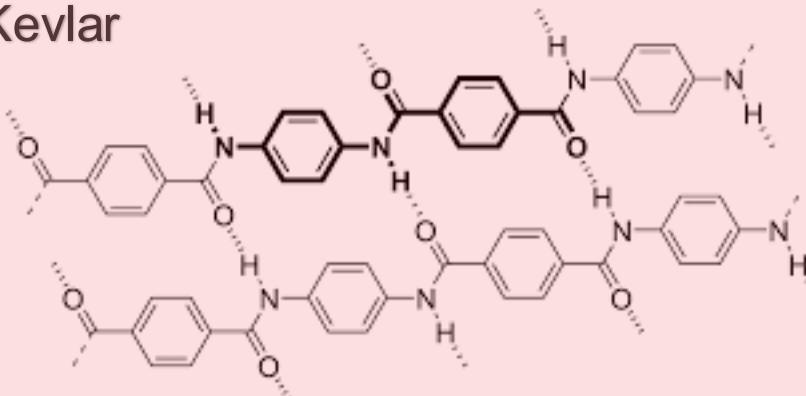
PET

Polyester
Carbonated drinks

Nylon 6,6

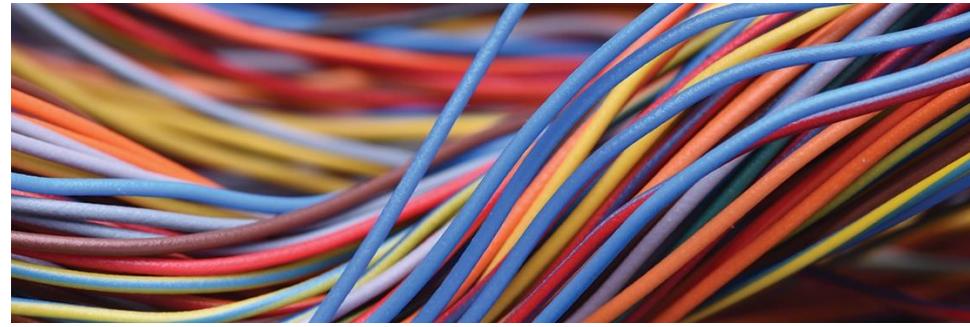
Polyamide
Textiles

Kevlar

aromatic polyamide
Engineering materials

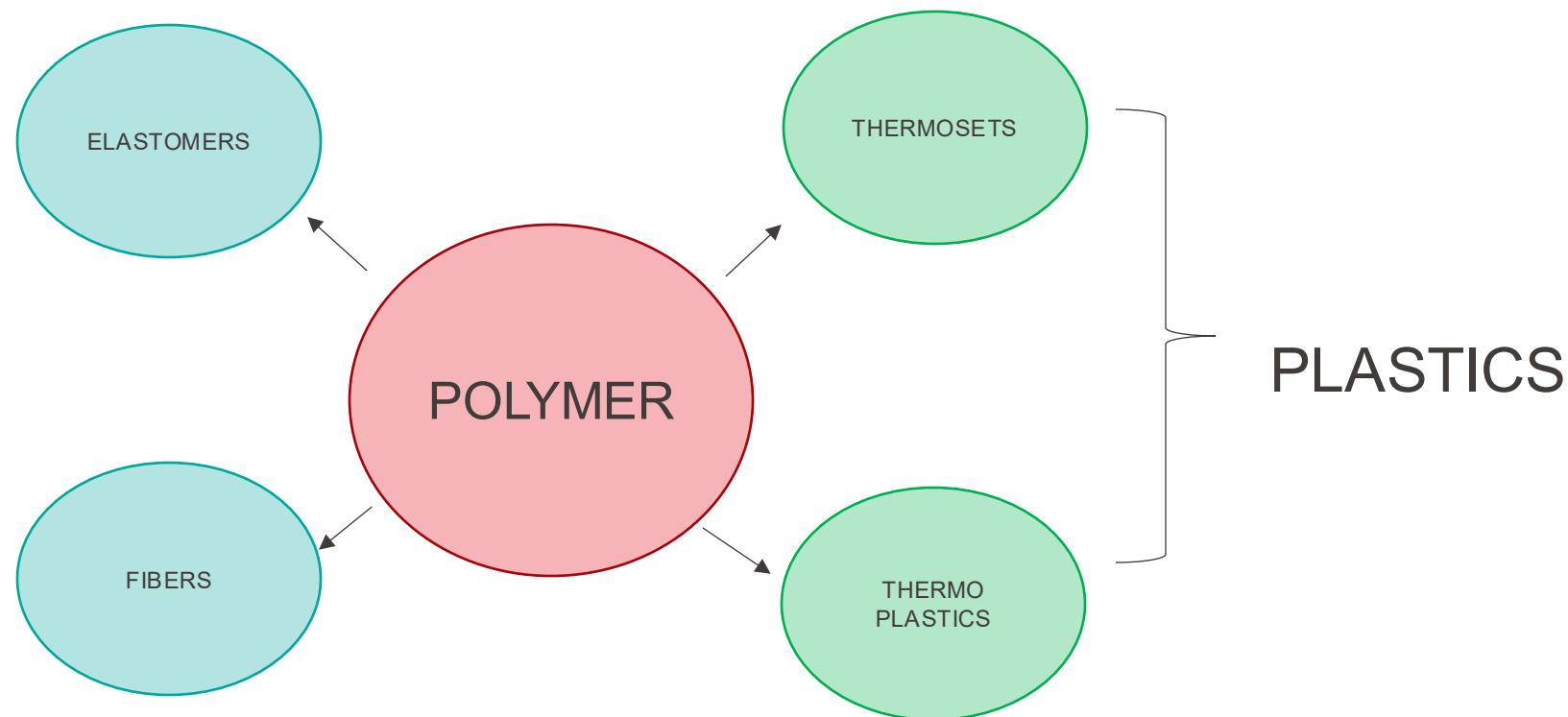
General characteristics of plastics

- Cheap & versatile
- Excellent strength to weight ratio
- Highly processible
- Non-reactive – inert, not easily corroded, BUT can be degraded and broken down into microplastics by UV-light, abrasion
- Good insulators
- Tailorable appearance and properties
- Life cycle & end of life complex for synthetic polymers



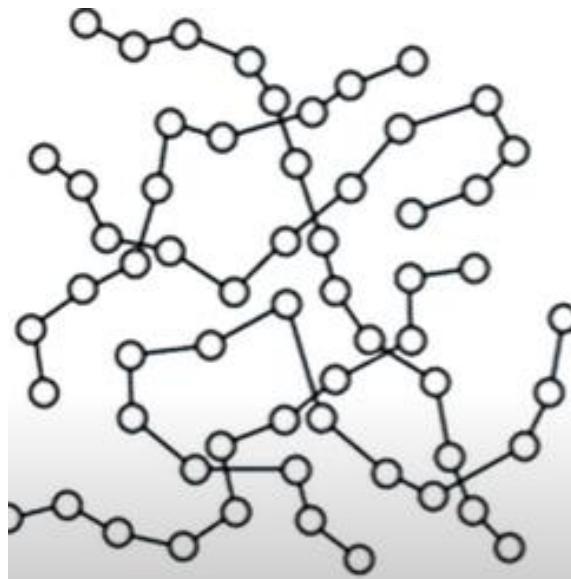
Polymer vs. plastic

- *Polymer* is from Greek for “of many parts”
- *Plastic* is from “*Plastikos*”, Greek for *pliable*



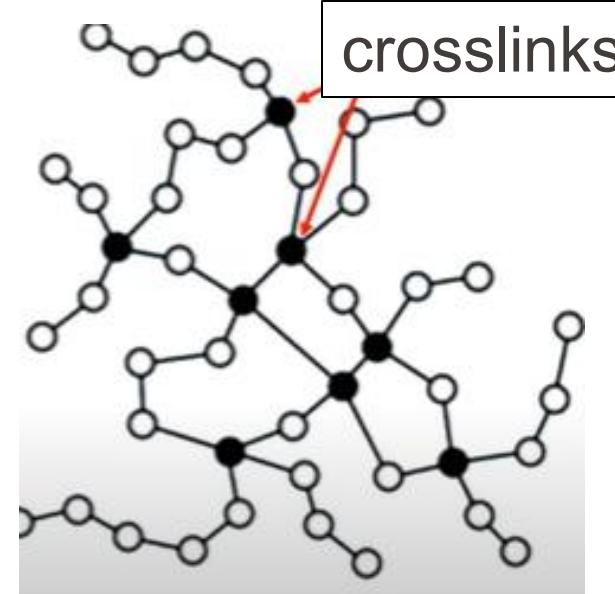
Thermoplastics vs. Thermosets

THERMOPLASTICS



- Weak intermolecular bonds
- Softens when heated
- Hardens when cooled
- Reversible

THERMOSETS



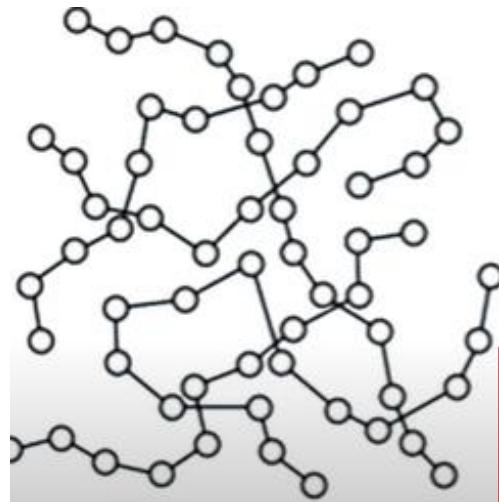
- Strong covalent bonds (“crosslinks”)
- Rigid, prevents melting when heated
- Cannot be reshaped or remolded

[thermoplastics vs thermosets](#)

<https://www.youtube.com/watch?v=4fTtrKPySm0>

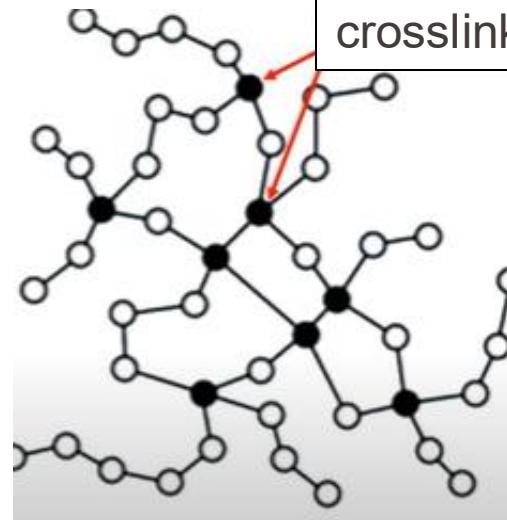
Thermoplastics vs. thermosets

THERMOPLASTICS



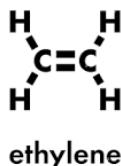
- Remoldable
- Recyclable
- High strength
- Shrink resistance
- Flexibility

THERMOSETS

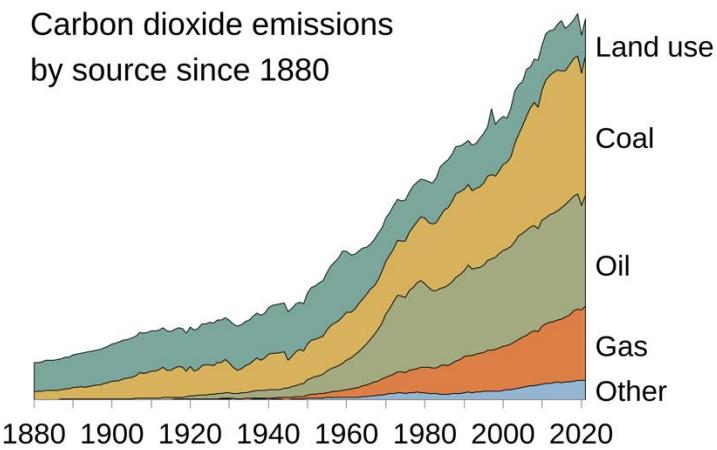
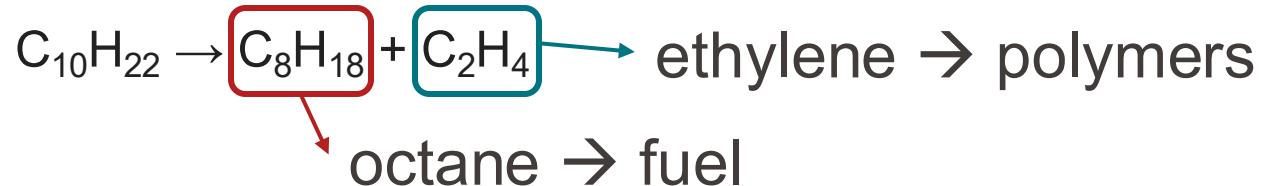


- Heat resistant
- Corrosion resistant
- High strength
- Mechanical creep resistance
- High specific strength (strength to weight ratio)

Plastic resources

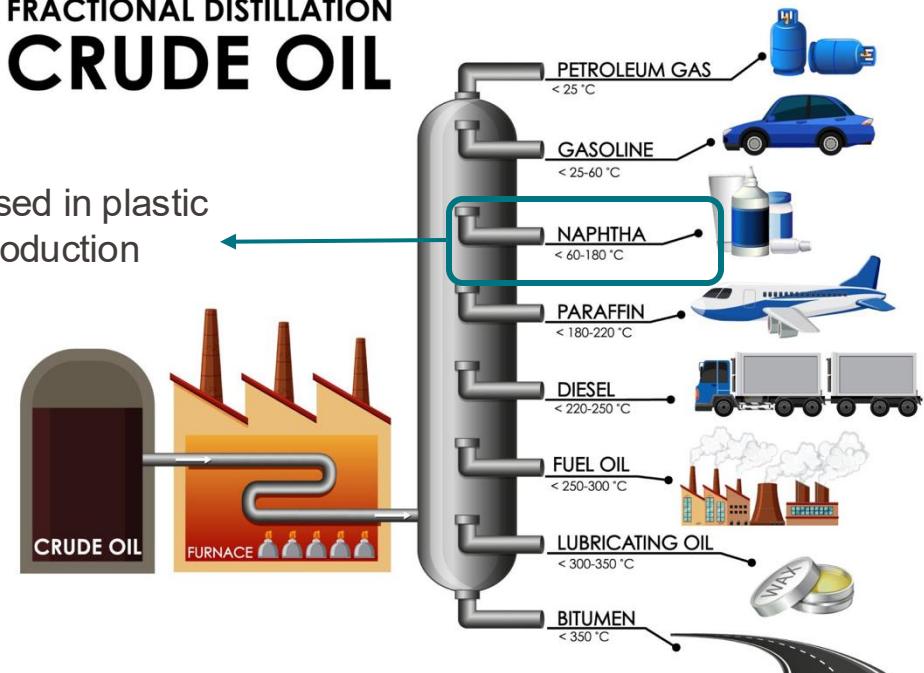


- Most plastic in use today comes from hydrocarbons derived from crude oil, natural gas, and coal – fossil fuels
- Crude oil is a combination of hundreds of different hydrocarbons
- Fractional distillation is used to separate them into their various components
- Higher molecular weight hydrocarbons are then selected and further processed (cracking) to yield lower molecular weight alkenes and alkanes (e.g., decane cracking)



FRACTIONAL DISTILLATION CRUDE OIL

Used in plastic production

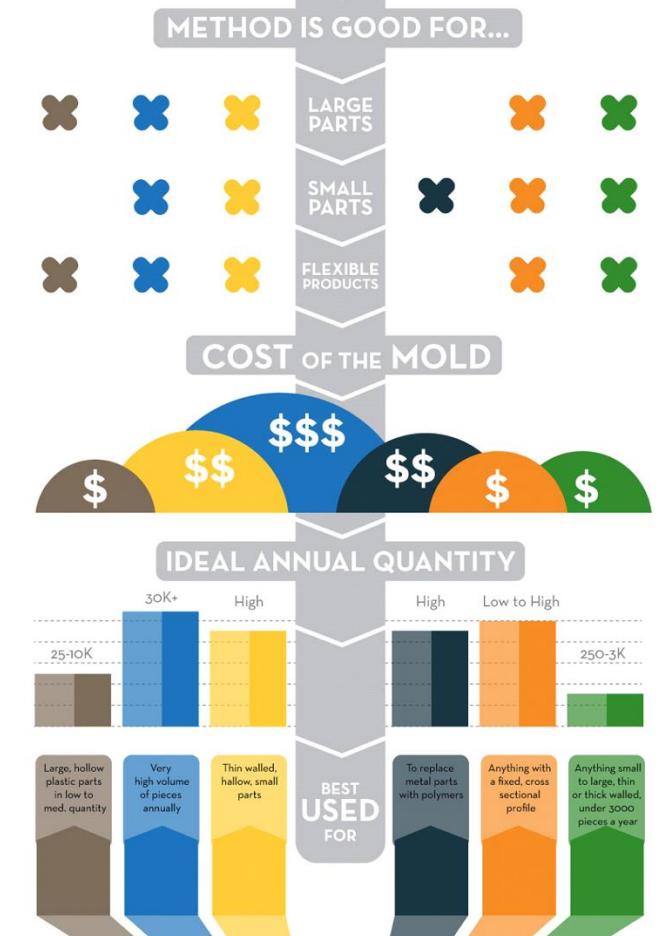
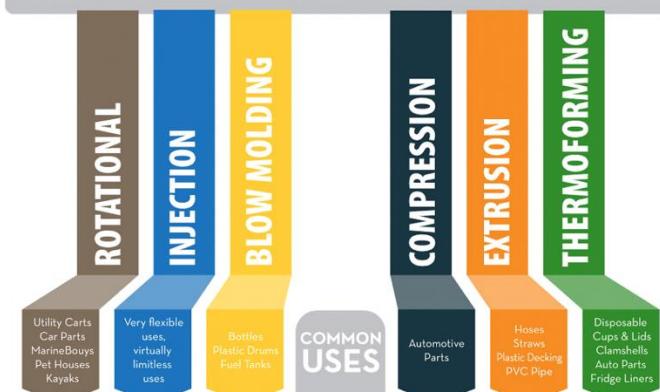


Plastic processing

- Rotational molding
- Injection molding
- Blow molding
- Compression molding
- Extrusion molding
- Thermoforming



PLASTIC MOLDING COMPARISON







Plasticene

A NEW GEOLOGICAL AGE...(last 60-70 years!)

n. & adj. (2011) an era in Earth's history, within the Anthropocene, commencing in the 1950s, marked stratigraphically in the depositional record by a new and increasing layer of plastic ([Stager, 2011](#), attributed to Matt Dowling). The history and etymology of plasticene in this sense is not related to the word *plasticene*, a common early spelling (for example, [Cooper, 1901](#)) of the popular molding clay, plasticine.

Useful and ubiquitous



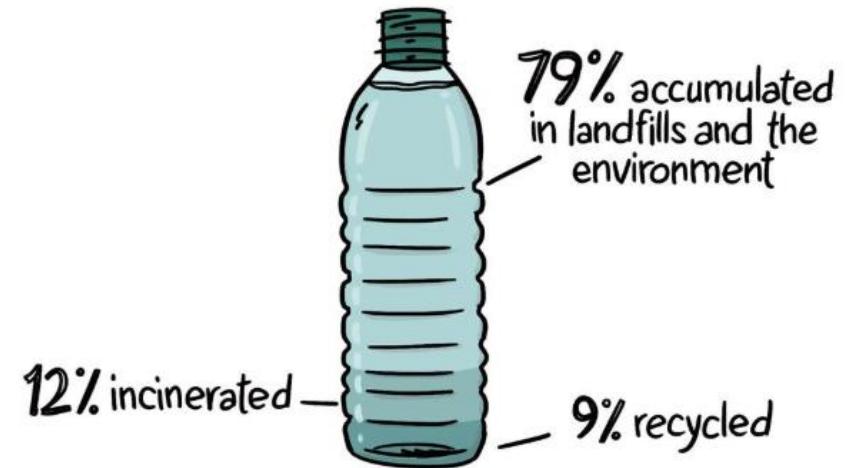
Plastics everywhere

Plastic promises: How'd we do?

Data from 1950-2015:

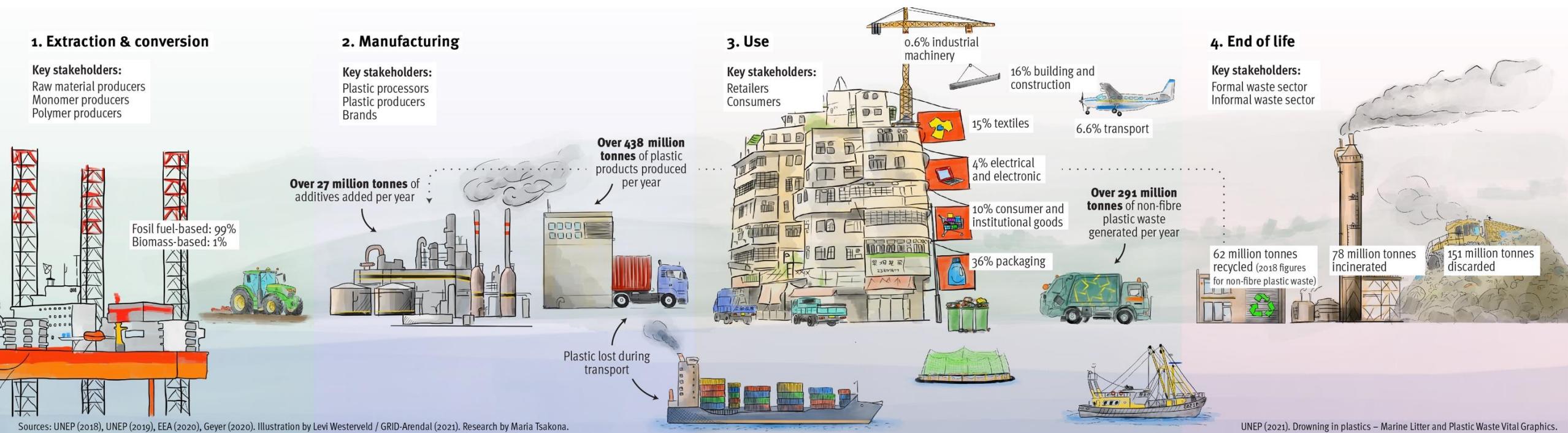
- 8300 Mt of plastics
- 6300 Mt of plastic waste
- 12000 Mt in landfill/nature projected by 2050
- Conventional plastics do not biodegrade
- Increased demand for plastics
- Permanent elimination only by incineration

Only about 30% of plastic ever produced is still in use — the rest has been disposed of in one of three ways:



Geyer R. et al., Science Advances 2017, 3(7), 1-5.

The life cycle of plastic



- Of the 438 million tons of plastic produced every year, only about 14% is recycled
- A total of 99% of feedstock for plastic production is fossil fuel-based, accounting for around 8-9% of global oil and gas consumption

Persistent and ubiquitous

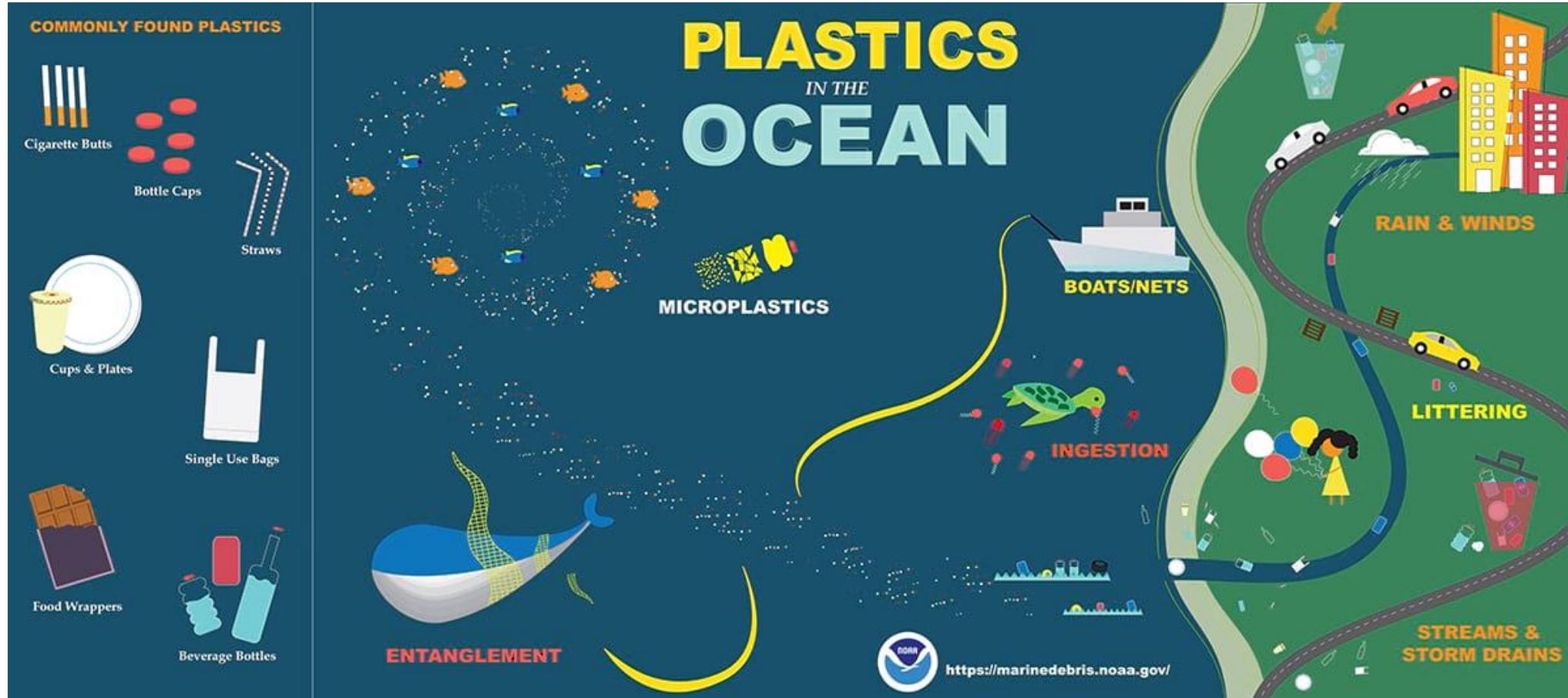


The New York Times

STEM WRITING CONTEST WINNER

An Unexpected Dinner Guest: Marine Plastic Pollution Hides a Neurological Toxin in Our Food

We are honoring each of the top eight winners of our Student STEM Writing Contest by publishing their essays. This one is by Vivian Li.



- 80% of marine pollution is plastic, 8-10 Mt of plastic/year
- Essentially all manmade plastics are still in existence (except what has been incinerated, approx. 9%)
- Plastics either break down into microplastics or form garbage patches



The Lifecycle of Plastics

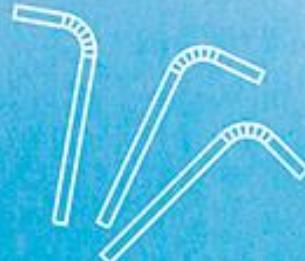
- 500-1000 years degradation time
- Even then, it becomes microplastics.
- **50-75 trillion pieces of plastic and microplastics in the ocean.**



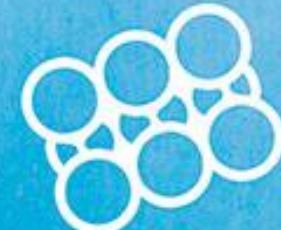
Plastic bag

20 years

Coffee cup

30 years

Plastic straw

200 years

6-pack plastic rings

400 years

Plastic water bottle

450 years

Coffee pod

500 years

Plastic cup

450 years

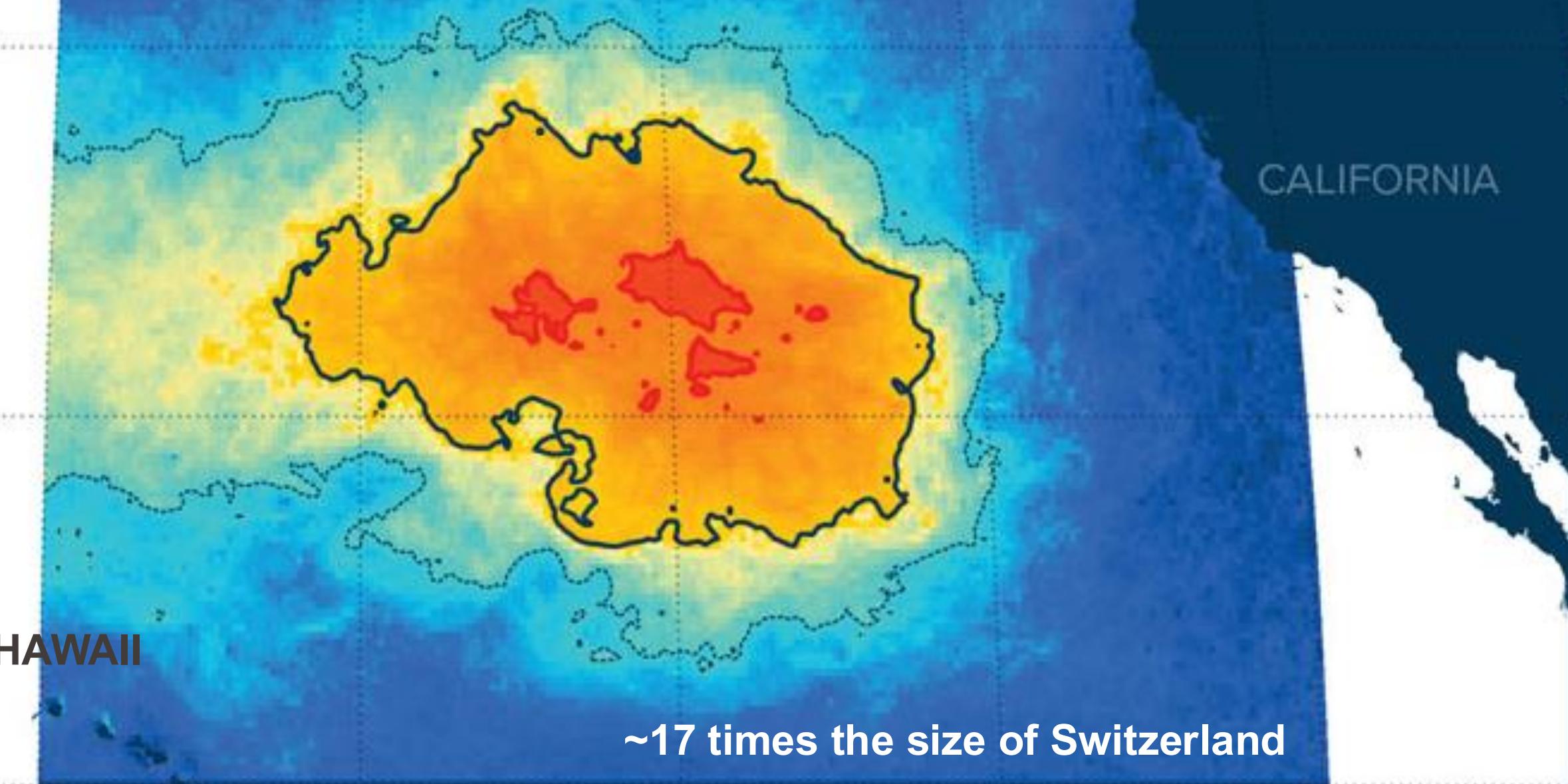
Disposable diaper

500 years

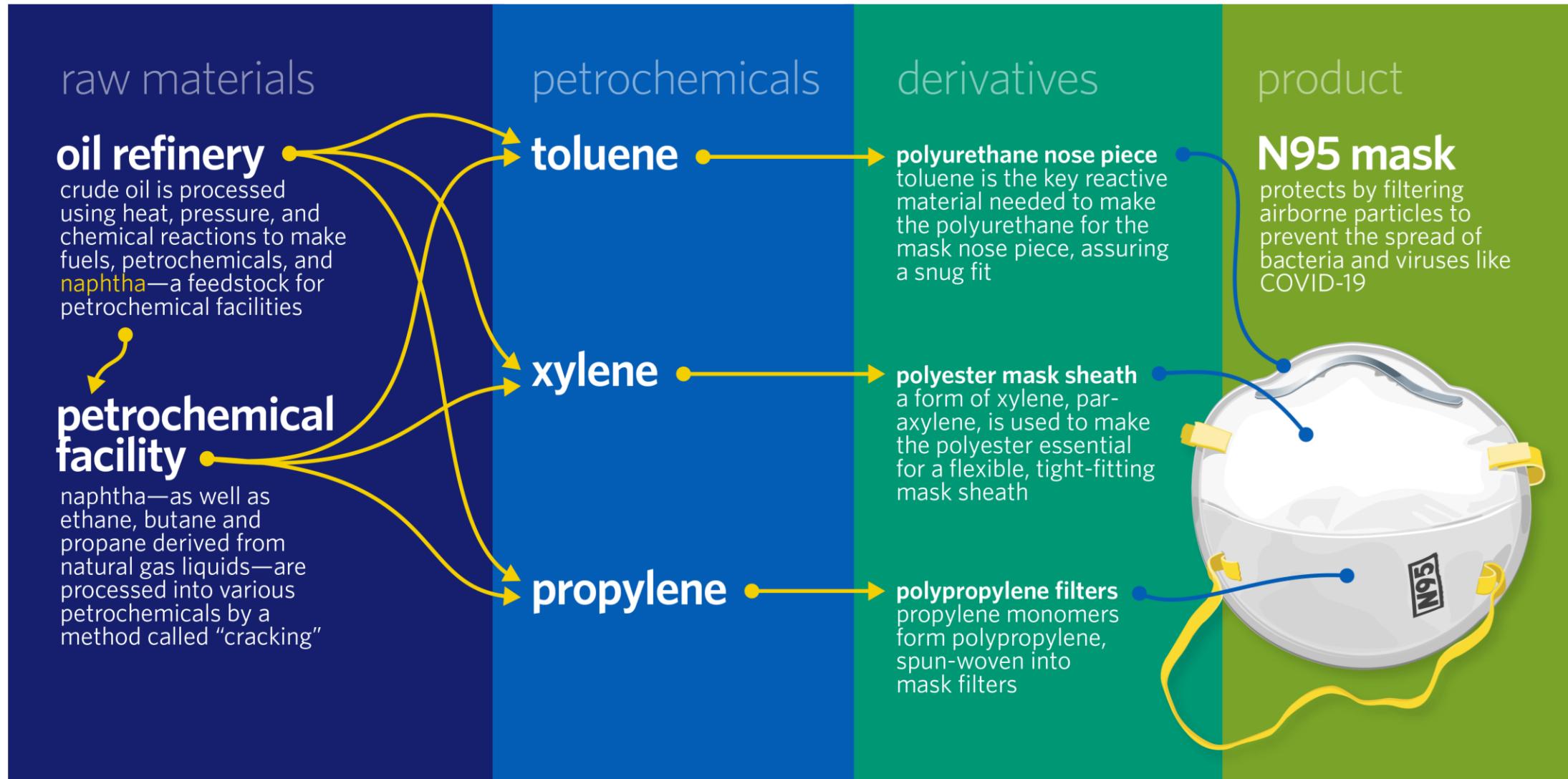
Plastic toothbrush

500 years

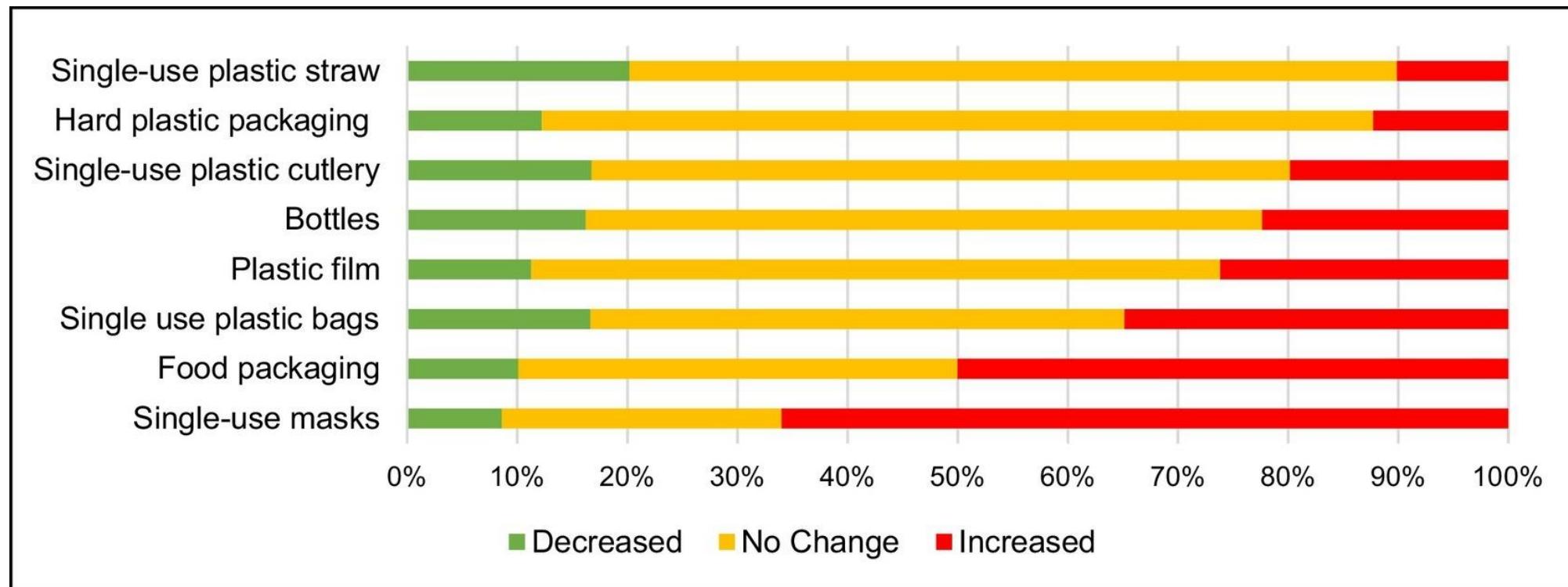
Great Pacific Garbage Patch



Pandemic plastics



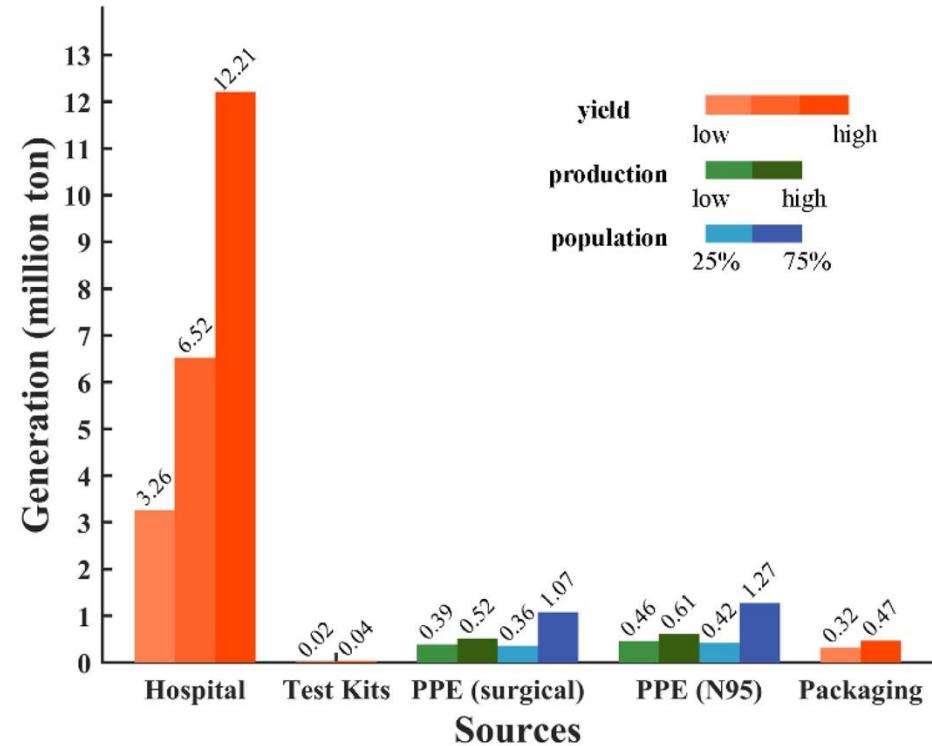
Pandemic plastics



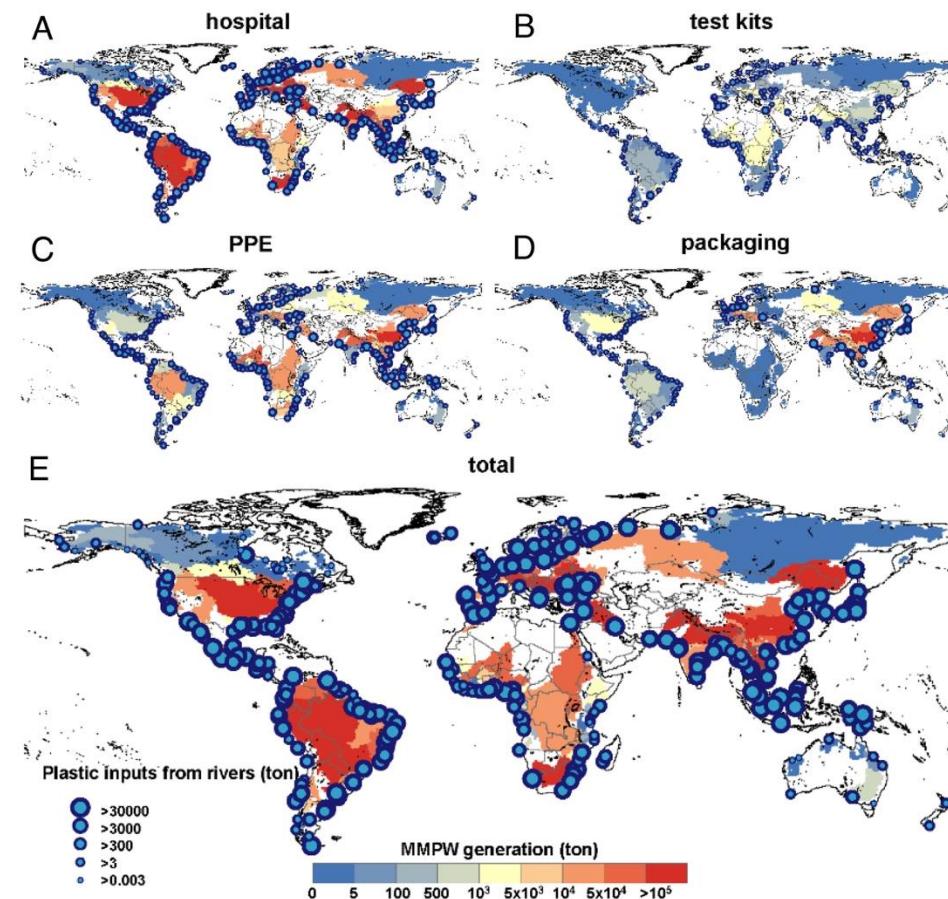
[SUPs in the Pandemic](#)

*Based on 202 respondents from 41 countries

Pandemic plastics in marine environment



- Excess mismanaged plastic waste (MMPW) 4.4-15.1 million tons (8.4 million tons best estimate)
- 87% from hospitals, 8% PPE by individuals, packaging 4.7%, test kits 0.3%



- Top river discharges from – Shatt al Arab, Indus, Yangtze River, Ganges, Danube, Amur
- 73% of discharge is from Asia, 11% from Europe

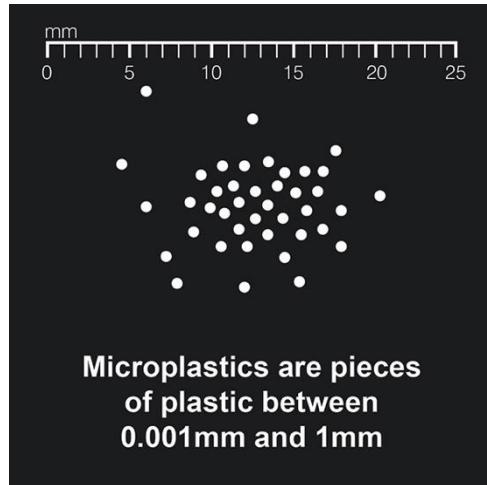
Persistent and ubiquitous



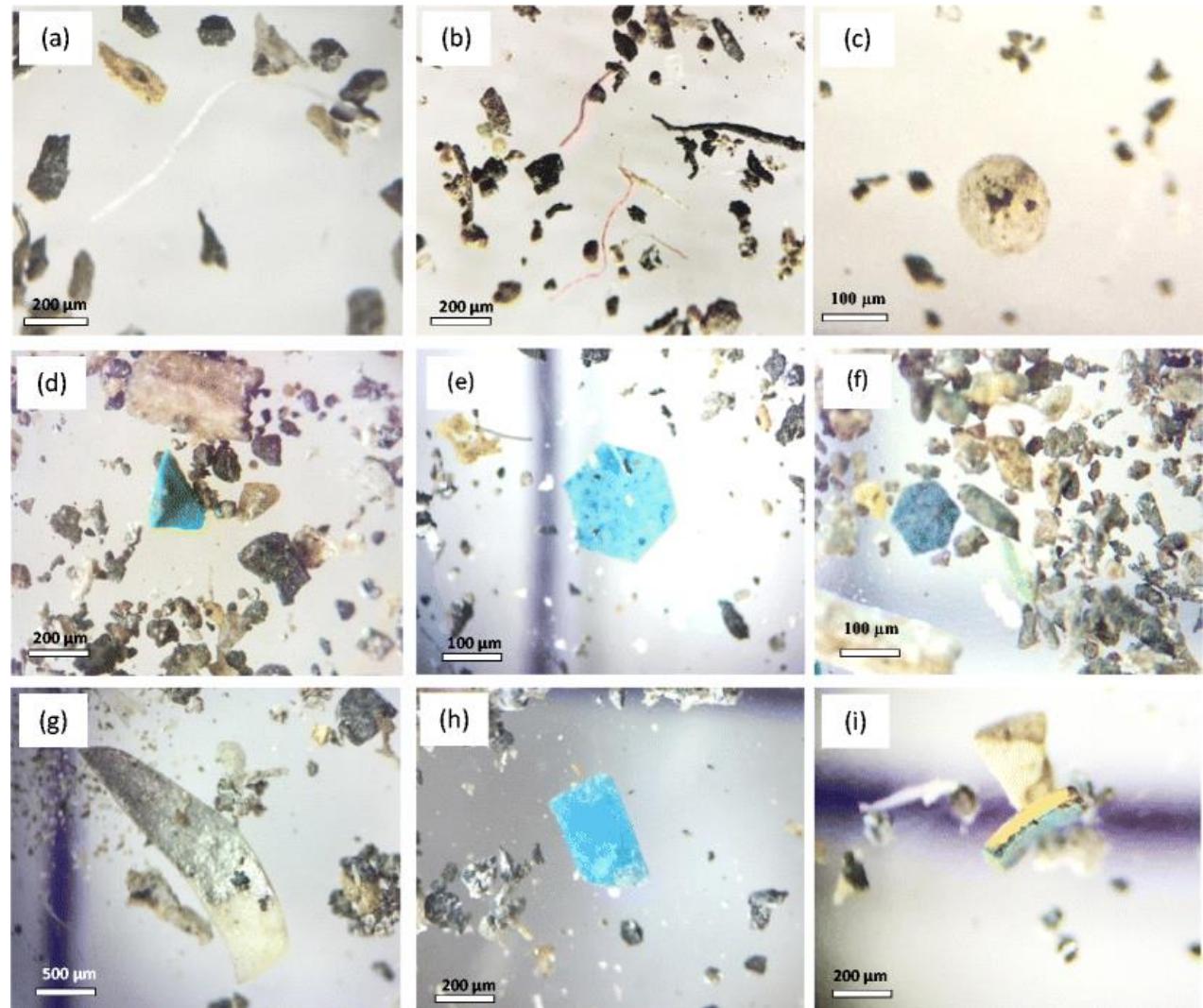


This collage of Arctic Ocean plastic, assembled by Andres Cozar, the lead author of a 2017 study "The Arctic Ocean as a dead end for floating plastics in the North Atlantic branch of the Thermohaline Circulation," and his colleagues, shows just some of the estimated billions of pieces of plastic in ice-free Arctic waters. (Nunatsiaq News file photo)

Microplastics



1. Plastic litter
2. From washing synthetic clothes
3. Cosmetics
4. Toothpaste
5. Abandoned and Derelict vessels
6. Car tires and rubbers
7. Scrubbing agents in toiletries



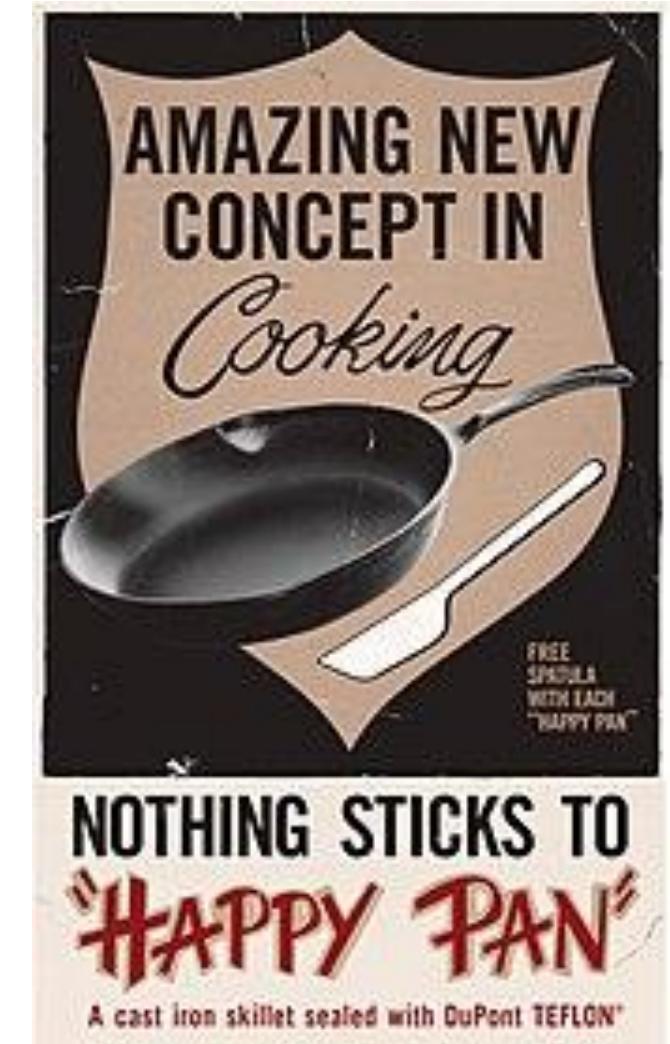
Microplastics in urban dust, Tehran

What went wrong?

- Perception of unlimited resources
- Plastics are largely stable and inert, e.g., PFAS “forever chemicals”
- Complicated EoL due to versatility in chemistry and function
- Mismanaged waste
- Limited recycling infrastructure
- Limited technologies for sorting and separation
- Global movement of microplastics
- Shipping trash may not be the answer (see China’s ban on plastic import)
- Hard to compete! Plastic is cheap, highly functional, growing in demand!



Easy, drop-in solution probably not possible



Packaging...

Next week!

Problematic because of plastics & multimatериалs, high volume, short life span, and recycling challenges



Key take aways

- Ceramics: main impacts related to mining, processing, and difficulties in recycling
- Plastics: it's complicated...
- Next week: Packaging and guest lecture on Sustainability at EPFL (what's the carbon footprint of a lab?)
- May 14: Glencore activity (please prioritize attendance – will not be recoded)
- May 21 and 30: Final project presentations! Summer in sight!